



Synthesis and electrical properties of ZnO-ITO and Al-ITO thin film by spin coating technique through sol gel process

Hardeli^{1*}, Hary Sanjaya² and Rahadian Zainul³

^{1,3}Physical Chemistry Laboratory, Padang State University, Padang, Indonesia

²Analytical Chemistry Laboratory, Padang State University, Padang, Indonesia

ABSTRACT

In this research we have investigated synthesis and electrical properties of ITO (Indium Tin Oxide) thin layers doped Aluminum (Al) and ZnO with spin-coating technique through sol-gel process and calcined at ± 550 °C for 1 hour. Effect of Al and ZnO doping on the conductivity of ITO with 0%, 1%, 3% and 5% w/v dopant concentration and the number of coating (4 and 5 layers) has reported. ITO-Al has 2 phases, these are rhombohedral and cubic, the crystal size is 67.31 nm. Meanwhile, ITO-ZnO are rhombohedral, cubic (bixbyite) and hexagonal (wurtzite), with crystallite size value was 67.4 nm. The surface morphology data indicated film thickness was 3.4 μm (ITO-Al) and 0.974 nm (ITO-ZnO). The electrical properties shows that the optimum film in 4 layers coating with the addition of 5% doping value was 17 $\text{k}\Omega/\text{cm}^2$ (ITO-Al) and 5-layered by addition of 5% of doping ZnO (80.800 $\text{k}\Omega/\text{cm}^2$). Meanwhile, ITO-Al thin film with 4 layers coating without doping was 9.331 $\text{k}\Omega/\text{cm}^2$, and for ITO-ZnO (5 layers) coating without doping was 11.796 $\text{k}\Omega/\text{cm}^2$. Al and ZnO doping decrease the electrical conductivity of ITO.

Keywords : ITO, Sol-gel, Spin-coating, Thin film, Al and ZnO doping

INTRODUCTION

Indium Tin Oxide thin film, has a high transparency in the visible region and low electrical resistivity(1). ITO thin film composed of In_2O_3 doped with SnO_2 . Based on Alam research the addition of 10% SnO_2 can improve the conductivity of ITO thin film, so that it has a low resistivity(2). Thus, ITO thin film has a band gap larger than 3.5 eV and has transparent properties in visible light region(3). ITO film is usually in the form of amorphous and have low conductivity at low temperatures, so high heating temperature needed to modify the optical properties, electrical, and crystal structure. The resistivity will be reduced by increasing the calcination temperature above 400°C(4). The higher the calcination temperature, the electrical conductivity will be better.

Transparent Conducting Oxide (TCO) glass is a material coated by a metal oxide, which has optical transparency in the area of the electromagnetic spectrum and have electrical conductivity. ITO (Indium Tin Oxide) is a type of TCO which has transparency properties and high electrical conductivity(5). The combination of these properties causes ITO thin film has been used in the development of today's technology such as LCD (Liquid Crystal Display), LED (Light Emitting diodes), solar cells, optoelectronics, gas sensors(6). Improved the performance of ITO thin film becomes an important topic to explore both on an industrial scale and on a laboratory scale considering the necessary for the use of the ITO thin film in daily life.

One of the aspects that can improve the performance of the ITO thin film is the addition of doping, such as the insertion of other atoms in a semiconductor that aims to change the value of conductivity(7) Zainul et al reported Al metal have potentially as negative charge carrier in surface of material(8), so that can considerate as doping in

material. Doping acts to minimize the width of the band gap so that the energy required for the excitation of electrons from the valence band to the conduction band getting smaller(9). Some of metals and oxides can use as dopant, such as Fe, Cu, Ni, Al, ZnO, TiO₂ and almost all oxides. In this research, ZnO and Al metal use as dopant for ITO.

ZnO used as doping in this study because according to Suprayogi, ZnO is a compound that is not toxic, readily available, resistant to radiation and high thermal stability(10). In addition, ZnO has a wide band gap of 3.37 eV (at a temperature of 300K)(11). The addition of a new band of ZnO on ITO band gap will cause a decrease in the amount of energy required to electrons excitation from the valence band to the conduction band, so that it can increase the electrical conductivity of ITO(9).

The ITO film deposition using metal doping of group IIIA, among the group IIIA metal is aluminum (Al) (12). The function of aluminum doping is to increasing the value of the conductivity of ITO by occupying space in the structure of ITO and substituting Al atoms in the structure of ITO. Conductivity value of aluminum is 3.8×10^7 (Ω -m)⁻¹. According to Rosa (2009) aluminum is very easy to join and produce a low barrier (12).

In order to produce high electrical conductivity of the ITO doped ZnO and Al thin film, there are several parameters that should be modified include the formation of a material element composition, crystal size and thickness of the coating(13). The composition of the elements is one of the factors that determine the success of the preparation of the ITO doped ZnO and Al thin film, therefore in this study the influence of the composition of which will be seen by varying the doping concentration of ZnO and Al on the ITO precursor. Intake of the Al and ZnO doping concentration variation refers to the study Hoong, these are 0%, 1%, 3%, and 5%(14).

Thin film preparation are also greatly influenced by the thickness layer factor. Preliminary research that has been conducted by the author and supported by research of Daoudi showed that the number of coating layers effect on the electrical conductivity produced in which the use of 1, 2, 3, 4, 5 layers obtained better conditions at 4 and 5 layers(15). The results of the study being a reference for the ITO doped ZnO and Al thin film number of coating variation.

There are several deposition methods for the preparation of thin film, including facing target sputtering, dc magnetron sputtering, sol-gel process(10; 16; 17). Sol-gel method have the advantages among others due to the deposition process is simple, the cost of making thin film is cheaper, can be used for the manufacture of thin film with a large area without the use of expensive and complex equipment(18). Sol-gel method consist of some coating techniques, these are spin-coating, dip-coating (19) and spray-coating(20). Comparison of the three techniques, the author uses spin-coating technique in the preparation of the ITO doped ZnO and Al thin film because it has the advantage of relatively cheaper cost, more homogeneous, lower process temperatures(21). Spin-coating merupakan suatu teknik untuk mendeposisikan lapisan tipis yaitu melalui percepatan larutan pada substrat yang diputar. Proses spin-coating dilakukan dengan menggunakan alat spin-coater berkecepatan tinggi dalam jangka waktu tertentu. Semakin cepat putaran, akan diperoleh lapisan tipis yang semakin homogen dan tipis(22).

To determine the properties of the ITO doped ZnO and Al thin film is carried out characterization by SEM, XRD and digital multimeter. SEM is used to observe the surface morphology and thickness of the film layer. XRD is used to specify the existent phases and crystal size. The resistivity of the ITO doped ZnO and Al thin film is measured using a digital multimeter, where the measurement is performed under direct sunlight.

EXPERIMENTAL SECTION

A. Equipment and Materials

The equipments used were beaker glass, measuring cylinder, spatula, sprayer bottle, pipette, analytical balance, pincers, conventional spin-coater, glass substrates of 10 x 25 x 1 mm, volumetric flask, volumetric pipette, neck erlenmeyer, ultrasonic cleaner, oven, furnace, XRD, SEM, digital multimeter.

The materials used were distilled water, 96% ethanol, 21% nitric acid, ammonia p.a, (Zn (CH₃COO)₂.2H₂O), InCl₃.4H₂O, acetyl acetone p.a, SnCl₂.2H₂O, iso-propanol p.a, acetone p.a, polyvinyl pyrrolidone (PVP), nitrogen gas, ethanol 96%, nitrogen gas, Acetyl acetone, methanol pa, iso-propanol, polyvinyl pyrrolidone (PVP) and AlCl₃.6H₂O

B. Glass Substrates Cleaning

The glass substrates cleaning procedures has been adapted from Kesim procedure(20). Substrates must be free of grease and dust. Glass substrates were treated ultrasonically for 120 min with a row in soapy water, acetone, ethanol, and finally distilled water and dried at $\pm 110\text{ }^{\circ}\text{C}$ for ± 60 min after cleaning.

C. Preparation of ITO Precursor

The preparation procedures has been adapted from Yang(23). ITO precursor made from mixing In and Sn in molar ratio 9:1. $\text{InCl}_3 \cdot 4\text{H}_2\text{O}$ concentration maintained at 0.5 M was dissolved in acetyl acetone and incubated for ± 8 hours at $\pm 60\text{ }^{\circ}\text{C}$. $\text{SnCl}_2 \cdot 4\text{H}_2\text{O}$ dissolved in iso-propanol. In and Sn solution was mixed at room temperature and added PVP as an adhesive agent.

D. Preparation of ZnO Doping

The preparation of ZnO doping has been adapted from working procedure of Maddu(24). $(\text{Zn}(\text{CH}_3\text{COO})_2 \cdot 2\text{H}_2\text{O})$ was dissolved in 10 ml of ethanol. 2 ml of ammonia was added while stirring. 21% HNO_3 was added step by step until reaching pH = 6. Variations were made in the doping concentration of 1%, 3% and 5% w/v.

E. Preparation ITO doped ZnO Thin Film

The ITO doped ZnO sol deposited using a soft brush and small puddle of a fluid resin on the center of a substrate, then rotated in the spin-coater for 3 minutes at 500 rpm. Heating is done for ± 120 minutes at $\pm 80\text{ }^{\circ}\text{C}$. Coating is done by 4 and 5 layers. To form metal oxide, the glass substrate calcined at $\pm 550\text{ }^{\circ}\text{C}$ for ± 60 minutes. The ITO doped ZnO thin film ready to characterized.

F. Preparation of Al Doping and ITO doped Al Thin Film

The preparation of Al doping has been adapted from working procedure of Maddu(24). The process ITO doped Al thin film as shown in figure 1.

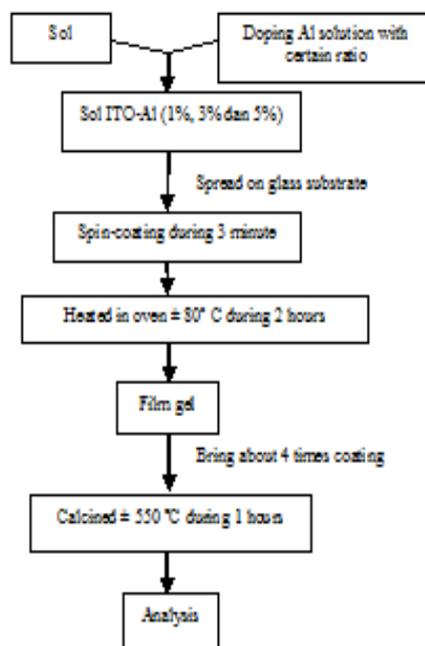


Figure 1. Systematics Procedure

RESULTS AND DISSCUSSION

The XRD pattern of ITO doped ZnO and Al was confirmed with ICSD-ICDD (*Inorganic Structure Database - International Center for Diffraction Data*), there are three type of crystals of ITO doped ZnO and two type for ITO doped Al. They are rhombohedral, cubic(*bixbyite*) and hexagonal (*wurtzite*) for ITO/ZnO, and rhombohedral and cubic (*bixbyite*) for ITO/Al. It is indicated of the thin layer growing with polycrystalline structure.

From Figure 2 there is a highest peak at $30.6501^{\circ} 2\theta$ position, the peaks of ITO-Al 5% [International Centre for Diffraction Data - Inorganic Crystal Structure Database (ICSD-ICDD)] was obtain trough the Scherrer equation,

crystal size is 67.31 nm. For ITO/ZnO, the some highest peaks at $2\theta = 21.48^\circ$, 30.56° , 35.44° , 51.01° , 60.65° , and the highest peak at $2\theta=30.56^\circ$ with sharp intensity was indicated of formation ITO-doped Al.

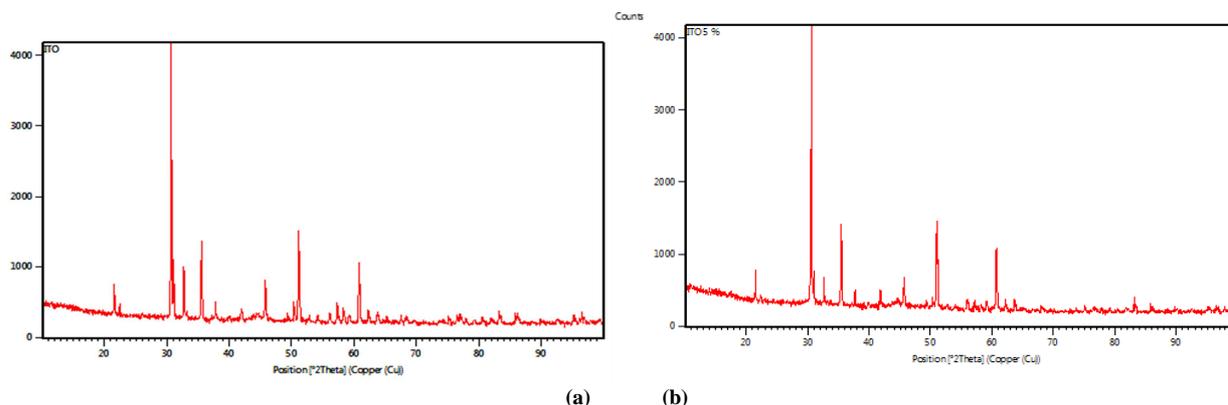


Figure 2. XRD from ITO doping with (a) ZnO 5% and (b) Al 5%

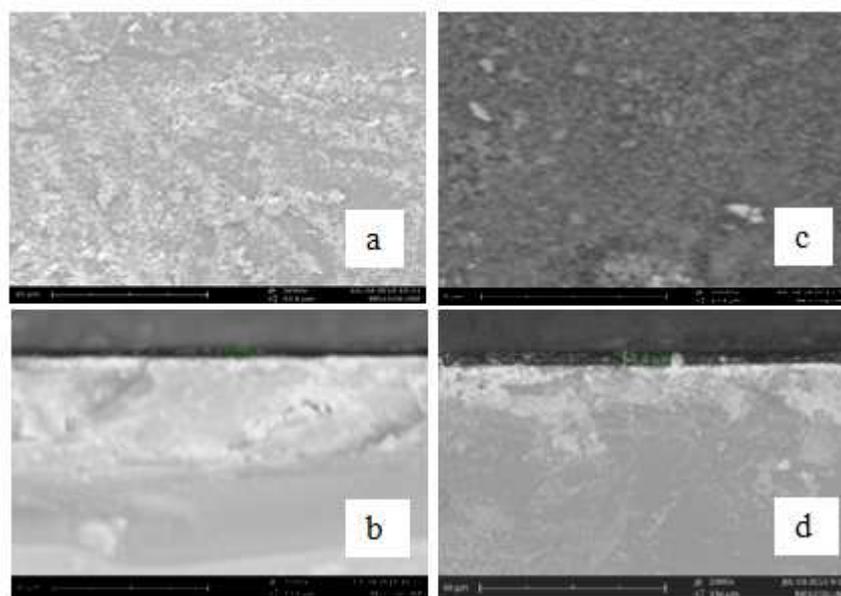


Figure 3. SEM images of the surface of the ITO-ZnO 5% thin film after being calcinated at 500°C with magnification (a) 5000 times, (b) 2000 times and surface of the ITO-Al 5% thin film after calcinating at a temperature of 550°C (c) 5000 times, (d) 2000 times magnification

SEM characterization results shown in Figure 3 (a) and (c) shows that the thin layer of ITO-ZnO 5% and ITO-Al 5% by particles deposited ITO-ZnO 5% and ITO-Al 5% hardly spread but still formed aggregates or clusters of crystals ITO-ZnO 5% and ITO-Al 5% and some particle size. SEM results showed the limitations of cracks and formation of aggregates on the surface of deposition results. It was by the presence of air bubbles on the surface of the substrate during the hydrolysis process of drying or low temperature, the greater deposition temperature, bring on the film surface become more homogeneous, so aggregate will be reduced and the gelatin process occur quickly and made the particle tied on glass substrate first before its spread. The process of layer fabrication was influenced by low rate of conventional spin-coater.

The thickness of thin film can be determined by SEM figure 3 (b) and (d), where in transverse position (cross-section) at 2000 times magnification the average of thickness is $\pm 3,4 \mu\text{m}$ (ITO-Al) and $0.974 \mu\text{m}$ (ITO-ZnO). Cheong was reported in general the thin layer have 100-200 nm thickness(6). It was indicated that formation of thin layer is thick. The thicker thin layer was produced will make the decreasing of thin layer transmittance. The results of measurements on thin-layer ITO multimeter without doping are shown in Table 1.

Table 1. Effect times coating for resistance of ITO

Times coating	Resistance (kΩ/cm ²) Of ITO
1	10 ⁵
2	1689.573
3	22.011
4	9.331
5	11.796

From the data were obtained, optimum coating on a thin layer of ITO at 4 times coating where it has the resistance value 9.331 kΩ/cm², so 4 and 5 layers can be referable in layers variation to determine the effect of doping for ITO conductivity. The measurement results of ITO thin layer with Al doping 5% are shown in Table 2.

Table 2. Measurement of resistance of 4 and 5 times coating for ITO-Al and ITO-ZnO

Dopant concentration (% w/v)	Resistance of 4 times coating (kΩ / cm ²) (Al dopant)	Resistance of 5 times coating (kΩ / cm ²) (Al dopant)	Resistance of 5 times coating (kΩ / cm ²) (ZnO dopant)	Resistance of 5 times coating (kΩ / cm ²) (ZnO dopant)
0	9.331	11.79	9.3	11.8
1	31.95	35.9	4740	312100
3	50.7	57.50	12760	264000
5	17	1271	6070	80800

When compared to the conductivity of a thin layer ITO-Al coating between 4 and 5 layers the coating, it can be seen that the constraints generated by a thin layer with 4 times coating the coating has a lower barrier than 5 layers the coating. This is caused of the greater number of layers to achieve the optimum conditions the better electrical, the highest conductivity of thin film at 4 times coating with Al doping 5% the result is 17 kΩ / cm². The addition effect of the addition of Al and ZnO doping on the electrical resistance value of ITO thin film can be described in Figure 4.

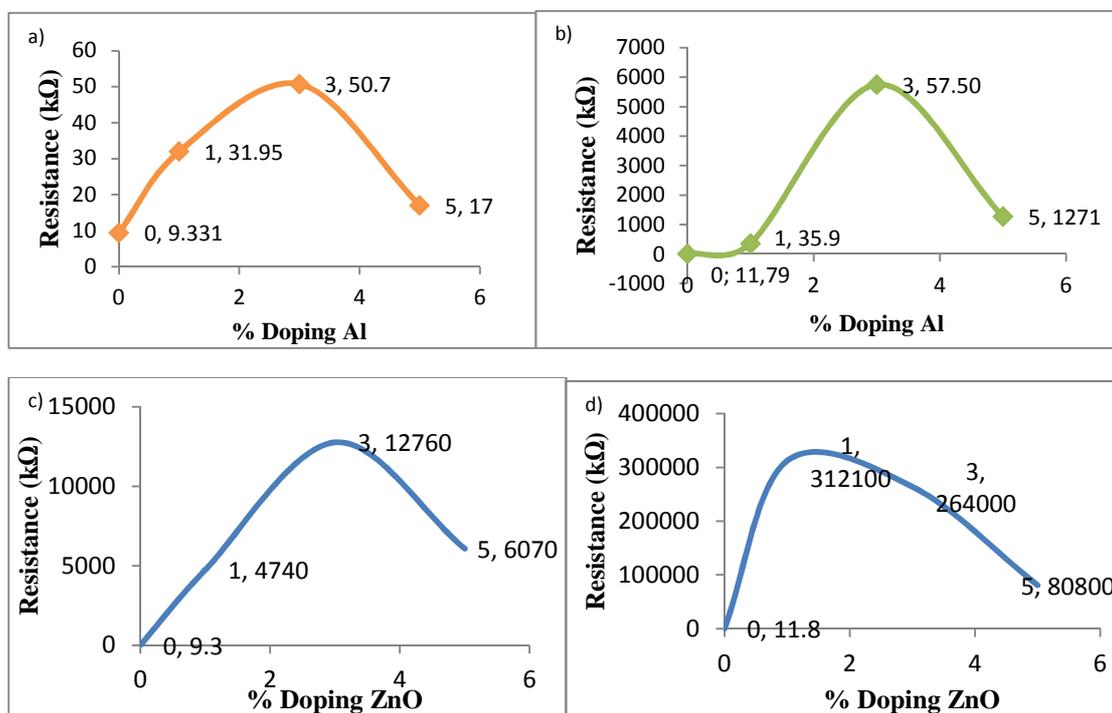


Figure 4. Resistance thin layer of ITO-Al a) 4 times coating, b) ITO-Al 5 times, and thin layer of ITO-ZnO c) 4 times coating and d) 5 times coating

The fabrication of a thin layer of ITO-Al 5% have been successfully carried out, but the resulting electrical conductivity was low. In general, addition of doping will increase the conductivity of ITO. Sistesya reported this is caused by decreasing the band gap of material for excitation of electron (15). Caused by addition of the band gap, it

will make the band gap became small and the photon energy for excite the electron will decrease, and the conductivity will increase. ITO have 3.5 eV band gap, and ZnO have 3.37 eV band gap.

Based on Figure 4, addition ZnO doping (1%) can decrease the conductivity from 9.331 kΩ/cm² to 4.74 kΩ/cm² (4 times coating), but after 5 times coating decrease to 6.070 kΩ/cm². Formation of ITO-ZnO thin layer make the low conductivity, because modified spin-coater from magnetic stirrer have low spin (500 rpm). According Kesim, to make good thin film with the 1.5 kΩ/cm², we need spin velocity is 1000 rpm(20). The high angular velocity from spin coater will make distribution of matter homogenously and will increase the optical and electrical properties.

Effect of calcination in inert condition is improve formation of material. In this case, N₂ gas will make synthesis in inert condition so that the formation thin layer will free from contaminants. N₂ gas will be function initiate ITO precursor formation. Beside that, InCl₃.4H₂O is high hygroscopic, so that will change became oxide metal when contact with atmosphere. In First mixing the material with dopants, there is not forming the valency between ITO and dopants, but only heterogenously mixture. This is indicated that ZnO is an isolator for ITO. An was reported that formation ITO thin layer under flow N₂ gas by calcination was considered by increasingly of oxygen vacancy concentration. It will make increase the charge carrier concentration and electrical conductivity was produced (3). Zainul et al reported that effect of calcination temperature in formation Cu₂O/CuO which after calcination at 400 °C was obtained percent oxygen is 16.5 % and 500°C percent oxygen on the surface of the Cu plate is 15.27%. Increasing of oxygen on the surface of the plate, will lead to an increase in the photocatalytic ability of the semiconductor material, since the formation of the conduction band(25).

CONCLUSION

This research has successfully to made a thin layer of ITO-Al dan ITO-ZnO with Spin-coating technique through Sol-Gel process, this is evidenced by the characterization of XRD, SEM and Multimeter. The optimum coating on a thin layer of ITO-Al accure at 4 times coating with the addition of 5% Al doped with electrical resistance generated is 17 kΩ/cm². The value of electrical resistance of ITO layer is lower than the value of the electrical resistance of a thin layer of ITO-Al 5%, the doping of Al tends to lower the electrical conductivity of ITO glass. This indicates that the doping is less suitable to the ITO glass. The ITO thin film have smaller resistance than the ITO doped ZnO thin film. Doping ZnO tend to lower the value of the electrical conductivity of ITO material. Better conditions of the preparation of ITO doped ZnO thin film was 5 layers by the addition of 5% doping ZnO with resistance 80.8 kΩ /cm².

REFERENCES

- [1] Ding Z. **2010**. *Journal of Nanomaterial*. 2010:1-5
- [2] Alam MJ, D.C.Cameron. **2000**.455-9
- [3] Skjaervo HS. **2013**. *Disertation. Norwegian University of Science and Technology*
- [4] Ogi T. **2006**. *Journal of Nanoparticle Research*:343-50
- [5] Helen SS. **2013**. *Norwegian : University of Science and technology Department of Meterial Science and Enegineering*.
- [6] Cheong D. **2011**. *Journal of the Korean Ceramic Society* 48:516-9
- [7] Gareso LP. **2012**. *Fisika Semikonduktor. Makassar: Lembaga Kajian dan Pengembangan Pendidikan*
- [8] Zainul R, Alif A, Aziz H, Arief S, Dradjad S, Munaf E. **2015**. *Research Journal of Pharmaceutical, Biological and Chemical Sciences* 6(4):353-61
- [9] Sistesya D, Susanto H. **2013**. *Youngster Physics Journal* 1:71-80
- [10] Suprayogi D. **2014**. *Jurnal Fisika* 3:19-26
- [11] Lupan O. **2009**. *Solar Energy Materials & Solar Cells*:1417-22
- [12] Rosa ES. **2009**. *Jurnal Elektronika, Bandung, UPI* 9
- [13] Sueta N. **2008**. *Thesis. FMIPA. University of Indonesia*
- [14] Hoong LJ. **2013**. *International Journal of Emerging Trends in Engineering and Development* 4:496-501
- [15] Daoudi K. **2002**. *Material Science and Engineering*:313-7
- [16] Kim S-M. **2009**. *Journal of the Korean Physical Society* 55:1996-2001
- [17] Stoica TF. **2008**. *Journal of Optoelectronics and Advanced Materials* 2:684-8
- [18] Yang YR. **2012**. *Research Article*:1-7
- [19] Brinker C, J. **1991**. *Thin Solid Films*:97-108
- [20] Kesim MT. **2012**. *Thesis. Middle East Technical University*
- [21] Adem U. **2003**. *Thesis. Middle East Technical Univeristy*
- [22] Saputra RI. **2012**. *Jurnal Teknik Pomits* 1:1-3
- [23] Yang LL. **2009**. *Thin Solid Films*:4979-83
- [24] Maddu A. **2006**. *Jurnal Sains Materi Indonesia* 7:85-90

[25] Zainul R, Alif A, Aziz H, Yasthopi A, Arief S, Syukri. 2015. *Journal of Chemical and Pharmaceutical Research* 7(11):57-67