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JUNWANG TANG

Editor PPT

Biography

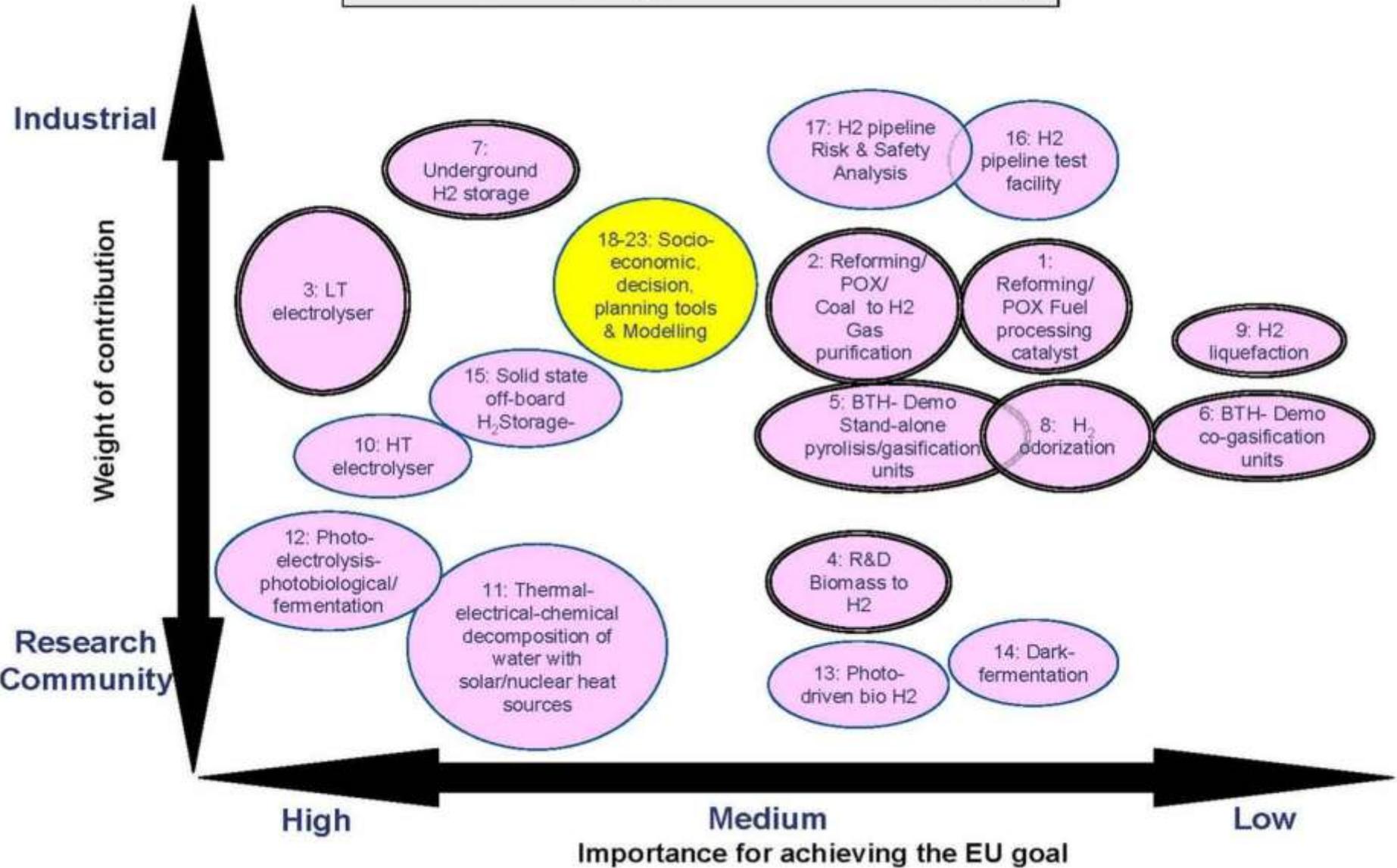
- Dr. Tang received his PhD in Physical Chemistry in 2001 from Dalian Institute of Chemical Physics. Between 2002-2005, he had been a Japan Society for the Promotion of Science (JSPS) Fellow & NIMS researcher working on solar energy conversion in NIMS, Japan. After that, he moved to the Department of Chemistry, Imperial College and then the Department of Chemical Engineering, UCL as a Senior Researcher, Lecturer in Energy, and then promoted Senior Lecture and Reader. He has been recognized by an Outstanding President's Award of Chinese Academy of Sciences (2001), JSPS Fellowship (2003) and Young Scientist Award by the International Association of Catalysis Societies (2008).
- He is the Guest Editor-in-Chief of a special issue of International Journal of Photoenergy, 2012, Associate Editor of Chinese Journal of Catalysis and sits on the editorial board of several international journals. He is an Honorary Lecturer in Chemistry at Imperial College London and Adjunct Professor in Chinese Academy of Sciences and Nanjing Tech University. Up to now, he has published ~100 papers with a total citation of >3100.

Research Interests

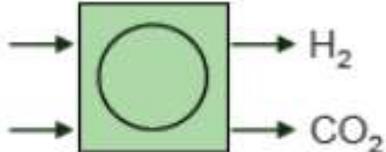
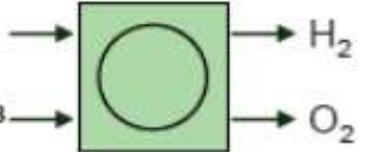
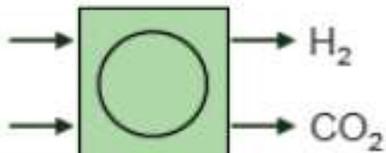
- ▣ His Research Interests lie in Nanostructured crystals and films synthesis, in particular by microwave promoted flow chemistry; Solar H₂ synthesis; CO₂ capture and conversion to a renewable fuel (Artificial Photosynthesis); CH₄ conversion; Photocatalytic contaminant decomposition; Biomaterials; Microwave catalysis.

Solar Hydrogen Synthesis

Sustainable Hydrogen Production and Supply



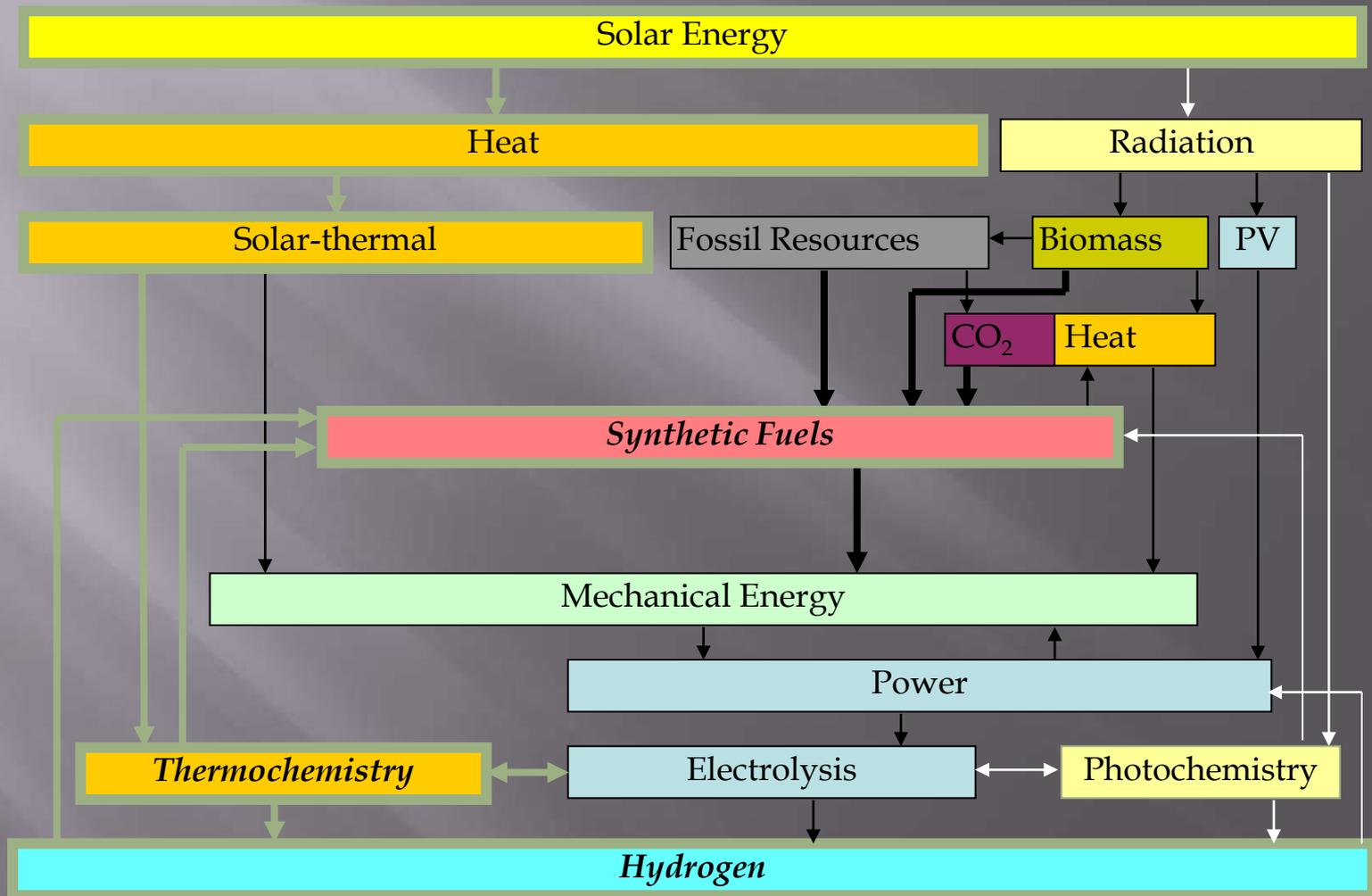
Hydrogen production – benchmark processes for solar technologies

Technology	Process	Governing reaction ¹	Variations
SMR Steam Methane Reforming		$\text{CH}_4 + 2\text{H}_2\text{O} \rightarrow 4\text{H}_2 + \text{CO}_2$	<ul style="list-style-type: none"> ▪ On-site SMR ▪ Central SMR ▪ Central SMR + CCS
WE Water Electrolysis		$2\text{H}_2\text{O} \rightarrow 2\text{H}_2 + \text{O}_2$	<ul style="list-style-type: none"> ▪ On-site WE ▪ Central WE
CG/(IGCC) Coal Gasification /Integrated Gasification Combined Cycle		$\text{C} + 2\text{H}_2\text{O} \rightarrow \text{CO}_2 + 2\text{H}_2$	<ul style="list-style-type: none"> ▪ CG ▪ CG + CCS ▪ IGCC ▪ IGCC + CCS

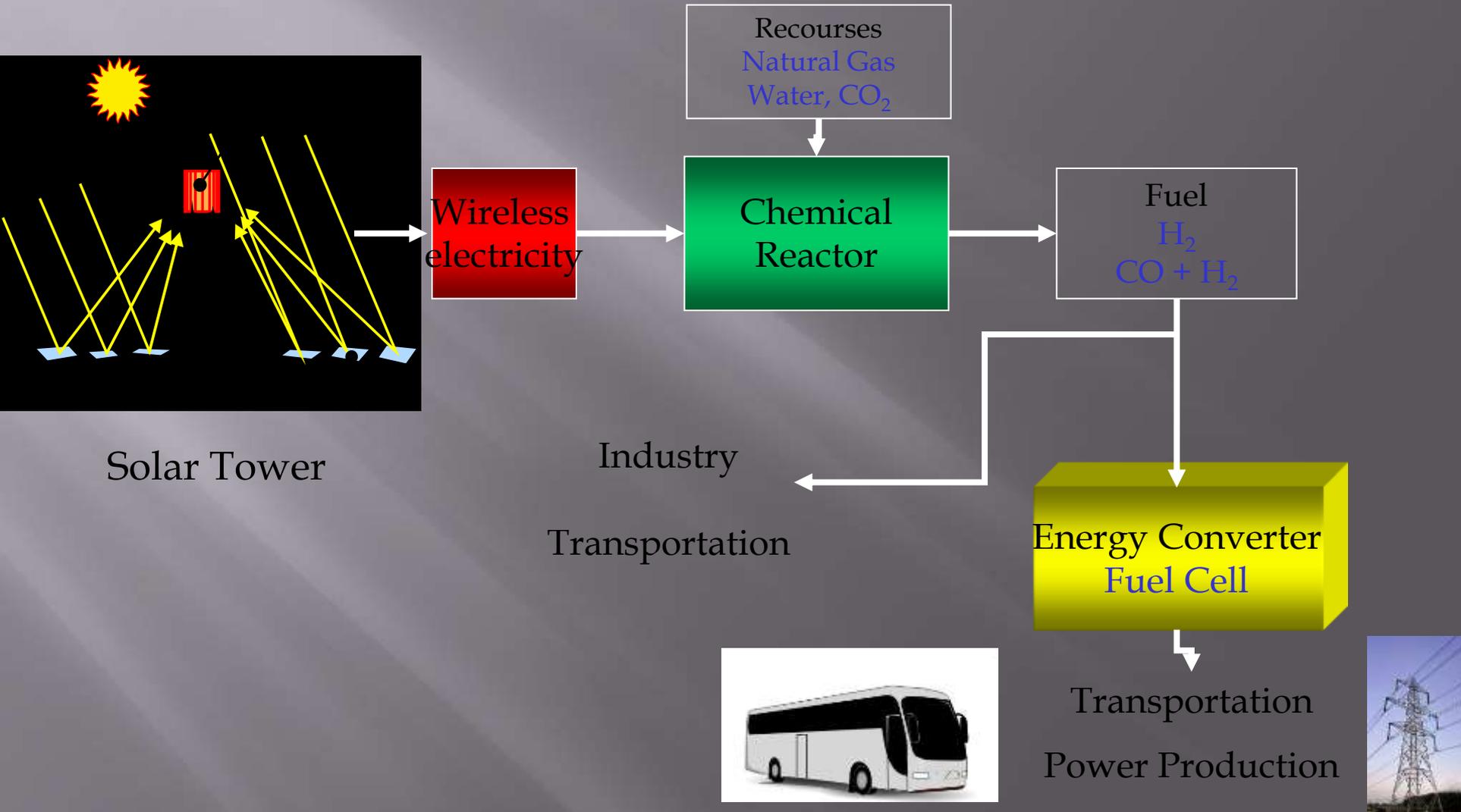
¹ Simplified reaction

² Includes co-firing with biomass

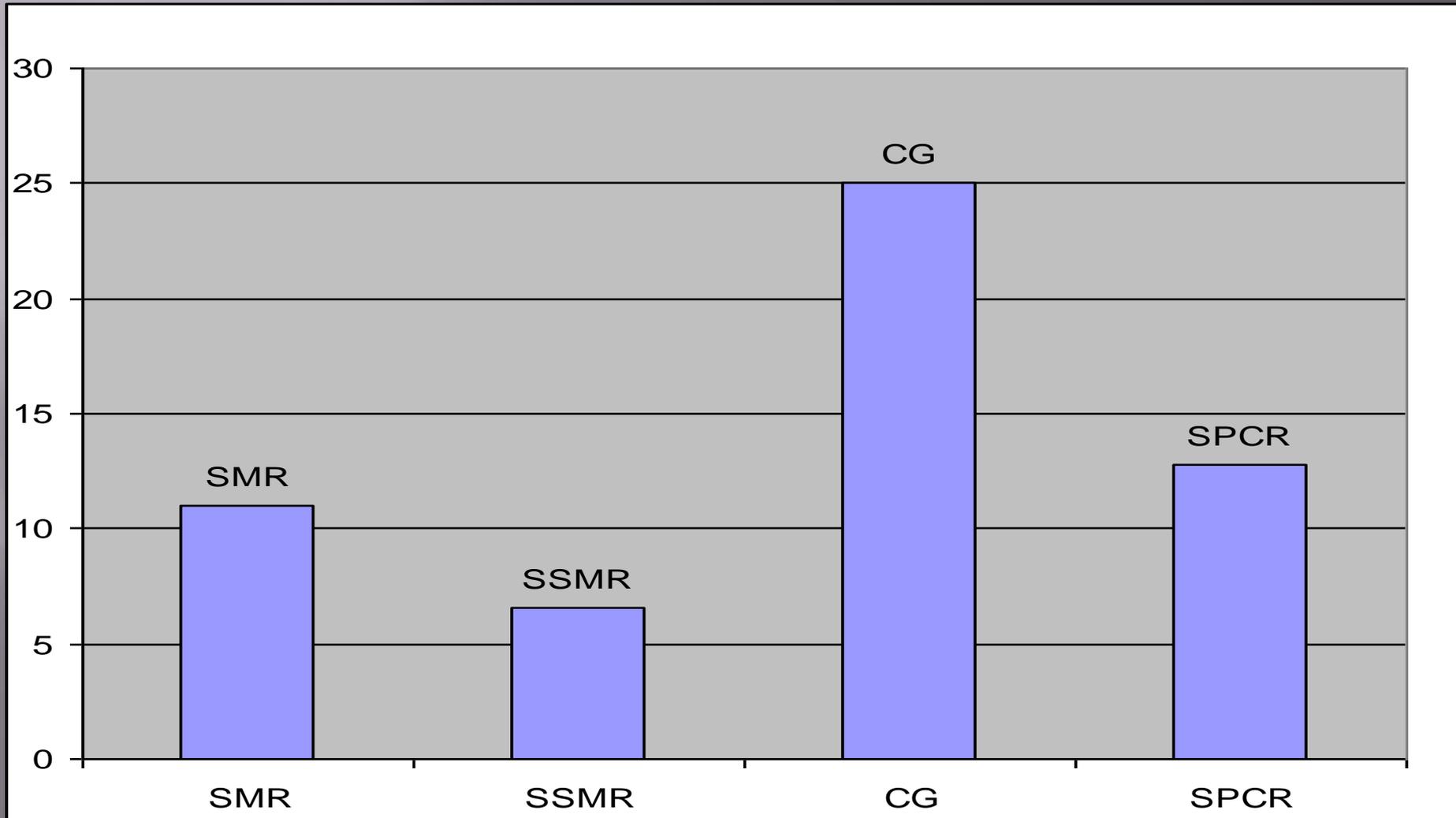
Energy Routes



Principle of the solar fuel production



CO2 Reduction by solar heating of state of the art processes like steam methane reforming and coal gasification



Efficiency comparison for solar hydrogen production from water

Process	T [°C]	Solar plant	Solar-receiver + power [MWth]	η T/C (HHV)	η Optical	η Receiver	η Annual Efficiency Solar – H ₂
Direct solar driven water splitting	25	No need	semiconductor	NA	NA	NA	>10%
Electrolysis (+solar-thermal power)	NA	Actual Solar tower	Molten Salt 700	30%	57%	83%	14%
High temperature steam electrolysis	850	Future Solar tower	Particle 700	45%	57%	76,2%	20%
Hybrid Sulfur-process	850	Future Solar tower	Particle 700	51%	57%	76%	22%
Hybrid Copper Chlorine-process	600	Future Solar tower	Molten Salt 700	49%	57%	83%	23%
Nickel Manganese Ferrit Process	1800	Future Solar dish	Rotating Disc < 1	52%	77%	62%	25%

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