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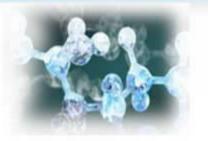
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SSN: 2161-0444

## Temple University School of Pharmacy

Department of Pharmaceutical Sciences

Moulder Center for Drug Discovery Research

Daniel Canney, Ph.D.

Chair and Director of Graduate Studies

#### About TUSP ...

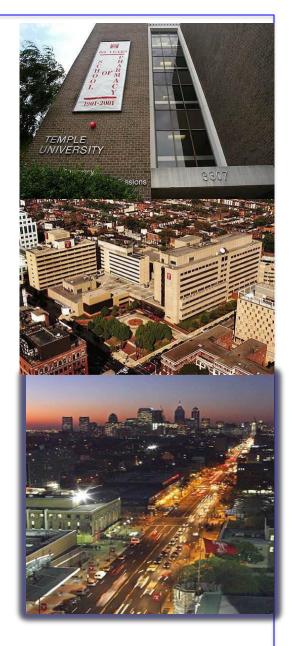
- Temple University School of Pharmacy is one of the oldest pharmacy schools in the country (Founded 1901) and is part of a major educational and research institution.
- Located on a comprehensive health sciences campus which includes: Schools of Pharmacy, Medicine, Dentistry, Health Professions, Temple Hospital, Temple Children's Medical Center
- TUSP maintains high standards while adhering to the University's mission of providing affordable, high quality education to those who wish to learn regardless of income

#### Current T.U.S.P. Centers and Facilities:

- Moulder Center for Drug Discovery Research (MCDRR)
- In Vitro ADME and Pharmacokinetics Laboratory
- cGMP facility
- Proteomics facility
- Jayne Haines Center for Pharmacogenomics and Drug Safety

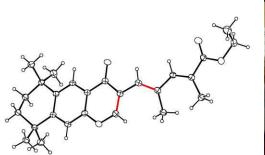
#### Current Programs

- Doctor of Pharmacy (PharmD) degree
- MS and PhD in Pharmaceutical Sciences
  - Concentrations: Pharmaceutics, Medicinal Chemistry, & Pharmacodynamics
- DBMD Program through the Office of International Studies
- MS Quality Assurance/Regulatory Affairs (premier US program)



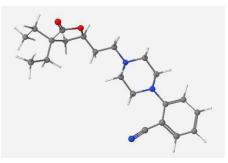
#### Daniel J. Canney, Ph.D., R.Ph.

Chair, Department of Pharmaceutical Sciences



RAR ligand  $(IC_{50} = 40 \text{nM})$ 





Muscarinic ligand  $(IC_{50} = 17 \text{ nM})$ 

Rong Gao, Siva Annadurai, Safura Nantogma, Richie Bhandare, Otito Iwuchukwu, (Shyam Desai, Weilin Sun; graduates of program)

#### Lead Optimization, Structure-Activity Relationship (SAR) Studies

- Cholinergic receptor (nicotinic and muscarinic) ligands
- Serotonergic receptor ligands and ligands for other GPCRs
- Retinoic acid receptors (RAR) ligands

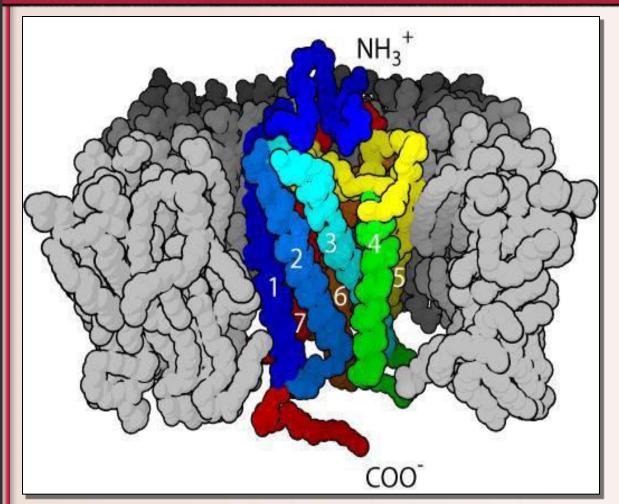
# Lead modification approaches in the design, synthesis and evaluation of novel musearinic ligands

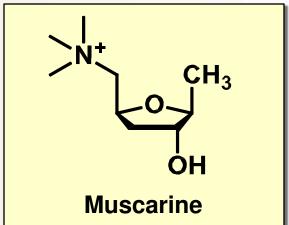
Dr. Daniel J. Canney

## **Lead Modification – Muscarinic Ligands**

- Background Information
  - Muscarinic receptors
  - □ Crystal structures
  - ☐ Ligands and allosteric regulators
- Research Design and Results
  - Lead molecules and specific aims
  - Molecular modification strategy
    - Region 1: cationic center
    - Region 2: hydrogen bonding region
    - Region 3: linker
    - Region 4: carbonyl oxygen
- Summary
- Acknowledgement

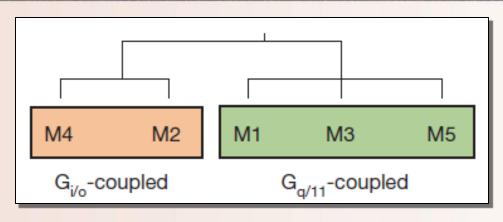
#### Muscarinic receptors - G-Protein Coupled Receptors





**GPCR** 

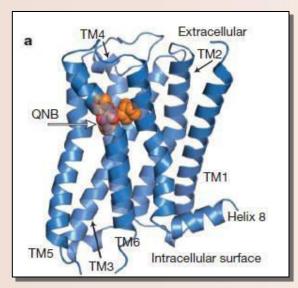
#### **Muscarinic Receptors Families**



Kruse AC and etc, Nature., 2012, 482, 552-559

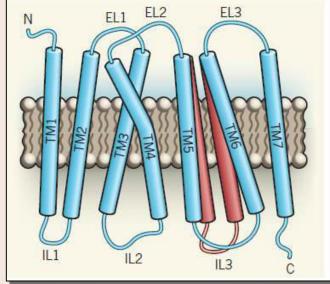
	Distribution	Therapeutic potential		
$M_1$	CNS	Alzheimer's disease (agonist) Parkinson's disease (PD; antagonist)		
$M_2$	CNS, Heart	Alzheimer's disease (antagonist)		
$M_3$	CNS, smooth muscles	OAB, COPD, IBS (antagonist)		
$M_4$	CNS	Parkinson's disease (PD: antagonist)		
M <sub>5</sub>	CNS	Parkinson's Disease: Addiction?		

#### **Crystal Structures**



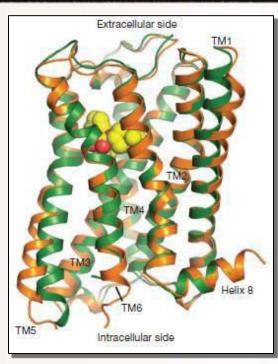
#### M<sub>2</sub> receptor with QNB

Haha K, Kruse AC and et. al. *Nature*, **2012**, 482, 547-552



Differences between M<sub>2</sub> and M<sub>3</sub> receptor subtypes.

Kow, RL; Nathanson, NM *Nature.*, **2012**, 482, 480-481



Overlap of M3 receptor (green) and M2 receptor (orange)

Kruse AC and et al *Nature.*, **2012**, 482, 552-559

#### **Muscarinic Ligands**

3a

#### M<sub>1</sub> antagonist, Ki=12.7nM, 6-35 folds selectivity

Lewis, L. M. and et.al. *Bioorg. & Med. Chem. Lett.* **2008**, 18, 885

7a

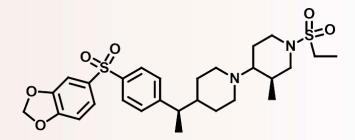
#### M₁ agonist, Ki=260nM

Malviya, M. and et.al. Bioorg. & Med. Chem. 2009, 17, 5526

13a

#### M<sub>2</sub> antagonist Ki=2.7nM, M<sub>2</sub>/M<sub>1</sub> selectivity of 40-fold

Kozlowski, J. A.and et. al. Bioorg. & Med. Chem. Lett., 2000, 10, 2255



16a

#### M<sub>2</sub> antagonist Ki=0.7nM, M<sub>2</sub>/M<sub>1</sub> selectivity of 109 fold

Kozlowski, J. A. and et.al. Bioorg. & Med. Chem. Lett., 2002, 12, 791



#### **Muscarinic Ligands**

M<sub>3</sub> antagonist Ki=2.5nM, M<sub>3</sub>/M<sub>2</sub> selectivity of 1100 fold

Mitsuya, M. and et.al. Bioorg. & Med. Chem. Lett., 2000, 8, 825

 $M_3$  antagonist Ki<10nM,  $M_3/M_2$  selectivity > 200 fold

Peretto, I. and et. al. J. Med. Chem. 2007, 50, 1571

34a

M₄ antagonist pKi=7.9, selectivity 30-50 fold

Varoli, L. and et. al. *Bioorg. & Med. Chem. Lett.*, **2008**, 18, 2972

#### **Muscarinic Ligands**

40a

#### M<sub>5</sub> antagonist

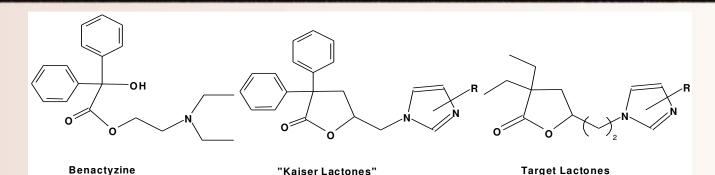
- 11-fold selectivity for M<sub>5</sub>/ M<sub>1</sub>, little activity at the M<sub>2</sub>-M<sub>4</sub>
- modest affinity (Ki = 2.24  $\mu$ M), potent (IC<sub>50</sub> = 0.45 nM)

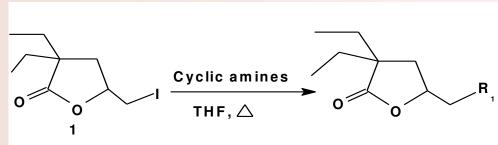
Zheng, G., Smith, A. M., Dwoskin, L. P., J. Med. Chem., 2013, 56, 1693-1703

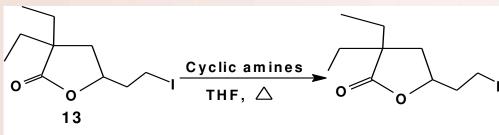
#### **Lead Modification – Muscarinic Ligands**

- Background Information
  - Muscarinic receptors
  - ☐ Crystal structures
  - ☐ Ligands and allosteric regulators
- Research Design and Results
  - Lead molecules and specific aims
  - Molecular modification strategy
    - Region 1: cationic center
    - Region 2: hydrogen bonding region
    - Region 3: linker
- Summary
- Acknowledgement

#### Lead Optimization – Addition of Arvl Rings







2. 
$$R_1$$
 = piperidine

3. 
$$R_1 = morpholine$$

$$R_1 = N$$

**5.** 
$$R_2 = CH_3$$

**6.** 
$$R_2 = CH_2CH_3$$

7. 
$$\mathbf{R_2} = \mathbf{CH_2CH_2CH_3}$$

8. 
$$\mathbf{R_2} = CH(CH_3)_2$$

14. 
$$R_1$$
 = Piperidine

15. 
$$R_1 = Morpholine$$

$$R_1 = N$$
16.  $R_2 = H$ 

17. 
$$R_2 = CH_3$$

18. 
$$R_2 = i-Propyl$$

#### Preliminary Data: Lead Modification and N-aryl-piperazines

#	structure	%inhib ¹	
L5	MeO N N	82	<b>4</b>
L9	N N C N CN	57	
RB1		18	
RB2	N CN CN	19	

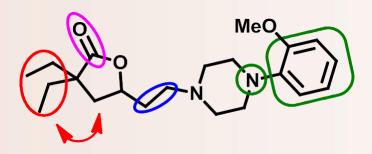
1. % inhibition at 10 µM against muscarinic receptors

**Hypothesis:** systematic modification of H-bonding, cationic and linker regions of lactone-based leads will improve ligand affinity (selectivity?)

Ahungena, A., Gabriel, J.L., Canney, D.J. Med. Chem. Res., 12:9, 2003, 481-5
Bhandare, R., Canney, D. J. Med Chem Res., 20, 2010, 558-565.

#### **Specific Aims**

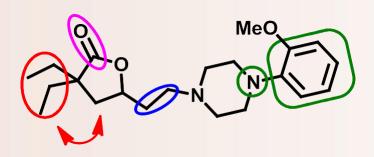
**Specific Aim 1**: Design a novel series of muscarinic ligands through addition of <u>aromatic substituents</u> and systematic modification of H-bonding, cationic and linker regions of lactone-based lead compounds to improve affinity, generate SAR data and ultimately develop subtype selective ligands.



Specific Aim 2: Develop efficient synthetic routes to the proposed ligands.

Specific Aim 3: Evaluate test compounds in muscarinic receptors binding assays

## **Lead Modification Strategy – The Specifics**



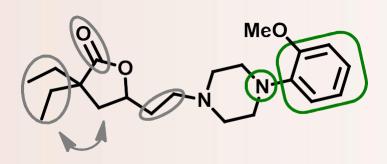
Hypothesis: It is possible to improve ligand affinity through systematic modification of H-bonding, cationic and linker regions of lactone-based lead compounds

Region 1: the physicochemical properties and position of the substituents on the cationic N-aryl piperazine region (and related heterocycles)

Region 2 the physicochemical properties and position of the substituents on the H-bonding lactone region

Region 3 the length and physicochemical properties of the linker

## **Lead Modification Strategy---Region 1**



Hypothesis: It is possible to improve ligand affinity through systematic modification of H-bonding, cationic and linker regions of lactone-based lead compounds

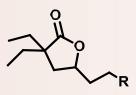
Region 1: the physicochemical properties and position of the substituents on the cationic N-aryl piperazine region (and related heterocycles)

**Regioin 2** the physicochemical properties and position of the substituents on the H-bonding lactone region

**Region 3** the length and physicochemical properties of the linker

## Region 1---Synthesis

R = commercially available piperazine



#	R	%inhib¹	#	R	%inhib¹	#	R	%inhib¹
L1	-ξ-N_N—	74	L2	HO -ξ-N N—	81	L3	-ξ-N_N——————————————————————————————————	61
L4	-ξ-N_N————OH	45	L5	-Ş-N N—	82	L6	-ξ-N_N——————————————————————————————————	75
L7	-ξ-N_N———OMe	56	L8	NC -Ş·N N−	83	L9	-ξ-n\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	57
L10	-ξ-N_N—	68	L11	-\$-N_N—	67	L13	-ξ-N_N_N_	58
L14	-ξ-N_N_	58	L15	-ξ-N_N—	53	L16	-ξ-N_N_F	63
L17	-ξ-N_N_Ph	56	L18	-ξ-N_N—\N=	64	L20	-ξ-N_N——NO <sub>2</sub>	70
L21	·ξ-N—N—Ph		L22	-ξ-N\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	33	L23	-ξ-N_N——NH <sub>2</sub>	47
L24	-ξ-N N	66	L25	-ξ- <b>N</b>	57	L26	-ξ-N	86

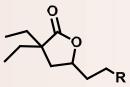
<sup>1. %</sup> inhibition at 10 μM against muscarinic receptors

#### Results

#	structure	% inhib²	#	structure	% inhib²
L21	Ph N Ph	99	L27	Ph OH Ph	96
L28	Ph Ph	99	L30	Ph	94
L31	Ph	99			

- <sup>1.</sup> IC<sub>50</sub> for muscarinic receptors
- 2. % inhibition at 10 µM for muscarinic receptors

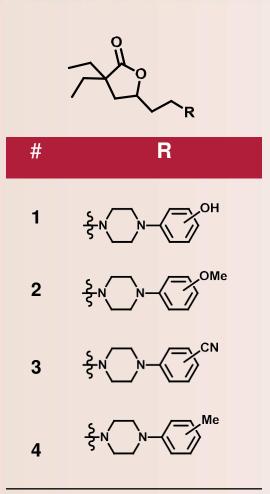


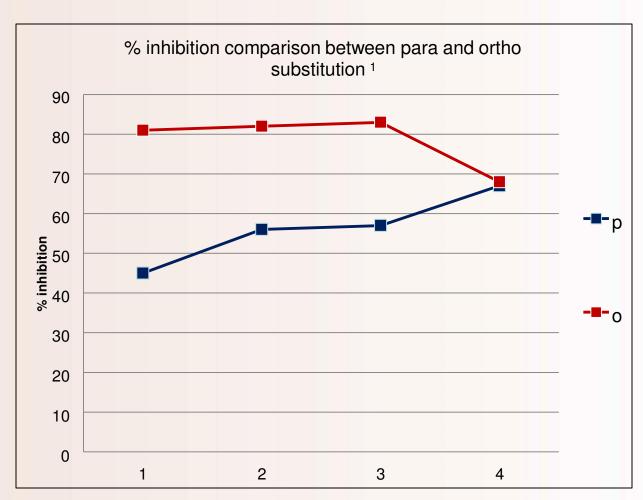


#	R	%inhib¹	#	R	%inhib¹	#	R	%inhib¹
L1		74	L2	НО	81	L3	ОН С	61
L4	;	45	L5	-ξ-NN—	82	L6	-ξ-N N- ( )	75
L7	ì	56	L8	·\$-N_N_\	83	L9	-ξ-N_N—	57
L10	-	68	L11	-ξ·N_N-\\\	67	L13	-ξ-N_N- \_CN	58
L14		58	L15	-ξ-N_N- \_\_	53	L16	-ξ-N N- F	63
L17		56	L18	-ξ-N_N—	64	L20	-}-N\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	70
L21		99	L22	-ξ-N_N—\N_)	33	L23	-ξ-N_N-\_NO <sub>2</sub>	47
L24	- ru	66	L25	-ξ-N N-(Ph	57	L26	-ξ-N_N—(	86
	HZZ			-ξ-N			-ξ-N	

<sup>1. %</sup> inhibition at 10 µM for muscarinic receptors

## Region 1---Results

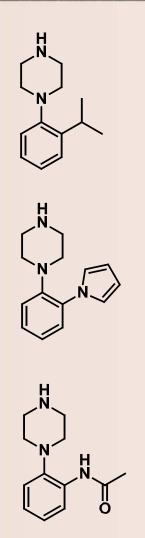




1. % inhibition at 10 µM against muscarinic receptors



## **Target N-aryl Piperazines**



## **Synthesis**

R	Yield (%)
i-Pr	12
t-Bu	0
1	0

L19

Percent inhibition: 96% <sup>1</sup>

1. % inhibition at 10 µM against muscarinic receptors

#### **Synthesis of N-Aryl Piperazines**

problems	solutions
Poor leaving group	Ns as leaving group
Unprotected amine	Ns as protecting group
High boiling point solvent	Microwave assisted
Long reaction time	reaction

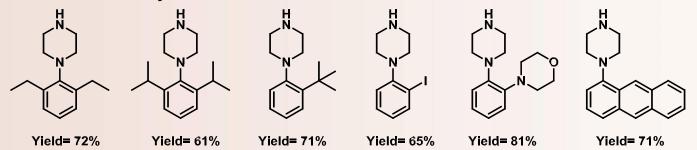
#### Results

#	R	Yield	#	R	Yield	#	R	Yield
6a		80%	6e		68%	6i		81%
6b		71%	6f		86%	<b>6j</b> [		71%
6c		72%	6g		65%	6k	N <sub>N</sub> 's	60%
6d		61%	6h		71%	61		66%

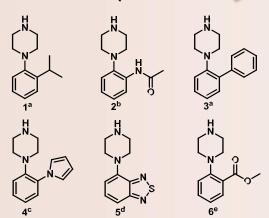
Gao, Rong.; Canney, Daniel., J. Org. Chem., 2010, 7451

#### Results

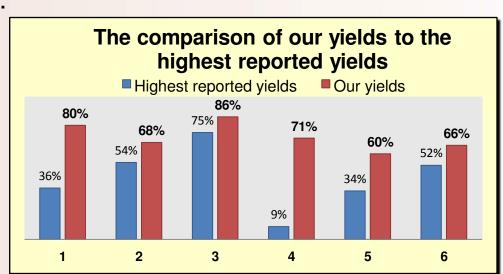
#### 1. Provides the only facile route to access these molecules



#### 2. Provides an improved method to access these molecules in terms of yield and reaction time (24~72h —> 1h).



<sup>a</sup>Elworthy T. R. Bantle G. W.; *J. Med. Chem.* **1997**, *40*, 2674. <sup>b</sup>Mills, S. G.; MacCoss, M.; PCT Int. Appl. WO 9825617A1. <sup>c</sup>Roche, F. H.; Eur. Pat. Appl. EP 0748800A2, **1996**; <sup>d</sup>Heinrich, T.; Seyfried, C.; U.S. Pat. Appl. US

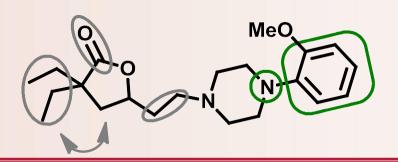


## Region 1---Results

#	R	%inhib¹	#	R	%inhib¹	#	R	%inhib¹
L33		96	L34	t Bu 71		L35	Ph	94
L36		87	L37	₹N_N-\ 52		L38	₹n _ n ,ο	63
				i-Pr N-N-			\_N	
L39		34	L40	i Pr 72		L41	₹n_n-(_)	48
L42	-n_n-( )	86	L43	₹N_N- 90			₹N_N-	
	N_S_N  \_//							
	₹N_N \			₹N_N				

<sup>1. %</sup> inhibition at 10 μM for muscarinic receptors

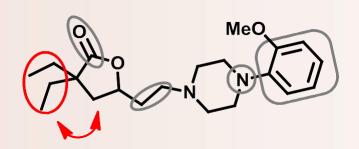
#### Region 1---Conclusions



Region 1: the property and position of the substituents on the cationic N-aryl piperazines (or similar N-heterocycles) affects the affinity of the ligands.

- Ortho substitution favored over para substitution
- Isopropyl, phenyl and iodo group are preferred groups for ortho substitution based on the series of compounds tested herein
- Biphenyl substitution provided the highest affinity compound (L21,99%; L28, 99%; L31, 99%) among those tested in the present study

## **Lead Modification Strategy---Region 2**



Hypothesis: It is possible to improve ligand affinity through systematic modification of H-bonding, cationic and linker regions of lactone-based lead compounds

**Region 1:** the physicochemical properties and position of the substituents on the cationic N-aryl piperazine region (and related heterocycles)

Region 2 the physicochemical properties and position of the substituents on the H-bonding lactone region

**Region 3** the length and physicochemical properties of the linker

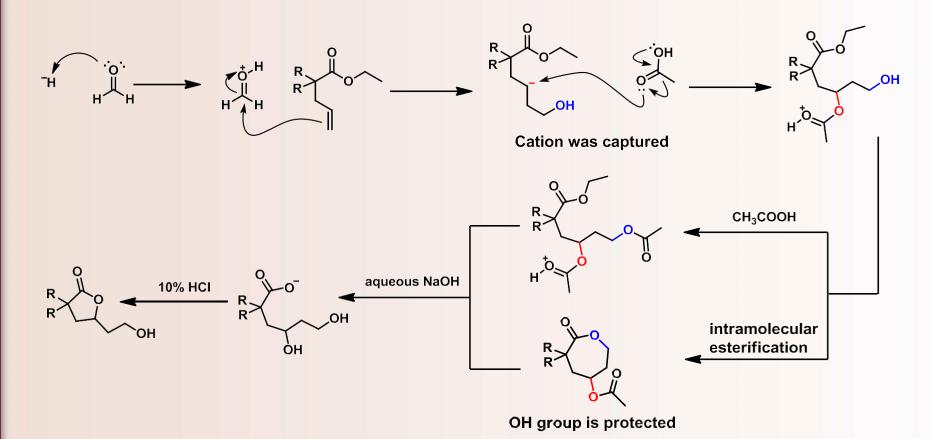
## Region 2--Design of Ligands

#### Region 2--Development of New Synthetic Route

	Reaction scheme	comment
Problematic reaction	BF <sub>3</sub> , (CH <sub>2</sub> O)n  19% + 15%  BBr <sub>3</sub> 35%	<ul><li>Low yield</li><li>Difficult separation</li></ul>
Mechanism	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	<ul> <li>Active cation</li> <li>Unprotected hydroxyl group</li> </ul>

#### **Solution – Modified Prins Reaction**

- Use H<sub>2</sub>SO<sub>4</sub> as catalyst instead of BF<sub>3</sub>
- Use acetic acid as solvent instead of DCM



#### **Region 2---Synthesis**

$$R_1$$
 $R_1$ 
 $R_1$ 

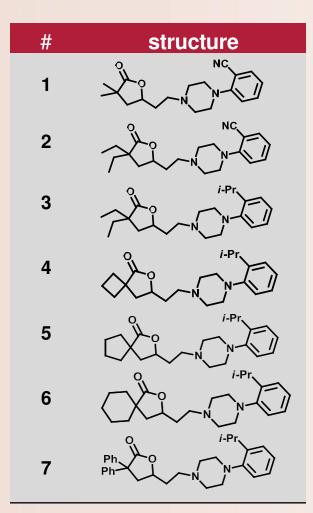
Reagents and conditions: (a) CH<sub>3</sub>COOH, paraformaldehyde, H<sub>2</sub>SO<sub>4</sub>; (b) NaOH, H<sub>2</sub>O, reflux, H<sub>2</sub>SO<sub>4</sub>;

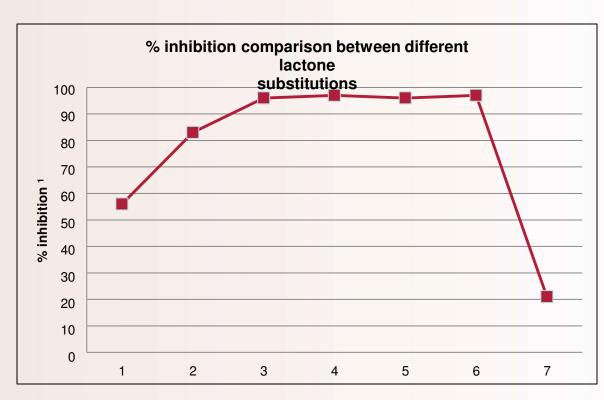
R <sub>1</sub>	compound #	Yield (%) <sup>a</sup>
Methyl	4a	81
Ethyl	4b	76
Spiro (4)	4c	74
Spiro (5)	4d	73
Spiro (6)	4e	73
Phenyl	4f	66

a Isolated yield

School of Pharmacy

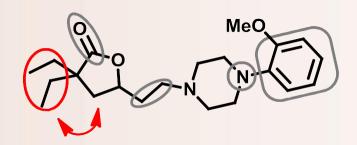
#### Region 2---Results





1. % inhibition at 10 μM against muscarinic receptors

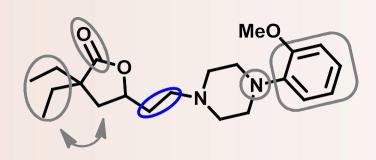
#### Region 2---Conclusions



Region 2: the property and position of the substituents on the H-bonding lactone region affects affinity of the ligands.

- Diethyl and spiro (4~6) substitution on the lactone region gave similar affinity
- In the alpha position, reducing the substitution size from diethyl to dimethyl or increasing the size to phenyl had negative effects on binding
- For substituents at the beta position, SAR data to be published soon

## **Structural Modification Strategy---Region 3**



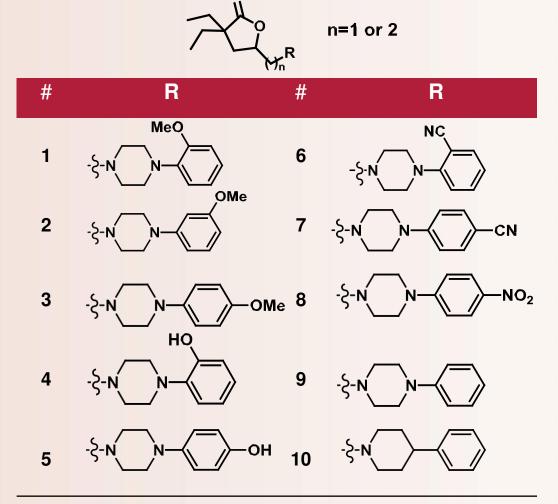
Hypothesis: It is possible to improve ligand affinity through systematic modification of H-bonding, cationic and linker regions of lactone-based lead compounds

**Region 1:** the physicochemical properties and position of the substituents on the cationic N-aryl piperazine region (and related heterocycles)

**Regioin 2** the physicochemical properties and position of the substituents on the H-bonding lactone region

Region 3 the length and physicochemical properties of the linker

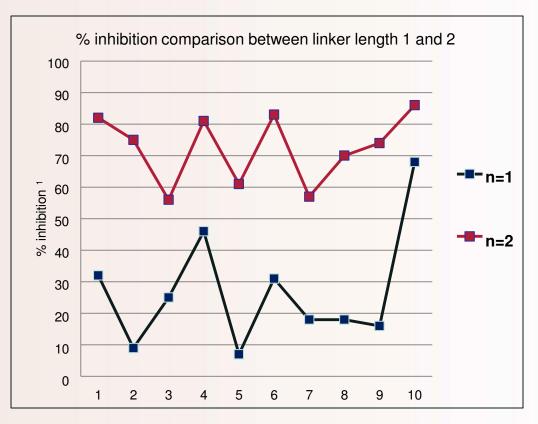
#### **Region 3---Design of Ligands**



1. % inhibition at 10 µM against muscarinic receptors

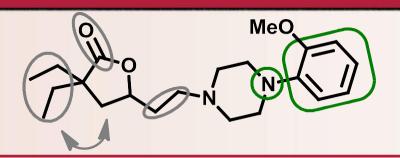
#### Region 3---Results

#	R	#	R
	-Ş-N N—	6	NC -Ş·N N−
2	-Ş-N_N——————————————————————————————————	7	-Ş-N_N—CN
	-ξ-N_N——————————————————————————————————	e 8	-ξ-N_N—\_NO <sub>2</sub>
4	-ξ-N N—	9	-Ş-N_N—
5	-ξ-N_N—————OI	10	-ξ-N



<sup>1. %</sup> inhibition at 10 µM against muscarinic receptors

#### Regions 1, 2, 3. Summary of Conclusions



**Region 1:** the property and position of the substituents on the cationic N-aryl piperazines (or similar N-heterocycles) affects the affinity of the ligands.

- Ortho substitution favored over para substitution
- Isopropyl, phenyl and iodo group are preferred groups for ortho substitution based on the series of compounds tested herein
- Biphenyl substitution provided the highest affinity compound (L21,99%; L28, 99%; L31, 99%) among those tested in the present study

Region 2: the properties, position of substituents on H-bonding region affects ligand affinity.

- Diethyl and spiro (4~6) substitution on the lactone region gave similar affinity
- In the alpha position, reducing the substitution size from diethyl to dimethyl or increasing the size to phenyl had negative effects on binding
- For the beta position, data to be reported in the near future.

Region 3: the length, electronic nature of linker affects affinity. N=2 is favored over 1.

The series is being evaluated further as potenital subtype selective ligands for muscarinic subtypes.

School of Pharma

## **Acknowledgements**

Dean's Office, Temple University School of Pharmacy (Dr. Peter Doukas)

Department of Pharmaceutical Sciences, School of Pharmacy: faculty, staff members and graduate students.

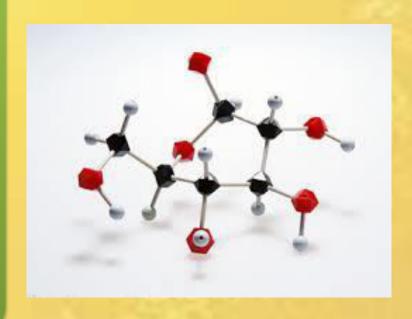
Moulder Center for Drug Discovery Research: Dr. Magid Abou-Gharbia (Director), Dr. Ben Blass and Dr. Wayne Childers

Canney Lab Members, Rong Gao (2013), Richie Bhandare (2012), Siva Annadurai (2011), Otito Iwuchukwu (2010), Safura Nantogma (2009), Shyam Desai (2009), Weilin Sun (2008)

National Institute for Mental Health (NIMH) - Psychoactive Drug Screening Program

## Medicinal chemistry Related Journals

- Drug Designing: Open Access
- Biochemistry & Pharmacology
- Advances in Pharmacoepidemiology & Drug Safety



## Medicinal chemistry Related Conferences

- ➢ 3rd International Conference on Medicinal Chemistry & Computer Aided Drug Designing
- ➤ 3rd International Conference and Exhibition on Pharmacognosy, Phytochemistry & Natural Products



