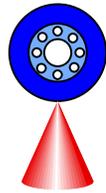




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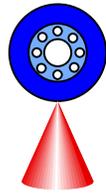
**THERMODYNAMIC AND TRANSPORT ANALYSIS
OF A BERNOULLI FLOW METER SYSTEM
FOR ANY REAL FLUID**

FOR

**NASA
MARSHALL SPACE FLIGHT CENTER**



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**For
NASA
Marshall Space Flight Center**

**by
Paul D. Van Buskirk, MSChE
Director of Technology
Quality Monitoring & Control**

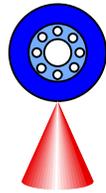
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Quality Monitoring & Control**

**April 2006
PTK-8, Revision 3**



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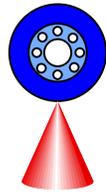


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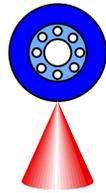
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A+ FLOWTEK, a NASA Spin-off Company

In a collaboration between NASA's MSFC and Quality Monitoring and Control (QMC) of Humble, Texas, state-of-the-art balanced flow meter technology has been developed. Based upon its work with NASA, QMC founded A+ FlowTek, also of Humble, Texas, to commercialize the balanced flow meter technology.

This Balanced Flow Meter invention has been nominated for NASA's 2006 Space Technology's Hall of Fame.

The Balanced Flow Meter determines the flow rate in piping, channel, and conduit systems. It provides highly accurate flow metering, flow limiting, or flow conditioning in any fluid flow system. As a flow meter, the technology provides flow measurement with minimal intrusion into the flow path and requires no moving parts. When compared to a standard orifice plate, it provides a 100% increase in pressure recovery, a 10-fold increase in accuracy, and a 15-to-1 reduction in acoustic noise generation. As a flow-limiting device, the technology can simulate fluid flow loads for facility certification and provide accurate flow limiting for safety and process control systems. The technology's sizing is more accurate than the currently used orifice plate technology, and it requires less space and cost when compared to venturi flow technology while maintaining similar performance. As a flow-conditioning device, the technology may improve process and equipment performance by conditioning fluid flow and fluid energy profiles around elbows, combustion chambers, pump inlets, etc.

Fluid flow measurements are used extensively in the processing industries for refineries and chemical, power, and pharmaceutical plants. The Balanced Flow Meter's applications to NASA's liquid propulsion systems and test facilities are numerous.

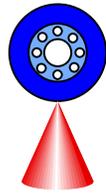
Disclaimer

This document contains information confidential and highly sensitive to QMC. The methods and procedure used are the sole property of QMC. These methods and procedures are used within the fluid processing industries for high accuracy flow measurement, and are the approaches taken by the US industry leaders in the field, such as Dresser-Rand, ABB, DuPont, Rosemont, etc. The development of these methods and procedures are used by QMC clients, such as DuPont, Dresser-Rand, and ABB. Rosemont and others in the field are considered QMC (and A+ FLOWTEK) competitors. The method and procedures provided herein enables QMC (A+ FLOWTEK) and associated clients, competitive leadership in the field and are not to be disclosed to any individual or entity without the written consent of QMC.

This document is provided to NASA/MSFC for understanding of the "head" meter/Bernoulli equation technical background and application for accurate flow measurement for any fluid system. These methods and procedures may be patented by NASA/MSFC and QMC, and are applicable to all components of the SSME and other



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new rocket designs. These components include conduits, valves, turbines, expanders, mixers, combustion chambers, and nozzles. The general applicability is for any mass, momentum, energy, power, and thermodynamic balance, as applied to any fluid, mixture, and reaction process of any phase.

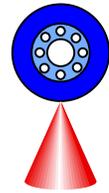
Additionally, the Modified Iterative Measurement Test (MIMT©) algorithm developed for flow meter error detection is owned exclusively by Heenan and Serth, Ltd. It is copyrighted by the American Institute of Chemical Engineering (AIChE). The algorithm is under exclusive contract with QMC. The methods and procedures for utilization of the MIMT© algorithm cannot be used by NASA/MSFC or any other entity, without the expressed written consent of QMC.

Questions concerning the technical basis or use of the information contained within are to be directed to: Jane Van Buskirk, CEO, Quality Monitoring & Control.



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Balanced Flow Meter Technical Basis: Units, and Fundamental Definitions ©

Foreword

The analysis and design of the Balance Flow Meter is based on the units, fundamental definitions, and nomenclature as used in the U.S. chemical process industries, such as refineries and chemical plants. The chemical industry engineer deals primarily with internal boundary layer flow (e.g., pipe flows) with mass, momentum, energy and entropy balances. This is different from the force and momentum basis as used for open-channel, external, inviscid-streamline, or potential flow analysis for other engineering applications.

Bernoulli's Theorem

Bernoulli's theorem is used in the solution of hydraulics problems. This theorem is a special case of the first (conservation of energy) and second laws of thermodynamics in which changes in the energy content of a fluid system are balanced against one another. Although friction losses are evident as heat, the theorem is concerned only with the pressure and energy changes, and heat is considered lost energy.

$$\left. \begin{array}{l} Z_a(\text{static head}) \\ + \frac{u_a^2}{2g}(\text{velocity head}) \\ + P_a V_a(\text{pressure head}) \\ + W(\text{mechanical work}) \end{array} \right\} = \left\{ \begin{array}{l} Z_b \\ + \frac{u_b^2}{2g} \\ + P_b V_b \\ + F(\text{friction}) \end{array} \right. \quad \text{Equation 1}$$

Each item in the above equation must be expressed in the same units, i.e., feet of head or pounds per square inch, etc.

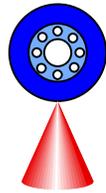
The Engineering Bernoulli Equation

The main application of the Balanced Flow Meter is for internal boundary-layer flow of single stream systems, such as the orifice, venturi or other *Bernoulli energy-head* meters. The design basis is the compressible-fluid steady-flow balance for total mass, energy and entropy. The equation uses *Bernoulli's theorem* for conservation of the various mass-specific energy-head forms and associated groupings. The *engineering Bernoulli equation* combines the energy and entropy balance into a single equation and is applicable to any incompressible, compressible or two-phase fluid. The steady state *engineering Bernoulli equation* for flow systems of any thermodynamic path is given by,

$$\Delta \left(\frac{\alpha v^2}{2g_c} \right) + \Delta \frac{gZ}{g_c} + \int \frac{dP}{\rho} + W_{shaft} + LW_{friction} = 0. \quad \text{Equation 2}$$



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and the downstream pressure at the datum is

$$P_{f2'} = P_{f2} + \frac{g}{g_c} H_{EL,2} \rho_s \quad \text{Equation 22}$$

where ρ_s is the density of the seal fluid in the lines.

The differential pressure transmitted to the differential pressure measuring device, for the same seal fluid density and provided the lines are at the same elevation, is

$$P_{f1'} - P_{f2'} = P_{f1} - P_{f2} + \frac{g}{g_c} (H_{EL,1} - H_{EL,2}) \rho_{f1} \quad \text{Equation 23}$$

The equation above is the measured differential pressure; when it is substituted into the flow equation, it results in,

$$m = \frac{\pi}{4} \sqrt{2g_c} \frac{d_F^2}{\sqrt{1-\beta^4}} \sqrt{(P_{f1'} - P_{f2'}) \rho_f} \quad \text{Equation 24}$$

This equation relates the measured differential to the flow rate, and the measured differential automatically adjusts for the potential-energy term for any orientation and gravitation field.

Fanning Friction Factor

Friction factors are used to determine viscous shear stress, pressure loss, radial velocity profiles, and momentum and kinetic energy correction factors in developed steady flow for pipes and conduits. The Fanning friction factor is used with Balanced Flow Meter (BFM) applications. Another friction factor in use is the Darcy-Weisbach. The relation between these friction factors is shown by use of the shear-stress τ representations,

$$f_{Fanning} = \frac{\tau}{(\rho v^2 / 2)} \cdot f_{Darcy-Weisbach} = \frac{4\tau}{(\rho v^2 / 2)} \Rightarrow f_{Fanning} = \frac{f_{Darcy-Weisbach}}{4} \quad \text{Equation 25}$$



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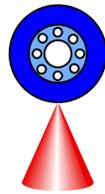
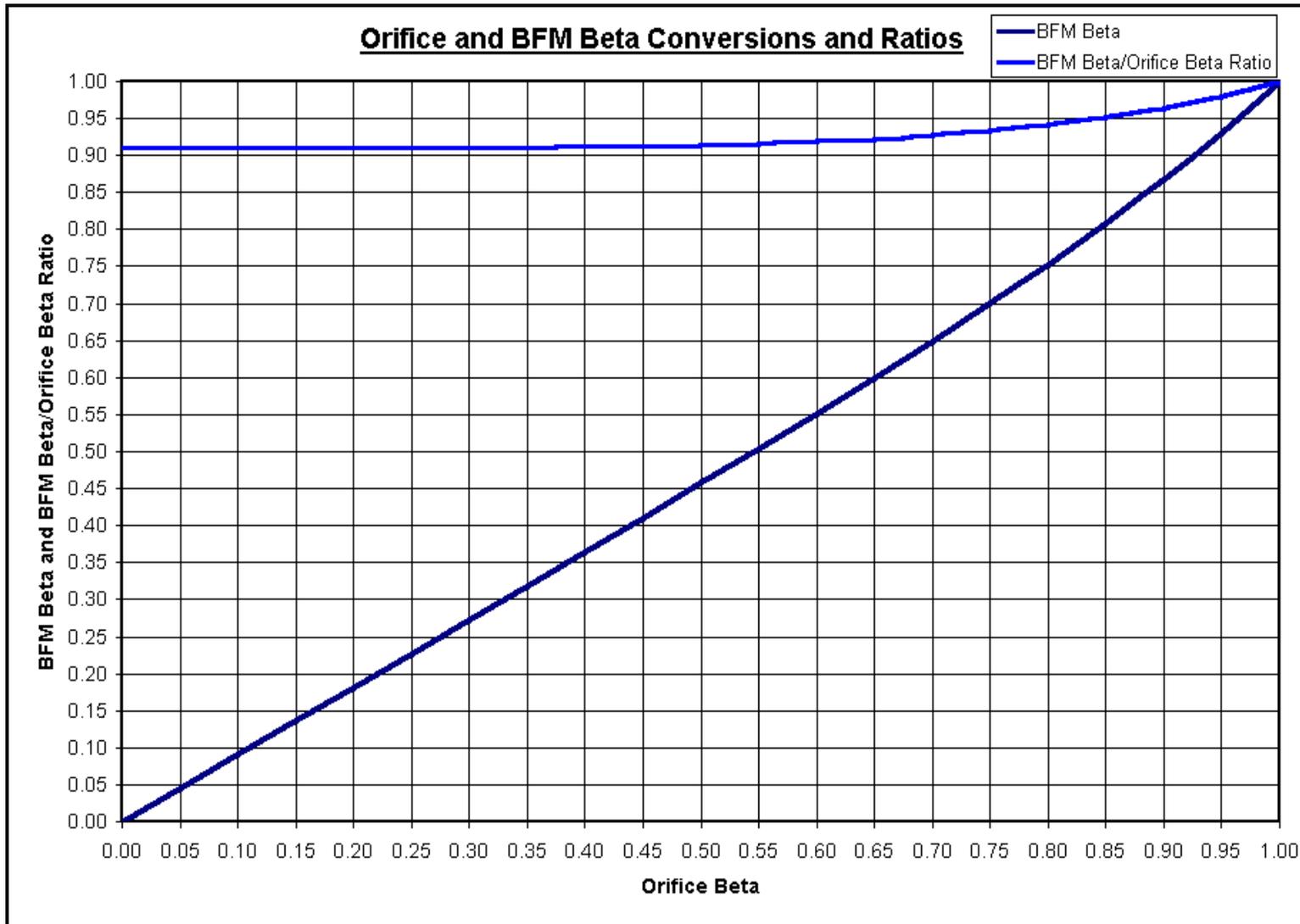


Figure 2 - Orifice to Balanced Flow Meter Beta Ratio Conversion





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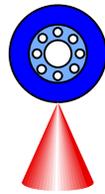
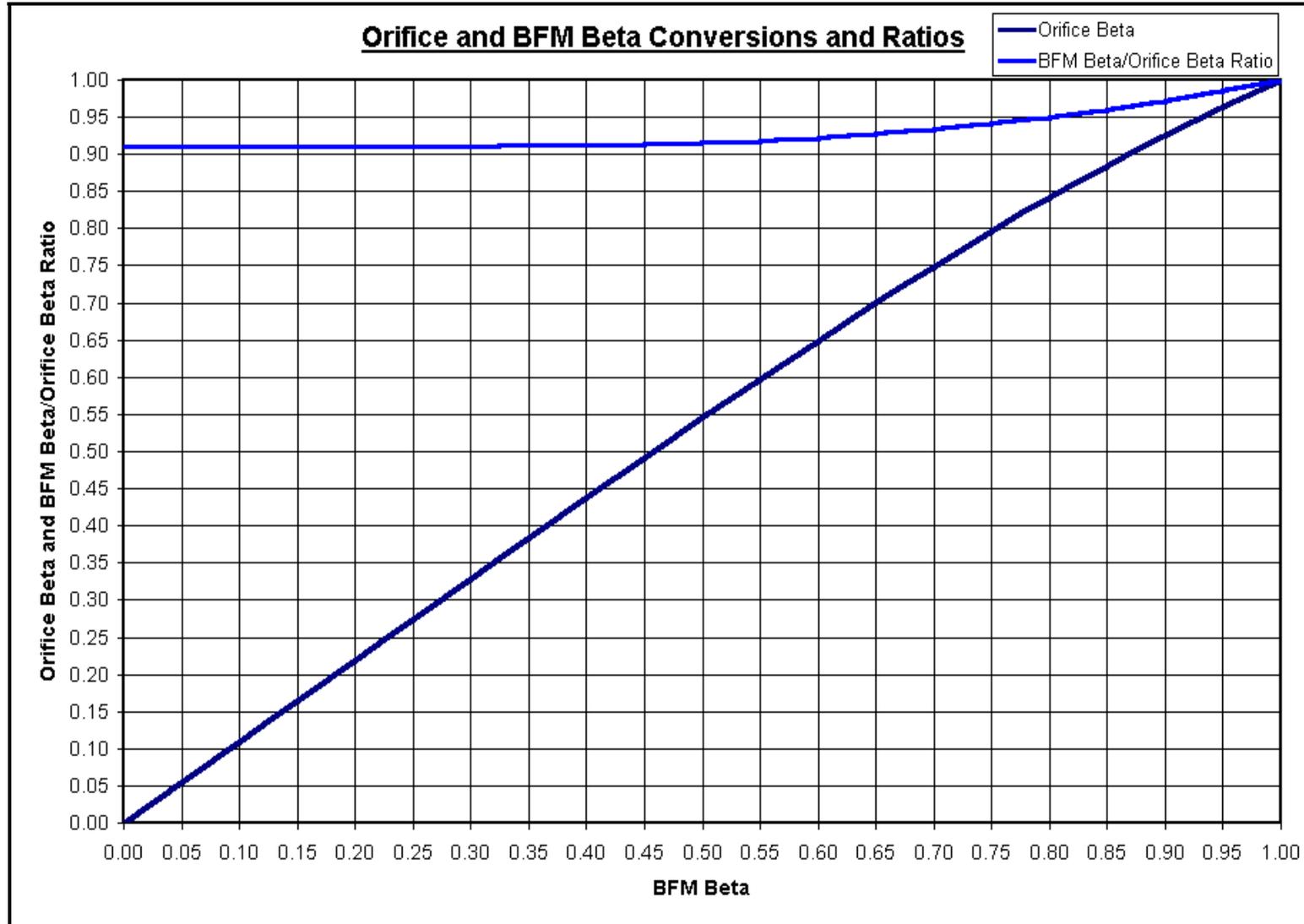


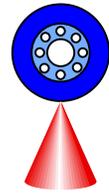
Figure 3 - Balanced Flow Meter to Orifice Beta Ratio Conversion





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PTK-802 Balanced Flow Meter Shear Stresses ©

Forward

The shear stress and potential for hole-surface erosion in a Balanced Flow Meter (BFM) is significantly reduced when compared to an orifice plate. High wall/surface shear leads to an increase in hole-erosion. The following provides the ratio of the hole-shears of the single-holed orifice plate to the multi-holed Balanced Flow Meter. This ratio provides an estimate of the reduction in erosion potential with the Balanced Flow Meter design when compared to an orifice plate with the same beta ratio. For a sharp-edged orifice, the erosion of the edge is highly increased. The Balanced Flow Meter holes for the standard plate design are square-cut to further reduce the potential for erosion.

Technical Summary

The figure on the next page shows the orifice to Balanced Flow Meter (BFM) shear stress ratio $\tau_{w,o} / \tau_{w,BFM}$. The ratio depends on the number of holes used with the particular Balanced Flow Meter design. The ratio is always greater than one (1) with the BFM plate, which shows that the erosion potential is significantly reduced with the Balanced Flow Meter. This figure demonstrates that the shear stresses are much higher with the single-holed orifice plate when compared to the BFM:

1. For the same flow,
2. Operating at the same process conditions, and
3. With the same beta β factor.

A concern with orifice plate is hole erosion that induces dimensional changes. The multi-holed Balanced Flow Meter provides a significant reduction in the shear stress and, accordingly, reduces the potential for hole-wall surface erosion. Additionally, the standard BFM holes are square, beveled, or bell-mouthed cut, *not* knife-edged. The following figure shows that the knife-edged orifice has considerable *higher* levels of shear stress.

The standard BFM design, with a 1-8 hole-layout, has 9 holes. This BFM design provides a 3 to 1 decrease in shear stress and erosion potential compared to the orifice plate. The decrease is 7 to 1 when compared to the knife-edged orifice plate.

Designs for saturated liquids and vapors use the 33-hole configuration, with a 1-16-16 hole-layout. This design provides a 5 to 1 decrease in surface shear stress. With this reduced shear stress, the potential for cavitation effects are reduced. Additionally, the multi-holed BFM essentially eliminates any vapor or liquid buildup upstream of the plate.

Further shear stress reduction is provided by using a 1-n-2n configuration as used in cavitating flows. An example is the 1-16-32 configuration, which has 49 holes. This layout configuration provides added holes for the same β specification.



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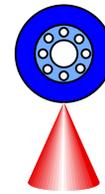
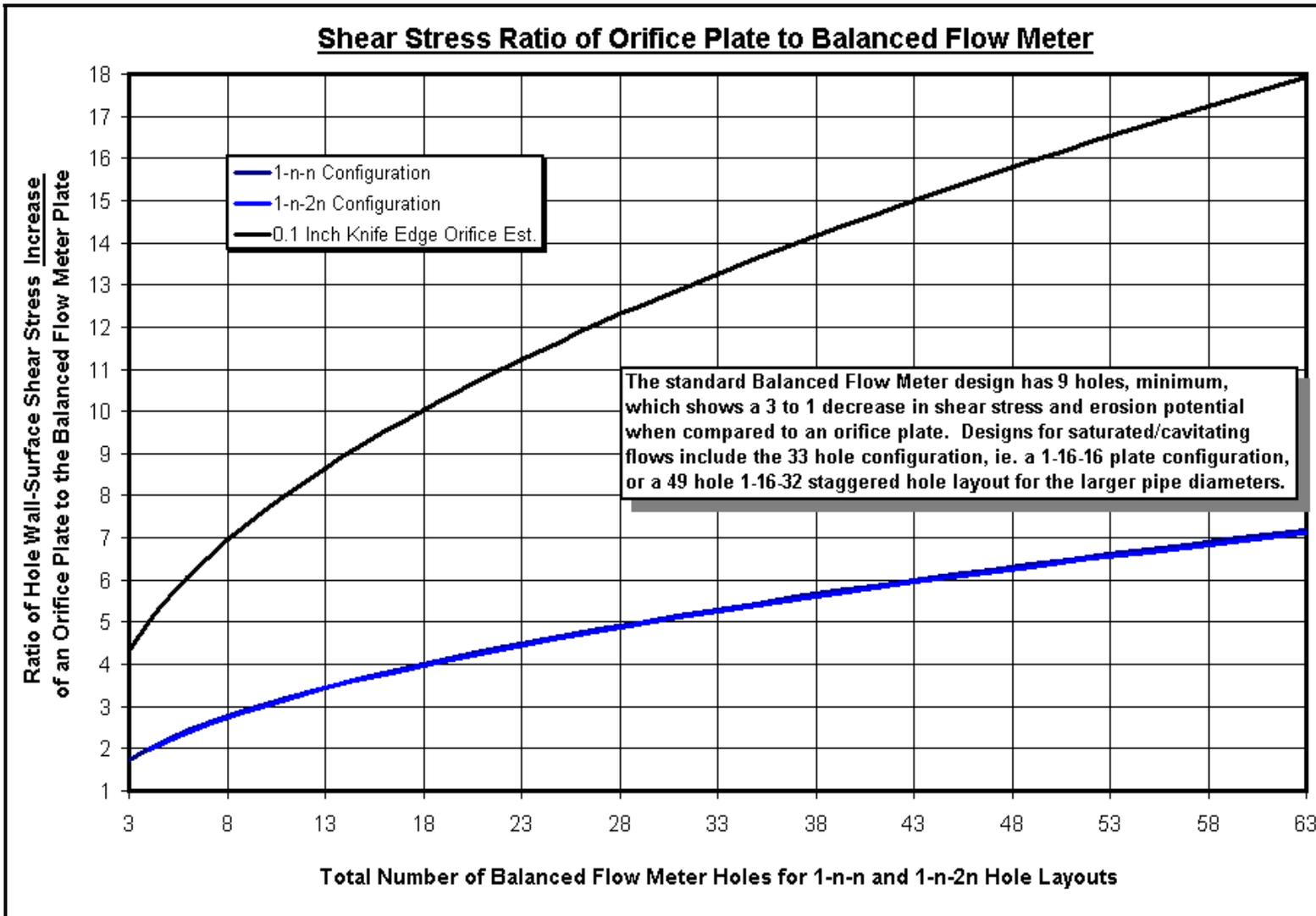
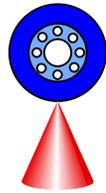


Figure 4 - Shear Stress Ratio Increase of the Orifice Plate to the BFM





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The area ratios are,

$$\frac{A_1}{A_o} = \frac{1}{1 + X_1 + X_2}, \quad \frac{A_2}{A_o} = \frac{X_1}{(1 + X_1 + X_2)}, \quad \text{and} \quad \frac{A_3}{A_o} = \frac{X_2}{(1 + X_1 + X_2)}. \quad \text{Equation 46}$$

The corresponding shear stress ratios are,

$$\frac{\tau_{w,o}}{\tau_{w,BFM}} = \left(\frac{1}{1 + X_1 + X_2} \right)^{1/2} + \left(\frac{nX_1}{1 + X_1 + X_2} \right)^{1/2} + \left(\frac{nX_2}{1 + X_1 + X_2} \right)^{1/2} = \frac{1 + (nX_1)^{1/2} + (nX_2)^{1/2}}{(1 + X_1 + X_2)^{1/2}}.$$

Equation 47

The equation above is the basis for the shear stress ratios as shown on the figure above. n is the number of holes in each ring of holes.



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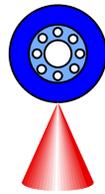
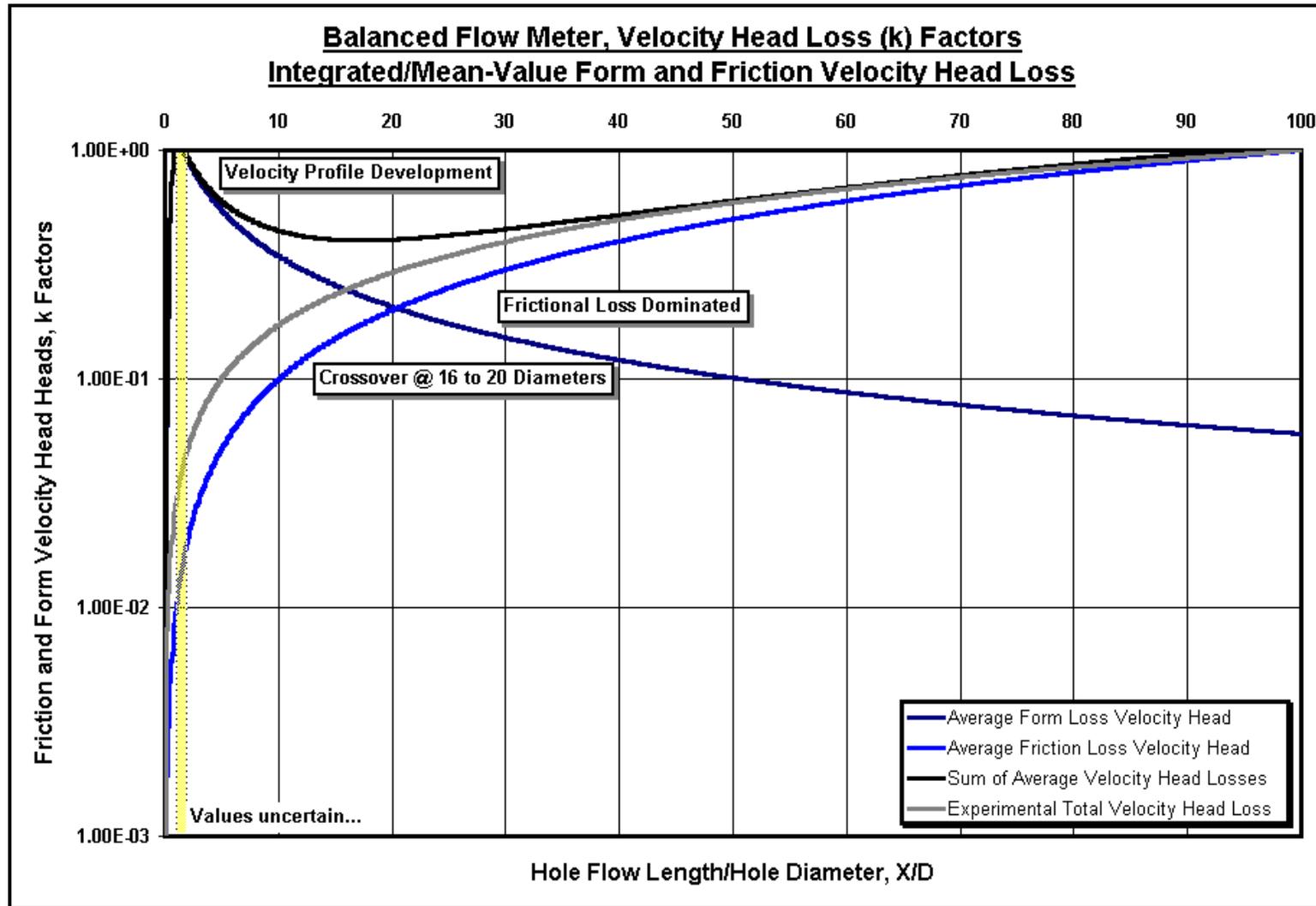


Figure 5 - Balanced Flow Meter Integrated Velocity Head Loss for Plate Thickness, Large Scale





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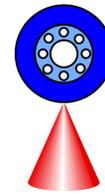
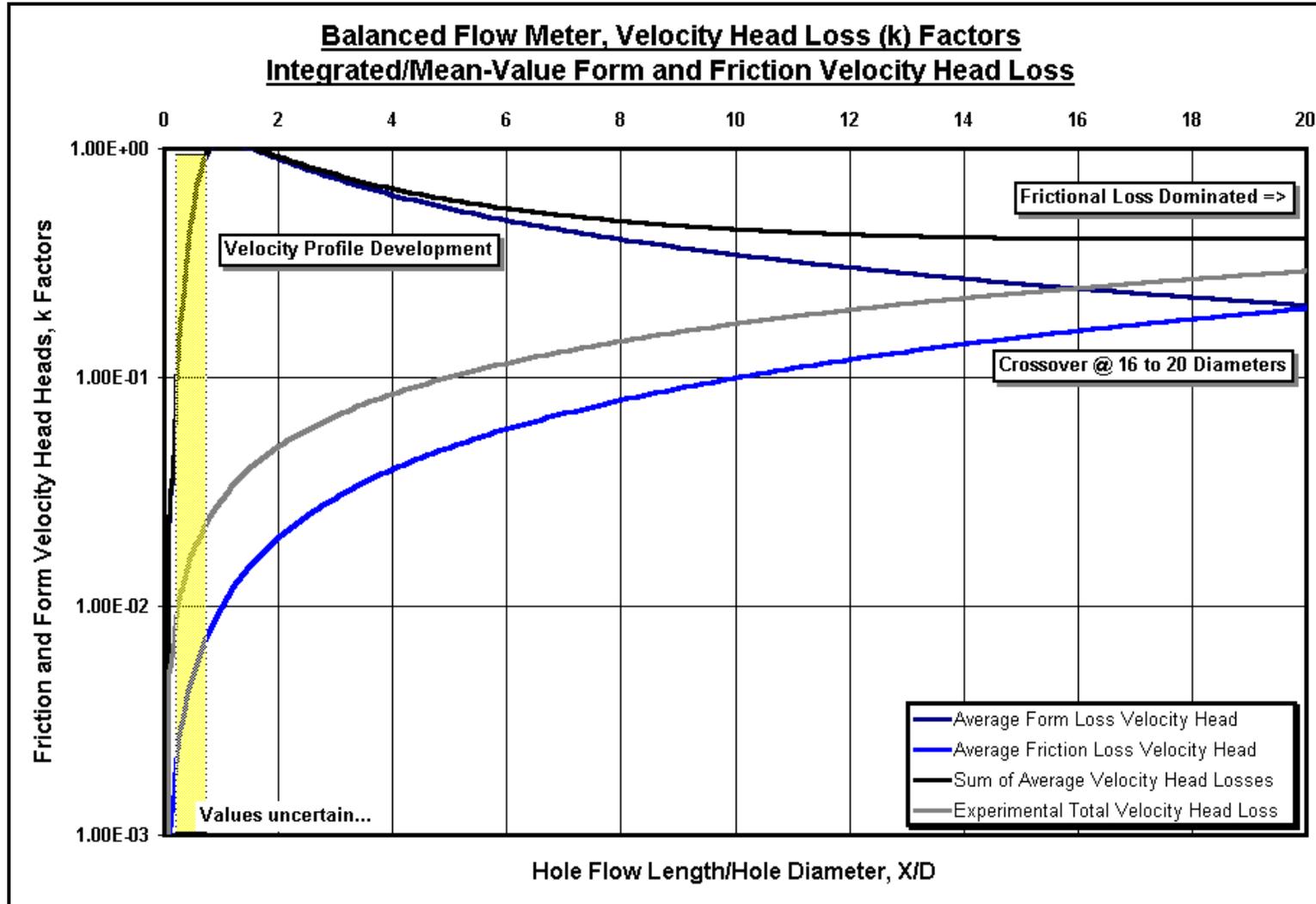


Figure 6 - Balanced Flow Meter Integrated Velocity Head Loss for Plate Thickness, Reduced Scale





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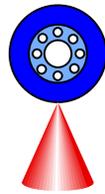
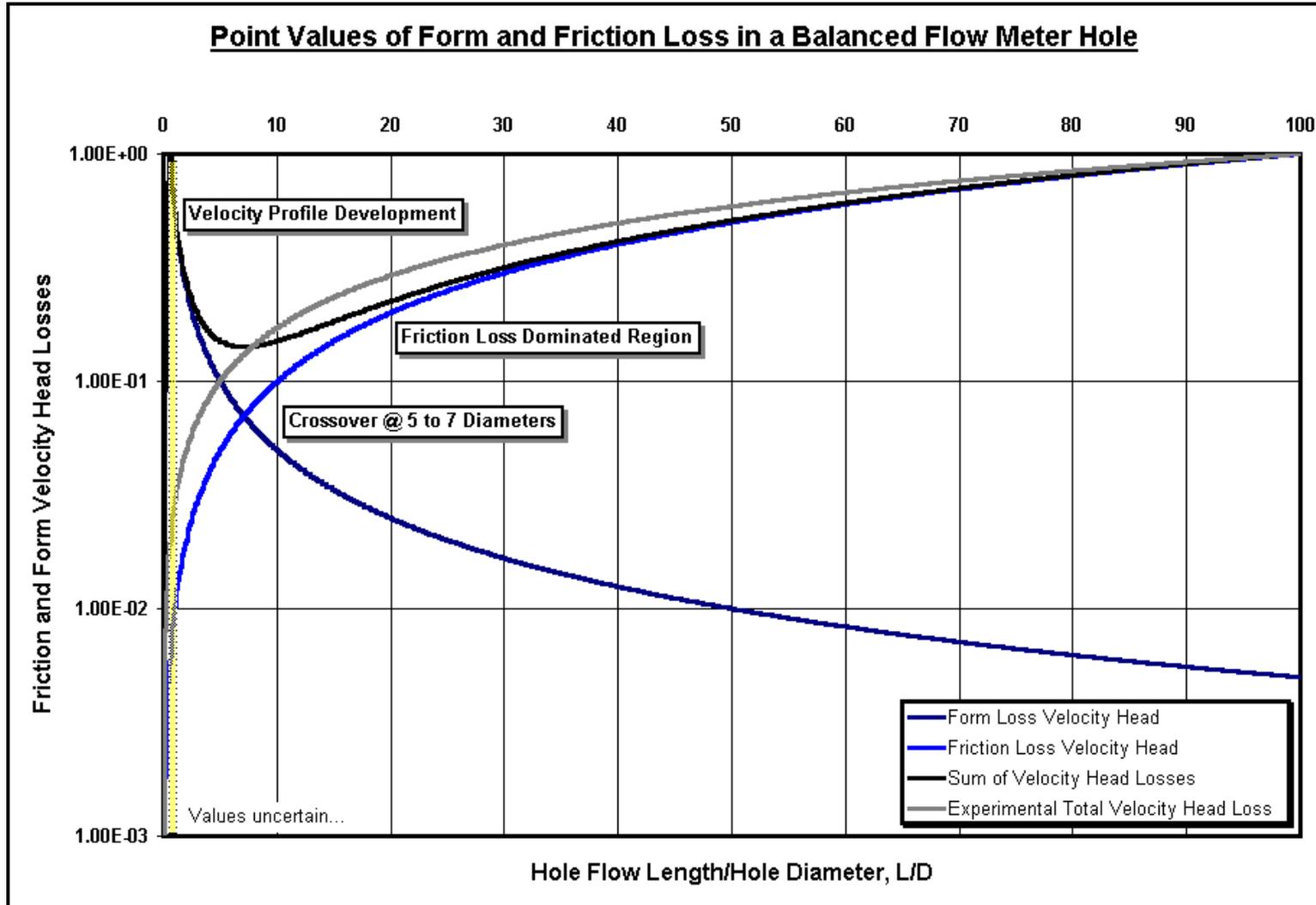


Figure 7 - Balanced Flow Meter Point Values Velocity Head Loss for Plate Thickness, Large Scale





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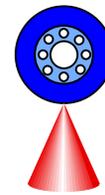
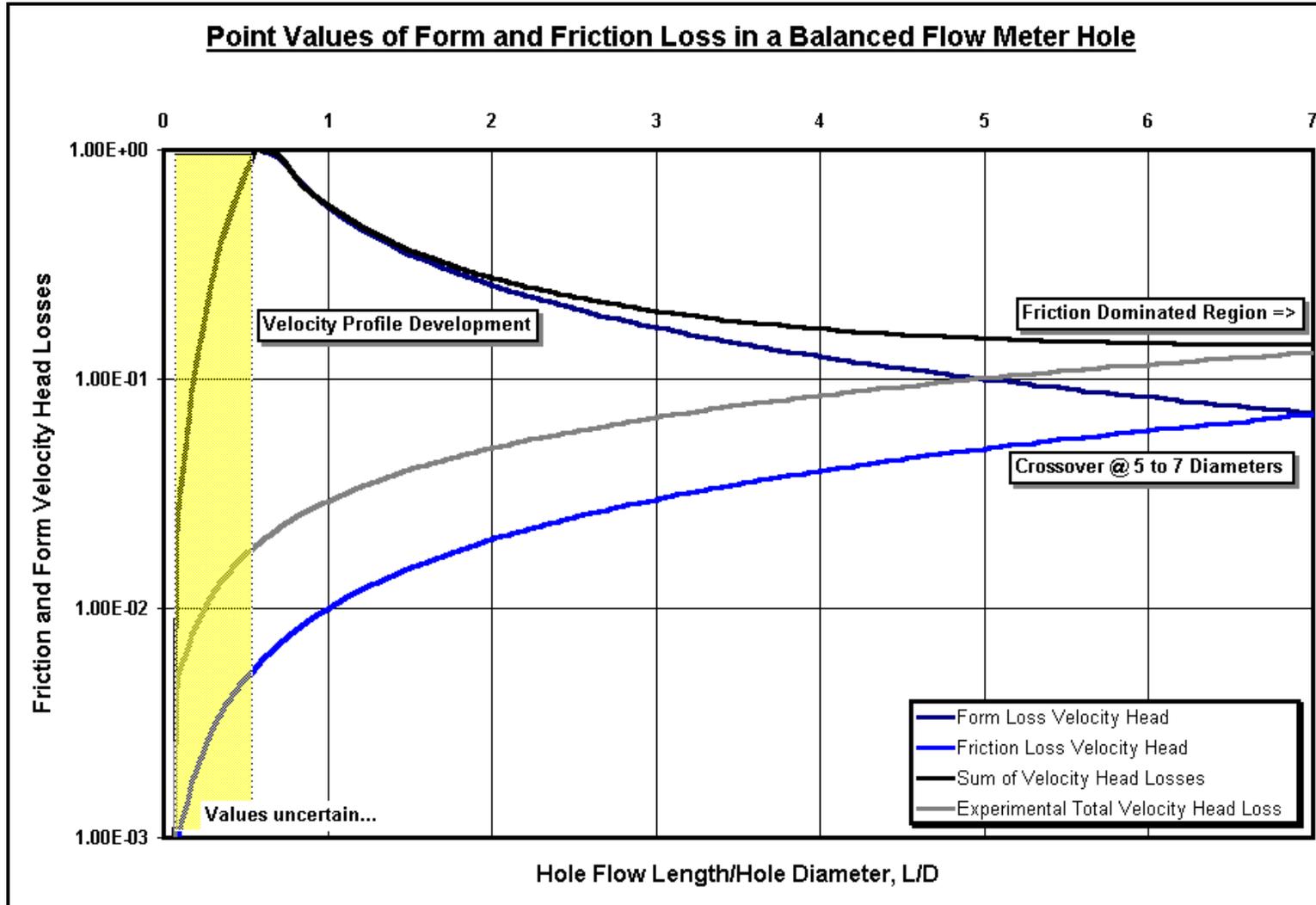


Figure 8 - Balanced Flow Meter Point Values Velocity Head Loss for Plate Thickness, Reduced Scale





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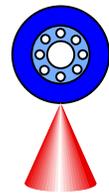
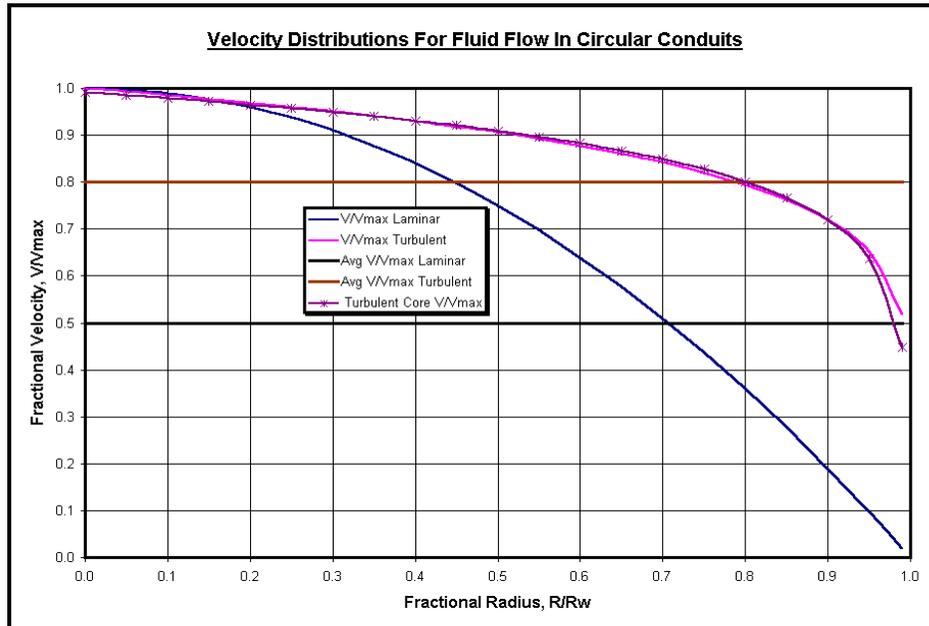


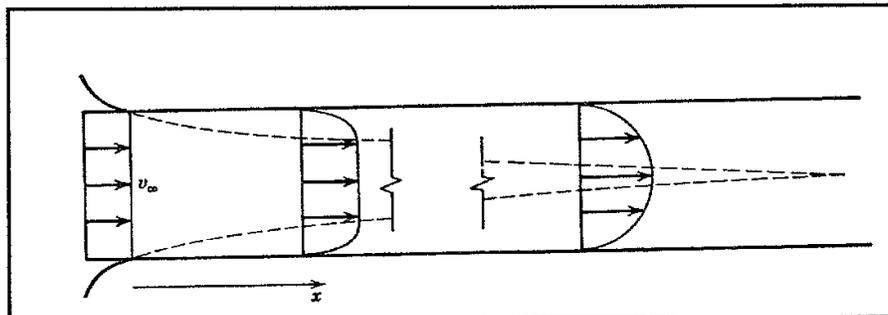
Figure 9 - Radial Velocity Distribution



The entrance velocity head losses, which result in pressure loss, are reduced with properly designed tapers and/or using nozzle or bell-mouth designs at the inlet. As an example for a tapered inlet, the following figure shows the initial velocity profile and final profile in laminar flow.

The figure below shows the velocity profiles with a tapered inlet. With this tapered inlet design, for either laminar or turbulent flows, a reduction in pressure loss due to inlet effects are reduced.

Figure 10 - Flow Entrance Effects

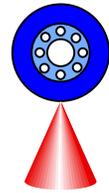


As the velocity profile shifts, from a uniform-constant velocity at the plate hole entrance to the final turbulent velocity distribution at some distance down the hole, a pressure loss occurs. These entrance effects typically last for 5 to 50 diameters dependent on Reynolds number and hole design. Frictional effects then dominate the pressure loss.



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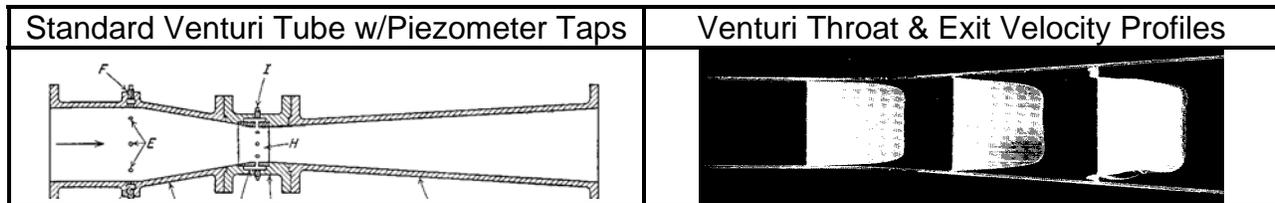
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corrects for both axial and radial gas-expansion due to eddies and associated radial-velocity effects. These concentric orifice-plate flow effects are shown in the above figure for orifice meter eddy-turbulence.

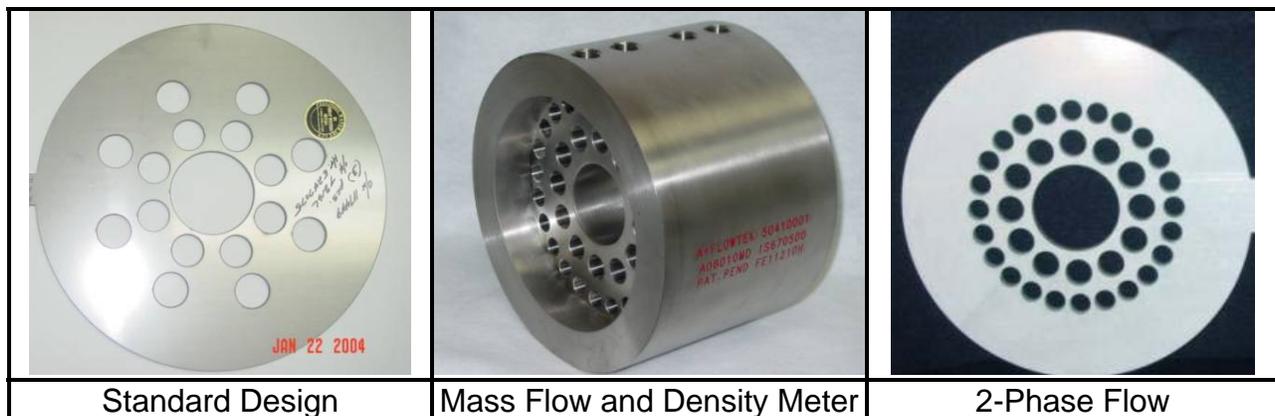
The venturi meter is well known for improved accuracy, repeatability and pressure recovery when compared to the single-holed orifice plate. The venturi γ equation is based on an adiabatic-isentropic expansion of the ideal gas since the gas expands only in the axial direction. In a venturi flow tube, there are no significant radial or axial eddies, and the gas expansion is in the axial direction only, with a well-defined and uniform velocity profile in the throat as shown below.

Figure 13 - Standard Venturi Profile



The design of the multi-holed Balanced Flow Meter (BFM) is based on the relation $\kappa \rho A v^\sigma = \text{constant}$ for each hole. This relation is optimized to provide the performance of a venturi flow tube in a single plate design. The following pictures provide various views for the configurations and wide applications of the Balanced Flow Meter plate.

Figure 14 - Balanced Flow Meter Plate Configurations and Applications

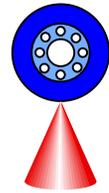


The performance of the Balanced Flow Meter follows that of the venturi meter, based on testing by NASA/Marshall Space Flight Center, and as verified by Texas A&M University, at Kingsville. The radial velocity and density variations are significantly reduced when compared to an orifice plate, as depicted on the following figure. As with the venturi tube, only axial variations of density and velocity are significant for a compressible fluid with the Balanced Flow Meter.



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PTK-805 The Balanced Flow Meter Mass, Momentum, Energy Balances, with "Real-Fluid" Equations-of-State and Multi-Sensors ©

Foreword

The design and sizing equations for the Balanced Flow Meter (BFM) are based on:

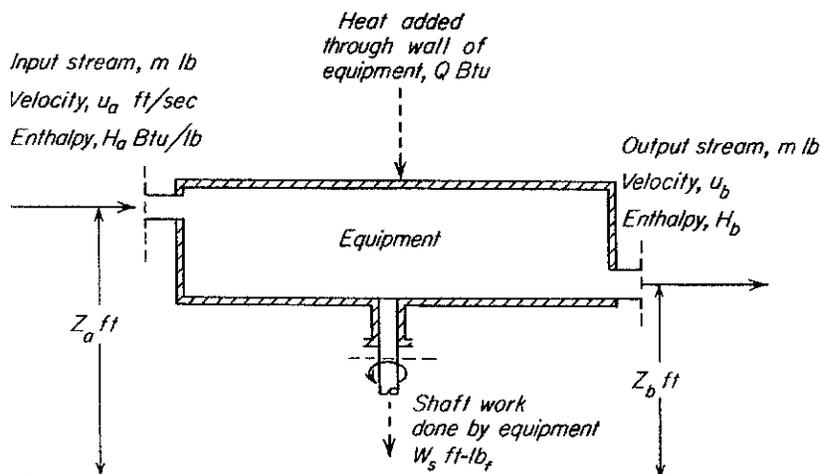
1. The Laws of Conservation of Mass and Energy,
2. Newton's Laws of Force, Momentum and Acceleration,
3. The Laws of Thermodynamics,
4. The Law of Corresponding States, for an Equation-of-State (EoS) with real-fluid physical, thermodynamic, and transport properties,
5. Boundary Layer Theory, with
6. The Balanced Flow Meter multi-holed sizing and layout relation, $\kappa\rho Av^\sigma = \text{constant}$ for each hole, that substantially reduce eddies and radial velocity and density effects.

Coupled with powerful non-linear model constrained statistical methods, such as Modified Iterative Measurement Test (MIMT[©]), with multiple pressure-temperature sensors, flow measurement errors may be reduced to negligible values. Also, pressure, temperature and flow sensor accuracy, repeatability and performance are monitored and verified. Continuous on-line calibration is provided with the MIMT[©] method.

Technical Basis

A steady flow process in which a single stream of any fluid material is flowing through any fixed-volume device is shown below.

Figure 16 - Diagram for a Steady Flow Process





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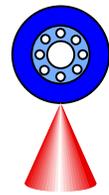


Figure 17 - Head-Meter Enthalpy-Entropy Plot

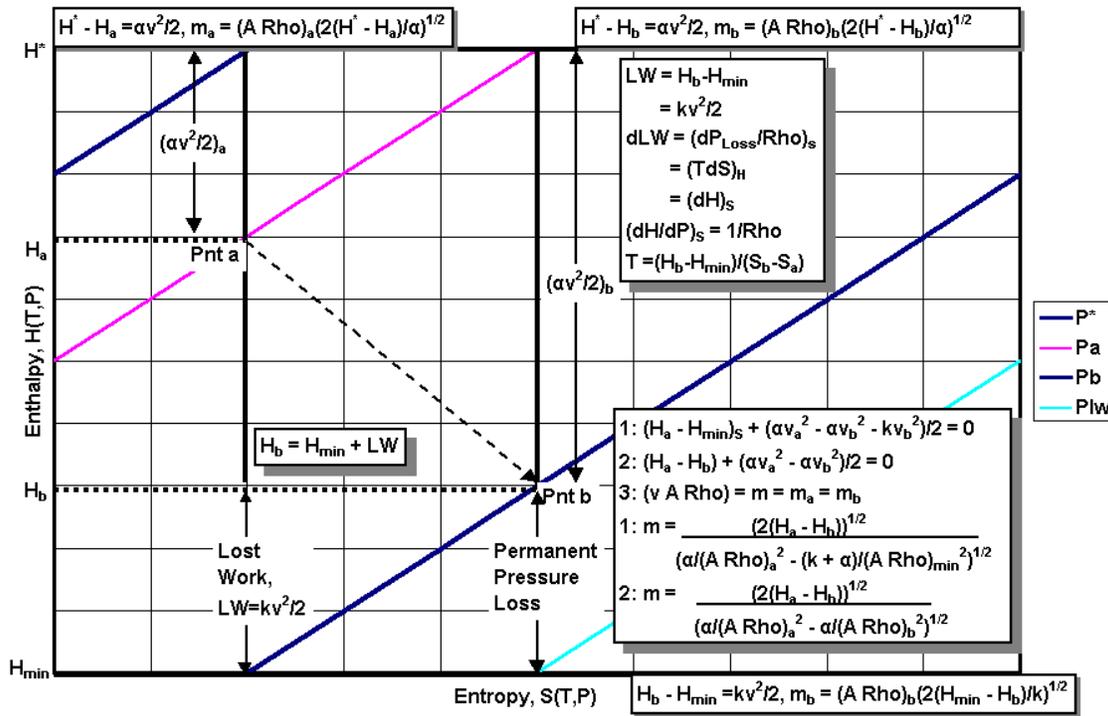
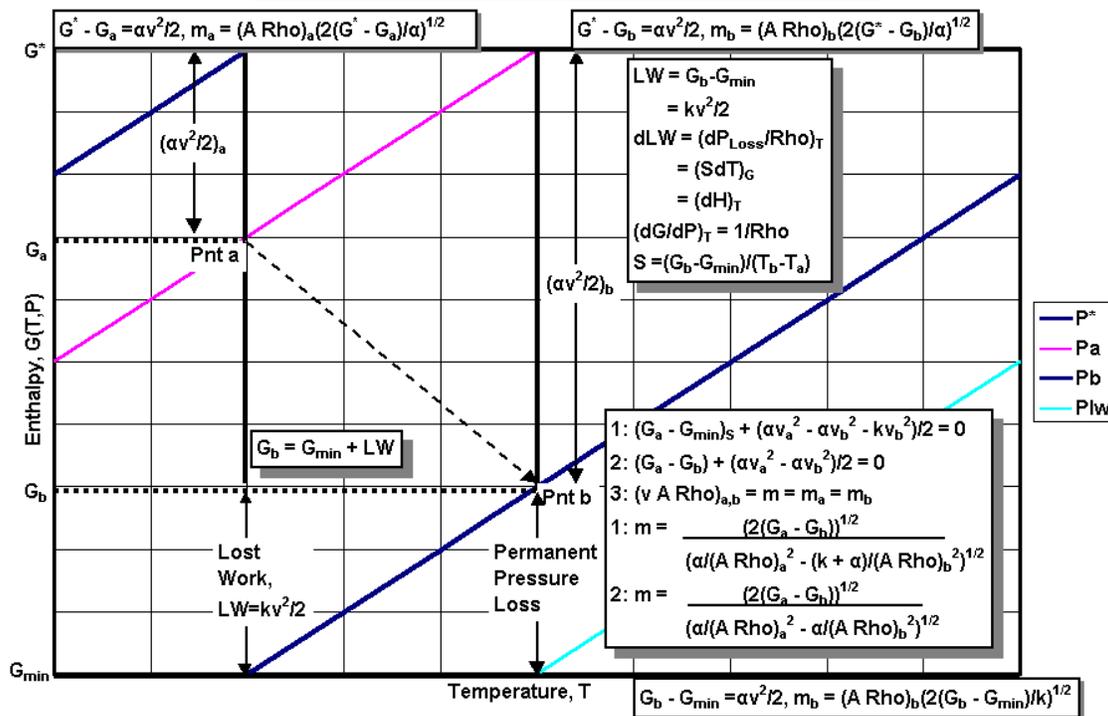


Figure 18 - Head-Meter Gibbs Free Enthalpy-Temperature Plot



Note: Use absolute values with $\sqrt{|\Delta' s|}$.



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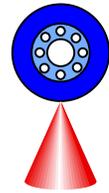


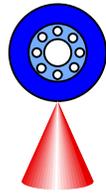
Table 5 - Summary Balanced Flow Meter Mass, Momentum and Energy Equations

| FLUID FLOW MASS, MOMENTUM, ENERGY AND ENERGY RATE EQUATIONS | | |
|---|---|---|
| Balance | Special Form | Steady State |
| Mass | Single Stream | $\Delta m = 0, \quad m_a = m_b = \text{Const.}$ $(\rho Av)_a = (\rho Av)_b = \text{Const.}$ $v = \frac{\vec{G}_z(P,T)\Re T}{P}$ at any a or b, & $\vec{G} \equiv m/A$ |
| Momentum | Control Volume of mass m_{tot} | $F = -\Delta \left(\frac{1}{g_c} \frac{\langle v^2 \rangle}{\langle v \rangle} m + PA \right) + m_{tot} \frac{g}{g_c}$ |
| Entropy | $E_f = \int_a^b \frac{dP_f}{\rho} \equiv \frac{kv^2}{2g_c}$ | $\int_a^b T dS = Q + E_f : \Delta S = \frac{Q}{T_{ref}} + \frac{E_f}{T_{ref}}$ |
| Total Energy-Bernoulli Theorem Head-Form | Any Path | $\Delta \left(U + P/\rho + \frac{1}{2g_c J} \frac{\langle v^3 \rangle}{\langle v \rangle} + \Phi_{PE} \right) = Q - W_{Shaft}$ |
| Mechanical Energy-Bernoulli Theorem Head-Form | Isothermal | $\Delta \left(\frac{1}{2g_c J} \frac{\langle v^3 \rangle}{\langle v \rangle} + \Phi_{PE} + G_T \right) + W_{Shaft} + E_f = 0$ |
| | Isentropic | $\Delta \left(\frac{1}{2g_c J} \frac{\langle v^3 \rangle}{\langle v \rangle} + \Phi_{PE} + H_S \right) + W_{Shaft} + E_f = 0$ |
| Bernoulli Equation Combined Entropy & Enthalpy Form | Path Dependent | $\Delta \left(\frac{1}{2g_c J} \frac{\langle v^3 \rangle}{\langle v \rangle} \right) + \Delta \Phi_{PE} + \int_a^b \frac{dP}{\rho} + W_{Shaft} + E_f = 0$ |
| Entropy Rate | Any Path | $\Delta(S(\rho Av)) = \frac{\dot{Q}}{T_\sigma} + \frac{mE_f}{T_\sigma}$ |
| Total Energy Rate | Any Path | $\Delta \left[\left(U + P/\rho + \frac{1}{2g_c J} \frac{\langle v^3 \rangle}{\langle v \rangle} + \Phi_{PE} \right) (\rho Av) \right] = \dot{Q} - \dot{W}_{Shaft}$ |
| Mechanical Energy Rate | Isothermal | $\Delta \left[\left(\frac{1}{2g_c J} \frac{\langle v^3 \rangle}{\langle v \rangle} + \Phi_{PE} + G_T \right) (\rho Av) \right] + \dot{W}_{Shaft} + mE_f = 0$ |



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| | | |
|---|----------------|---|
| Mechanical Energy Rate | Isentropic | $\Delta \left[\left(\frac{1}{2g_c J} \frac{\langle v^3 \rangle}{\langle v \rangle} + \Phi_{PE} + H_S \right) (\rho A v) \right]$ $+ \dot{W}_{Shaft} + mE_f = 0$ |
| Bernoulli Equation Combined Entropy & Enthalpy Rate | Path Dependent | $\Delta \left(\frac{(\rho A v) \langle v^3 \rangle}{2g_c J} \right) + \Delta m \Phi_{PE} + \int_a^b \frac{(\rho A v) dP}{\rho}$ $+ \dot{W}_{Shaft} + mE_f = 0$ |



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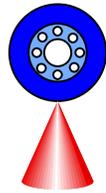


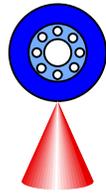
Table 6 - Summary BFM Equations-of-State & Thermodynamic Paths

| FLUID-FLOW EQUATIONS-OF-STATE AND THERMODYNAMIC PATHS | | |
|--|---|--|
| Fluid Type | Basic Form | Thermodynamic Path |
| Ideal Gas $\rho = \frac{\mathfrak{R}T}{P}$ $\lambda = \Theta \delta = \frac{C_p}{C_v}$ $\left(\frac{dU}{dP}\right)_T = 0$ | Isothermal $P / \rho = \text{Const.}$ | $\Delta G_T = \int_a^b \frac{dP}{\rho} = \mathfrak{R}T \ln \left(\frac{P_b}{P_a} \right)$ |
| | Isentropic $P / \rho^\lambda = \text{Const.}$ $T / \rho^{\lambda-1} = \text{Const.}$ | $\Delta H_S = \int_a^b \frac{dP}{\rho} = \int_a^b C_p dT = \frac{\lambda}{\lambda-1} \left[\left(\frac{P_b}{P_a} \right)^{\frac{\lambda-1}{\lambda}} - 1 \right] \frac{P_a}{\rho_a}$ |
| | Bernoulli $P / \rho^\delta = \text{Const.}$ $T / \rho^{\delta-1} = \text{Const.}$ | $\int_a^b \frac{dP}{\rho} = \frac{\delta}{\delta-1} \left[\left(\frac{P_b}{P_a} \right)^{\frac{\delta-1}{\delta}} - 1 \right] \frac{P_a}{\rho_a}$ |
| | Any Path | $\Delta H = \int_a^b dU + \int_a^b d \left(\frac{P}{\rho} \right) = \int_a^b T dS + \int_a^b \frac{dP}{\rho} = \hat{C}_{p,lm} (T_b - T_a)$ |
| Incompressible Liquid $\rho_a = \rho_b$ $C_p = C_v = \hat{C}$ | Isothermal | $\Delta G_T = \int_a^b \frac{dP}{\rho} = \frac{P_b - P_a}{\hat{\rho}}$ |
| | Isentropic | $\Delta H_S = \int_a^b \frac{dP}{\rho} = \frac{P_b - P_a}{\hat{\rho}}$ |
| | Bernoulli | $\int_a^b \frac{dP}{\rho} = \frac{P_b - P_a}{\hat{\rho}}$ |
| | Any Path | $\Delta H = \int_a^b dU + \int_a^b d \left(\frac{P}{\rho} \right) = \hat{C} (T_b - T_a) + \frac{P_b - P_a}{\hat{\rho}}$ |
| Single Phase $\ln \left(\frac{\rho_b}{\rho_a} \right) = \kappa (P_b - P_a) - \varepsilon (T_b - T_a)$ $\Delta T, \Delta P \leq \delta X \Rightarrow$ $\kappa = \text{Const.}$ $\varepsilon = \text{Const.}$ $\left(\frac{\partial \rho}{\rho} \right)_s = \kappa_s dP$ | Isothermal | $\Delta G_T = \int_a^b \frac{dP}{\rho} = \frac{1}{\kappa} \left(\frac{1}{\rho_b - \rho_a} \right)$ |
| | Isentropic | $\Delta H_S = \int_a^b \frac{dP}{\rho} = \frac{1}{\kappa} \left(\frac{1}{\rho_b - \rho_a} - \varepsilon (T_b - T_a) \right)$ |
| | Bernoulli | $\int_a^b \frac{dP}{\rho} = \frac{1}{\kappa} \left(\frac{1}{\rho_b - \rho_a} - \varepsilon (T_b - T_a) \right) = \frac{\kappa_s}{\rho_a} (e^{\kappa_s \Delta P} - 1)$ |
| | Any Path | $\Delta H = \int_a^b dU + \int_a^b d \left(\frac{P}{\rho} \right) = C_{v,lm} (T_b - T_a) + \frac{P_b}{\rho_b} - \frac{P_a}{\rho_a}$ |



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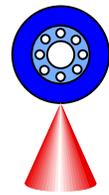


| | | |
|--|------------|---|
| Single Phase $\rho = \frac{z(T, P)\mathcal{R}T}{P}$ $P/\rho^\delta = \text{Const.}$ | Isothermal | $\Delta G_T = \int_a^b \frac{dP}{\rho} = \text{Ln}\left(\frac{P_b}{P_a}\right) \frac{(P_b/\rho_b - P_a/\rho_a)}{\text{Ln}\left(\frac{P_b/\rho_b}{P_a/\rho_a}\right)}$ |
| | Isentropic | $\Delta H_S = \int_a^b \frac{dP}{\rho} = \text{Ln}\left(\frac{P_b}{P_a}\right) \frac{(P_b/\rho_b - P_a/\rho_a)}{\text{Ln}\left(\frac{P_b/\rho_b}{P_a/\rho_a}\right)}$ |
| | Bernoulli | $\int_a^b \frac{dP}{\rho} = \text{Ln}\left(\frac{P_b}{P_a}\right) \frac{(P_b/\rho_b - P_a/\rho_a)}{\text{Ln}\left(\frac{P_b/\rho_b}{P_a/\rho_a}\right)}$ |
| | Any Path | $\Delta H = \int_a^b dU + \int_a^b d\left(\frac{P}{\rho}\right) = C_{v,lm}(T_b - T_a) + \frac{P_b}{\rho_b} - \frac{P_a}{\rho_a}$ |
| Real Fluid $\rho = \frac{z(T, P)\mathcal{R}T}{P}$ $\text{properties} = f(T, P)$ $\Delta H = \Delta H_S / \eta_S$ $\phi_S, \eta_S = f(\text{path})$ | Isothermal | $\Delta G_T = (H_b - H_a)_T - T(S_b - S_a)_T$ |
| | Isentropic | $\Delta H_S = (U_b - U_a)_S + ((P/\rho)_b - (P/\rho)_a)_S$ |
| | Bernoulli | $\int_a^b \frac{dP}{\rho} = \text{Ln}\left(\frac{P_b}{P_a}\right) \frac{(P_b/\rho_b - P_a/\rho_a) / \phi_S}{\text{Ln}\left(\frac{P_b/\rho_b}{P_a/\rho_a}\right)}$ |
| | Any Path | $\Delta H = \int_a^b dU + \int_a^b d\left(\frac{P}{\rho}\right) = U_b - U_a + \frac{P_b}{\rho_b} - \frac{P_a}{\rho_a}$ |



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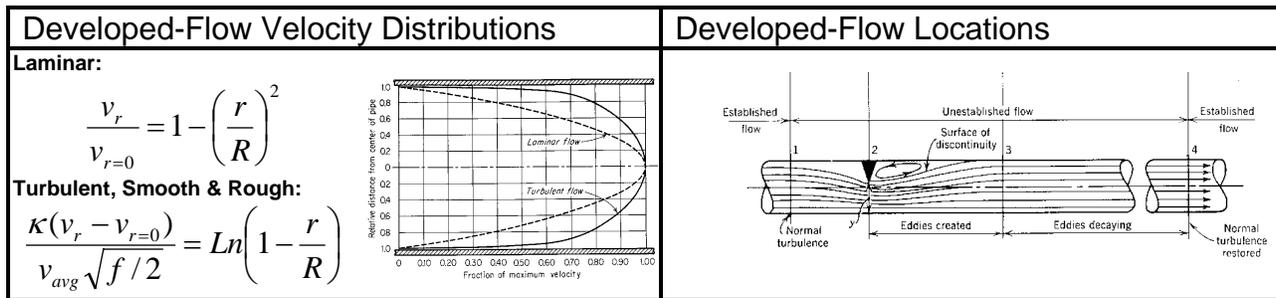
PTK-806 Balanced Flow Meter, Kinetic and Momentum Correction Factors ©

$u^2/2g_c$ is the kinetic energy of a pound mass of fluid flowing with the kinetic velocity u across the pipe flow area. The kinetic velocity may not be the same as the average flow velocity v determined by mass balance. The relation is $u^2 = \alpha v^2$, where α is the kinetic energy correction factor. α is calculated by the kinetic energy per pound E_K of fluid by,

$$E_K = \frac{\dot{E}_K}{m} = \frac{u^2}{2g_c} = \frac{\alpha v_{avg}^2}{2g_c} = \frac{1}{2g_c} \frac{\langle v_r^3 \rangle}{\langle v_r \rangle} = \frac{1}{2g_c} \frac{\int^A v_r^3 dA}{\int^A v_r dA} = \frac{1}{2g_c} \frac{\int_0^{2\pi R} \int_0^{2\pi R} v_r^3 r dr d\theta}{\int_0^{2\pi R} \int_0^{2\pi R} v_r r dr d\theta} \quad \text{Equation 65}$$

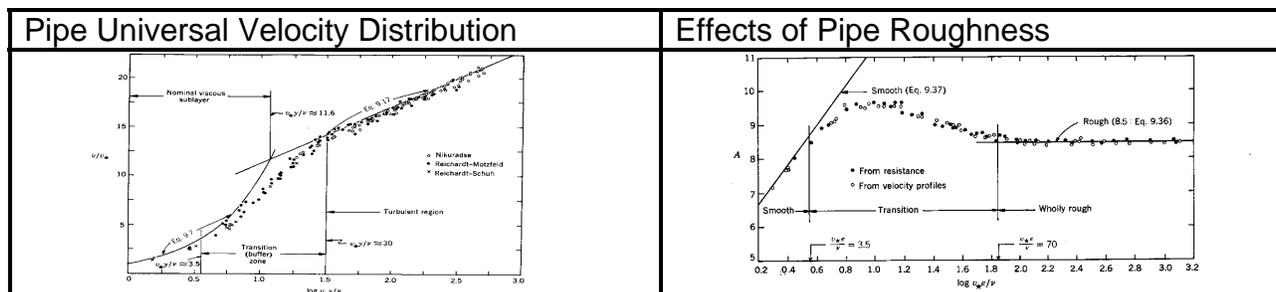
v_{avg} is calculated from $v_{avg} = m/\rho A$. For this integration, the local average velocity v_r , for axial velocity \bar{x} flow, must be known as a function of radius r and angle θ within the pipe. A similar procedure is used for the momentum correction factor β . The velocity profiles v_r for developed laminar or turbulent flows are shown in the following figure.

Figure 19 - Developed Flow Velocity Distributions and Locations



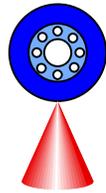
For standard head-meter design, fully developed flow is normally taken as 5 to 10 diameters, or more, upstream and downstream of the plate or constriction. An additional 5 to 10 diameters are required upstream for any fittings that may be in the pipe. The locations for established flow are shown as points 1 and 4 in the above figure. The Balanced Flow Meter design provides operations below 2 pipe diameters.

Figure 20 - Universal Velocity Distribution and Effects of Pipe Roughness





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| Fanning Friction Factor | |
|--|---|
| $f = \frac{16}{\text{Re}}$ | $\frac{1}{\sqrt{f/2}} = \frac{1}{\kappa} \text{Ln} \left(\text{Re} \sqrt{\frac{f}{8}} \right) - \frac{3}{2\kappa} + \varepsilon$ |
| $\text{Re} \equiv \frac{D\vec{G}}{\mu} \cong \text{const.}, \vec{G}_{@ \text{any } \bar{x}} \equiv m/A = \text{const.}, v_{\text{max}} @ r=0, \kappa \cong 0.407, \varepsilon \cong 5.67.$ | |
| Friction factors applicable to smooth pipes and subsonic flows. | |



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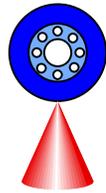
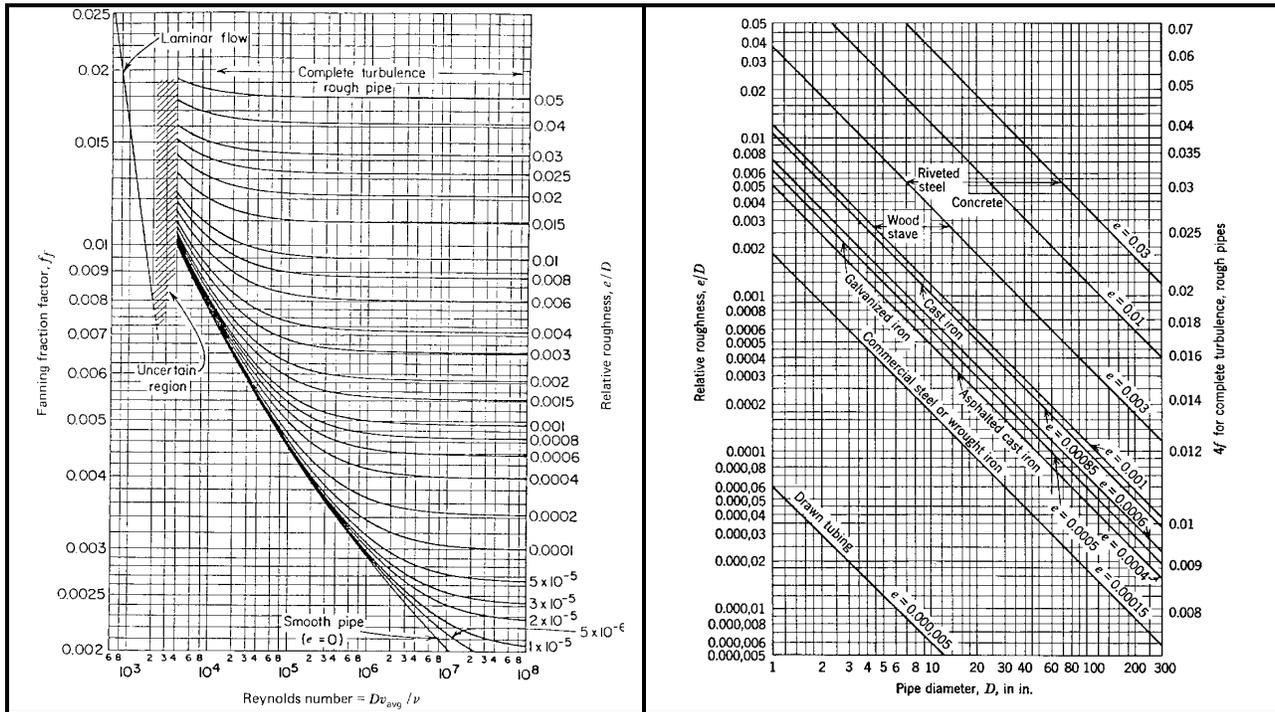


Table 11 - Turbulent Flow Properties for Rough Pipes & Tubes

| Velocity Distributions and Velocity Ratio | |
|--|---|
| Radial Velocity, v_{\max} @ $r = 0$ | $\frac{v_r}{v_{r=0}} \cong \left(1 - \frac{r}{R}\right)^{\frac{1}{n}}$ |
| Average Velocity | $\frac{v_{\text{avg}}}{v_{r=0}} = \frac{1}{1 + 10\kappa\sqrt{(f/2)}} = \frac{2n^2}{(n+1)(2n+1)}$ |
| Friction Factors with Relative Roughness Factor (e/D) | |
| Turbulent Flow, Fully Rough $(D/e)/(Re\sqrt{f}) \leq 0.01$ | $\frac{1}{\sqrt{2f}} = \text{Log}\left(\frac{D}{e}\right) + B$ |
| Turbulent Smooth/Rough Transition Flow | $\frac{1}{\sqrt{2f}} = \text{Log}\left(\frac{D}{e}\right) - \text{Log}\left(\frac{(\varepsilon - 1)(D/e)}{Re\sqrt{f}} + 1\right) + B$ |
| $f \pm 1.5\%, 10^8 \geq Re \geq 10^4, 0.05 \geq e/D \geq 0 \Rightarrow$ | $\frac{1}{\sqrt{f}} = -3.6\text{Log}\left[\frac{6.9}{Re} + \left(\frac{e/D}{3.7}\right)^{10}\right]$ |
| $Re \cong \frac{D\bar{G}}{\mu} \cong \text{const.}, \bar{G}_{@ \text{any } \bar{x}} \cong m/A = \text{const.}, \kappa \cong 0.407, \varepsilon \cong 5.67, B = 2.28, \text{Log} = \text{Log}_{10}$ | |

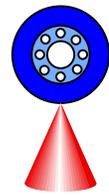
Figure 21 - Fanning Friction Factors and Roughness Parameters for Pipes & Tubes, e in feet





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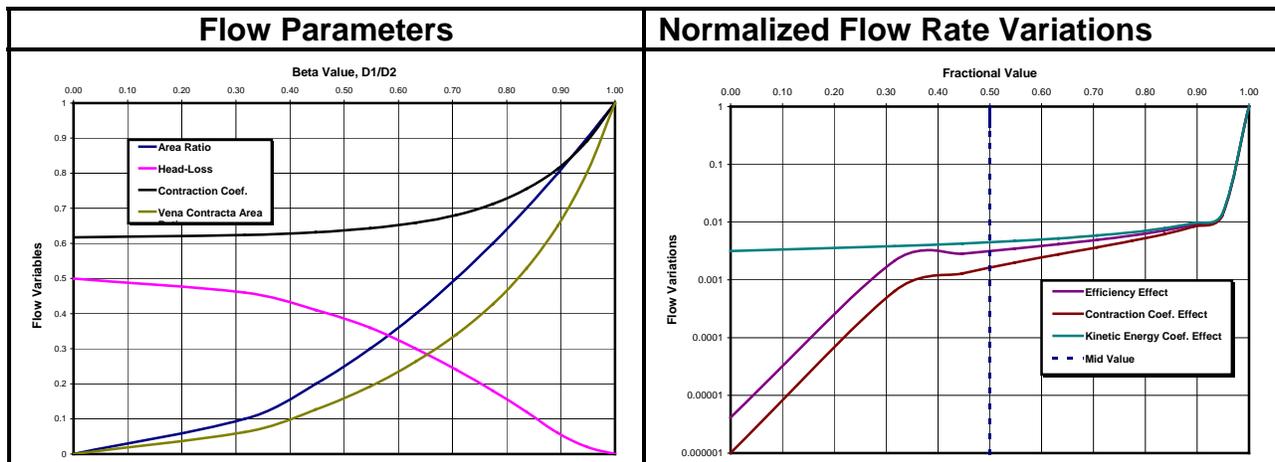


$$m = \rho_a A_a \sqrt{\frac{(2g_c J\Delta H + 2g\Delta Z \cos(\phi))\eta_{eff}}{\left(\left(\frac{1}{C_c \beta^2} \frac{\rho_a}{\rho_b}\right)^2 - \alpha_a \eta_{eff}\right)}} = \frac{C_D \gamma A_b}{\sqrt{1-\beta^4}} \sqrt{\rho_a (2g_c \Delta P + 2g\rho_a \Delta Z \cos(\phi))}$$

Equation 72

The following figures provide graphs of flow parameters versus the beta ratio and show the effects of parameter variations on flow rate. The normalized flow rate variations plot show that flow variations are relatively constant for fractional values from ~0.3 to ~0.9. For values above ~0.9, the efficiency, contraction, and kinetic energy effects have the same magnitude value. Fractional values below ~0.3 show an increasing divergence.

Figure 22 - Flow Parameters and Normalized Flow Rate Variations



Discharge coefficients are affected by parameter variations as shown above. For improved flow measurement accuracy, the discharge coefficient in the turbulent flow regime can be corrected. The C_D flow adjustment follows the general equation,

$$C_D = C_{D,Re=\infty} + \frac{b}{Re_D^n} \cong f(Re, \beta, D) .$$

Equation 73

This equation applies to all head-meters, of any fluid and phase. The coefficients $C_{D,Re=\infty}$, b and n are listed in many publications. As an example, for a venturi meter, b and n are taken as ~zero (0). From testing over a wide range of flows and beta factors, the Balanced Flow Meter (BFM) has similar b and n constants. (However, for the BFM, $C_{D,Re=\infty} = f(\beta)$ as a secondary affect). The following figures provide graphs of the orifice plate and multi-holed orifice plate discharge coefficients C_D . The multi-holed plate layout is based on a constant pitch to diameter ratio; the BFM is not.



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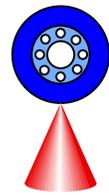
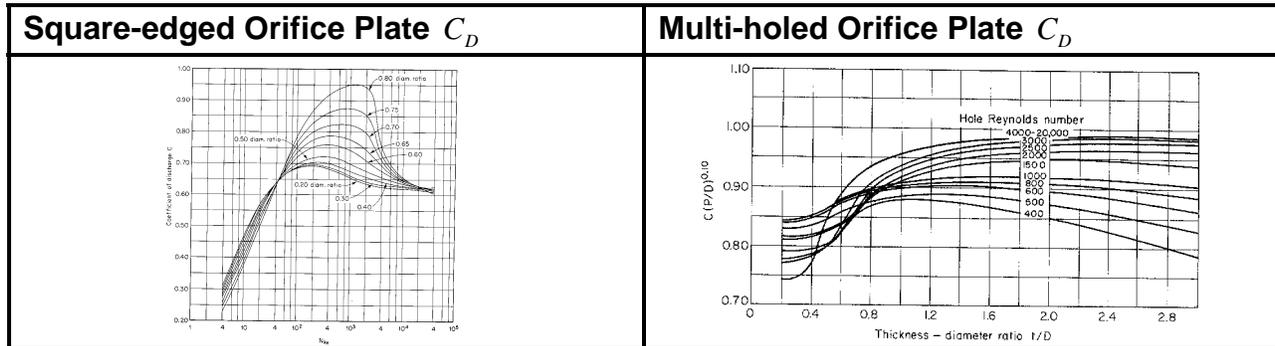


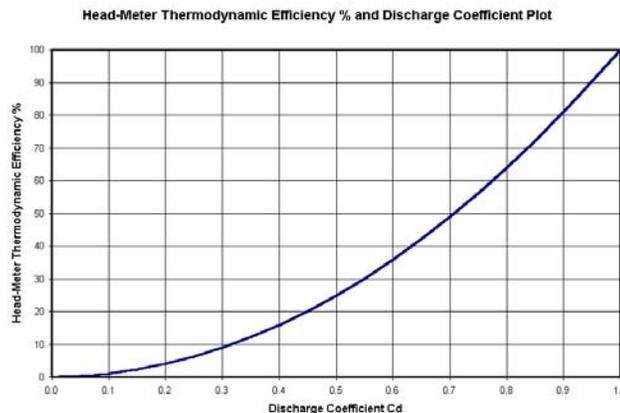
Figure 23 - Square-edged Orifice Plate C_D and Multi-holed Orifice Plate C_D



For thin plates $t/D < 0.4$ and for a wide range of beta factors, the Balanced Flow Meter discharge coefficient C_D is constant at ~ 0.89 . The high Reynolds number concentric orifice C_D is ~ 0.61 . The multi-holed thin plate C_D is ~ 0.74 at high Reynolds numbers. Thick BFM plates, with $t/D > 1$, have C_D 's with values close to ~ 0.99 , which is similar for the multi-holed orifice plate. The reason the C_D goes up with plate thickness is that the vena contracta velocity-head is recovered at t/D values greater than one (1). Frictional effects are minimal at thickness to diameter ratios, t/D 's, less than seven (7). Please see A+ FLOWTEK technical article PTK-803.

The relation of thermodynamic efficiency and discharge coefficient is provided below. For a discharge coefficient of 0.6, the thermodynamic efficiency is only 36%. This is a high percentage loss of available energy. The BFM efficiency is 80% for a thin plate, and nominally thick-plates provide 95+% thermodynamic energy recovery.

Figure 24 - Head-Meter Thermodynamic Efficiency η_{eff} and Discharge Coefficient C_D Graph



The following table gives discharge coefficients C_D for a range of beta factors. Values for fouled, and upstream and downstream elbow fittings are provided. These are representative values for thin plates. A comprehensive test with detailed results is



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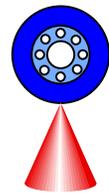


Table 14 - Basic Bernoulli Head-Meter Equation and Discharge Coefficient Based on Transport Phenomena

| Orifice and Venturi Meter Basic Bernoulli Head-Meter Equation ¹ | Discharge Coefficient Based on Transport Phenomena, BSL, 1960 ² |
|--|--|
| $w = \rho_2 S_2 \sqrt{\frac{-2\alpha_2 \int_1^2 (1/\rho) dp}{1 - \frac{\alpha_2}{\alpha_1} \left(\frac{\rho_2 S_2}{\rho_1 S_1}\right)^2 + \alpha_2 e_v}} \quad (15.4-3)$ | $C_D = \frac{\sqrt{\frac{2g_c J \Delta H_s}{\left(\frac{\alpha + k_f}{(\rho A)_s}\right)_b - \left(\frac{\alpha}{(\rho A)_s}\right)_a}}}{\frac{\gamma A_b}{\sqrt{1 - \beta^4}} \sqrt{2g_c \rho_a \Delta P}} \quad \gamma = \left(\frac{\left(\frac{\Delta H}{\Delta P}\right)_s (\rho_a (1 - \beta^4))}{\left(\left(\frac{\rho_a}{\rho_{b,s}}\right) - \beta^4\right)} \right)^{\frac{1}{2}}$ |
| | $C_D = \frac{1 - \beta^4 \left(\frac{\rho_b}{\rho_a}\right)^2}{\left(\alpha + k_f\right)_b - \alpha_a \beta^4 \left(\frac{\rho_b}{\rho_a}\right)^2} \approx Const.$ |
| | $C_D \equiv \frac{m_{act}}{m_{ideal}} = \frac{(\rho A v)_{act}}{(\rho A v)_{ideal}} = \frac{\rho_{b,f}}{\rho_{min,S}} \left(\frac{\alpha}{\alpha + k}\right)_b = \frac{\rho_{b,f}}{\rho_{min,S}} (\eta_{eff})^{1/2}$ |
| | $\eta_{eff} = \frac{v_{b,f}^2}{v_{b,ideal}^2} = \frac{(m/\rho)_{b,f}^2}{(m/\rho)_{b,ideal}^2} = \frac{H_b^* - H_{b,f}}{H_b^* - H_{min,S}} = \frac{\alpha v_{b,f}^2}{\alpha v_{b,f}^2 + k v_{b,f}^2} = \left(\frac{\alpha}{\alpha + k}\right)_b$ |
| 1) α defined by $Kinetic\ Energy = \frac{v^2}{\alpha 2}$ | 2) Contemporary basis $Kinetic\ Energy = \frac{\alpha v^2}{2}$ |



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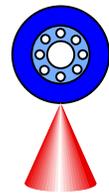


Table 18 - Head Meters with Contraction and Expansion

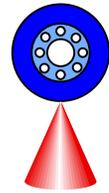
| Head Meter Permanent Pressure Loss with Contraction & Expansion | |
|---|---|
| <p style="text-align: center;">Balanced Flow Meter Pressure Recovery</p> | <p style="text-align: center;">Orifice Meter Pressure Recovery</p> |
| $k_{f,PH} = \left(\frac{1}{(\gamma C_D)^2} - 1 \right) (1 - \beta^4)_{to \text{ plate hole}} \quad \text{or} \quad k_{f,PT} = \left(\frac{(1 - \beta^2)(1 - \beta^4)}{(\gamma C_D)^2} \right)_{\text{pipe taps}} \quad \text{for } v_{\beta}$ $k_{f,PH} = \left(\frac{1}{(\gamma C_D)^2} - 1 \right) \left(\frac{1 - \beta^4}{\beta^4} \right)_{to \text{ plate hole}} \quad \text{or} \quad k_{f,PT} = \left(\frac{(1 - \beta^2)(1 - \beta^4)}{\beta^4 (\gamma C_D)^2} \right)_{\text{pipe taps}} \quad \text{for } v_{Pipe}$ $C_{\text{contraction, vena-contracta}} = C_{D, \text{orifice}} = \pi / (\pi + 2), \quad C_{D, BFM} = (C_{D, \text{orifice}})^{1/4}, \quad C_{D, \text{venturi}} = (C_{D, \text{orifice}})^{1/8}$ | |
| <p style="text-align: center; font-size: small;">Velocity Head-Loss Coefficients</p> <p style="text-align: center;">Pipe Taps - Beta Area Velocity Basis</p> | <p style="text-align: center; font-size: small;">Velocity Head-Loss Coefficients</p> <p style="text-align: center;">Pipe Taps - Pipe Area Velocity Basis</p> |

k_f is determined from measured values based on air testing, ranging from minimum flows to sonic velocities. Values are calculated by an ideal gas compressible fluid equation, verified from the NASA Gas Frictional-Flow Tables. The equation is shown on the next page. Testing results are given in the table below and shown graphically with the figure below.



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$$\bar{G}^2 = \left(\frac{m}{A_{Pipe}} \right)^2 = \frac{2g_c(1 - (\rho_b/\rho_a)^2)}{(1 - 1/\lambda)((\rho_b/\rho_a)^2 - 1)/2 - 2(1 + 1/\lambda)\ln(\rho_b/\rho_a) + k_f}$$

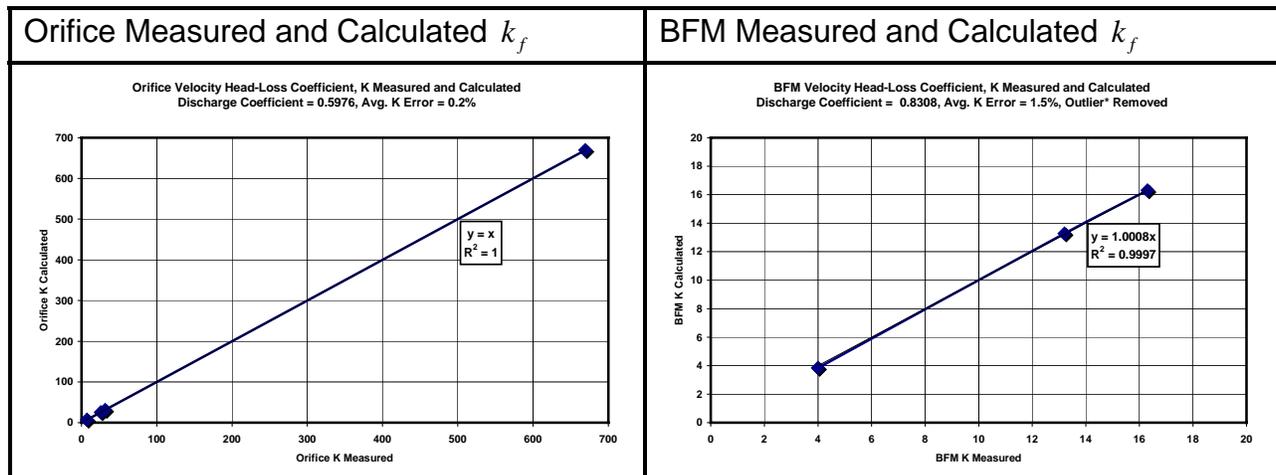
Equation 84

Table 19 - Measured and Calculated Velocity Head-Loss k_f for Pipe Taps

| Bernoulli Meter Beta, $\beta = D_a / D_b$ | 0.250 | 0.500 | 0.521 | 0.650 |
|---|--------|-------|-------|-------|
| Venturi k_f Calculated, $C_D = 0.940$ | 270.6 | 12.7 | 10.4 | 3.0 |
| BFM k_f Measured | 287.1* | 16.3 | 13.2 | 4.0 |
| BFM k_f Calculated, $C_D = 0.831$ | 346.4 | 16.3 | 13.3 | 3.9 |
| Orifice k_f Measured | 669.4 | 31.5 | 25.7 | 7.4 |
| Orifice k_f Calculated, $C_D = 0.598$ | 669.4 | 31.5 | 25.6 | 7.4 |
| Orifice to BFM k_f Ratio Reduction % | 233 | 193 | 195 | 185 |

With the table above, the BFM coefficient of discharge C_D value is 0.831 for a pipe tap configuration. The flange tap C_D value is 0.89+. All k_f values for the BFM and orifice are measured. Values shown for the venturi flow meter are calculated. Density corrections and k_f values for the venturi meter are estimated for a downstream pipe tap location. All k_f values are based on pipe tap permanent pressure losses and the pipe flow area as the velocity reference. Comparisons of the Balanced Flow Meter with the orifice plate show a k_f reduction of ~200%. k_f accuracies are shown below.

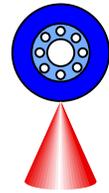
Figure 26 - Measured and Calculated Velocity Head-Loss k_f , Orifice and BFM Plate





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The problem in mathematical form is to minimize the sum of the normalized squared errors,

$$\phi(X_i) = \sum \frac{(X_i - X_i^*)^2}{\sigma_i^2}. \quad \text{Equation 89}$$

Subject to the constraints imposed by j mass (or model based) balance equations of the form,

$$\Psi_j(X_i) = \sum_i a_{ij} X_i = 0. \quad \text{Equation 90}$$

Where X_i is the i th corrected measurement, X_i^* is the i th observed measurement, σ_i^2 is the error variance of the i th measurement, Ψ_j is the balance constraint for the j th balance equation, and a_{ij} is the coefficient of X_i in the j th balance equation. The value of a_{ij} is +1 for an input stream, and -1 for an output stream.

The method of LaGrange is used to obtain the X_i values. This involves solving a set of simultaneous equations,

$$\frac{\partial}{\partial X_i} [\phi(X_i) + \sum_j \lambda_j \psi_j(X_i)] = 0, \quad \text{Equation 91}$$

$$\frac{\partial}{\partial \lambda_j} [\phi(X_i) + \sum_j \lambda_j \psi_j(X_i)] = \psi_j(X_i) = 0, \quad \text{Equation 92}$$

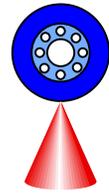
λ is a LaGrangian multiplier.

Any technique for solving the above equations will yield a set of adjusted flow rates (X_i) that will satisfy the mass-balance equations for all the nodes, i.e. systems. To apply LaGrange's method, it is necessary to know the error variance for each measurement, σ_i^2 . The value of σ_i^2 for each measuring instrument is estimated independently, typically from the manufacturer's specification, together with the engineer's knowledge of how the instrument actually performs.



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Gross-Error Detection

While LaGrange's method will always yield a good mass balance, the adjusted flow rates will be erroneous if there are gross errors in the data. Therefore, such errors have to be identified and bad measurements removed before proceeding with data-reconciliation.

Most methods for gross-error detection involve the use of statistical tests on the basis that random errors in the data are normally distributed. Considered are two general types of statistical tests:

1. Methods for analyzing least-squares residuals. The approach is to adjust the data using a least-squared-error analysis and then calculate a set of residuals,

$$l_i = X_i - X_i^* . \quad \text{Equation 93}$$

The outliers among the l_i values are those that exceed some number of standard deviations, $\sigma = 1.96$ for a 95% confidence level. Thus, a given residual is an outlier if the following is true:

$$\frac{l_i}{\sigma_{l,i}} > 1.96, \text{ or } \frac{l_i}{\sigma_{l,i}} < -1.96 . \quad \text{Equation 94}$$

where $\sigma_{l,i}$ is the standard deviation of the residual (the residuals are random variables, like the X_i values). The outliers are the results of gross errors.

2. Methods for analyzing nodal imbalances. The nodal imbalance is the left-hand side of the mass balance equation. This value is divided by the nodal standard deviation, σ_j which is the square root of the nodal variance:

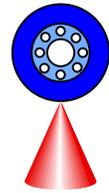
$$\sigma_j^2 = \sum_i \sigma_i^2 \quad \text{Equation 95}$$

Here, the summation extends over all the streams connected to node j . The resulting normalized statistic is denoted δ_j ,

$$\delta_j = \frac{\sum_i a_{i,j} X_i}{\sigma_j} . \quad \text{Equation 96}$$



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For a 95% confidence level, there are one or more gross errors in the measurements associated with node j when,

$$\delta_j > 1.96, \text{ or } \delta_j < -1.96.$$

Equation 97

Modified Iterative Measurement Test, MIMT©

Error detection methods have two principal drawbacks,

1. The least-squares procedure tends to spread the error over all the data so even the best measurements can have high residuals. When these residuals fail the test for outliers, the corresponding measurements are erroneously identified as having gross errors.
2. There is no provision to prevent unrealistic flow rates from being computed. If the algorithm fails to identify all the gross errors, the data-reconciliation procedure may generate negative flow rates, or absurdly large positive ones, for some of the streams.

The modified iterative measurement test, MIMT©, is designed to overcome these two problems. The test for outliers is applied in a stepwise fashion. After the least-squares calculation, the one measurement corresponding to the worst significant outlier is identified as having a gross error. This measurement is removed from the data set by nodal aggregation, and the least-squares analysis is repeated on the reduced data set to obtain residuals for the next round of tests. If unrealistic flow rates result when a particular measurement is removed, that measurement is put back in the calculation and the next-worst outlier is removed instead. When there are no more outliers among the remaining residuals, the iterations are terminated.

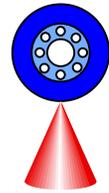
Example of MIMT© Gross-Error Detection

The following table provides a set of measurements that contain both random and gross errors. These data were developed by superimposing random and gross errors onto a perfect mass balance. The stream numbers are shown in column 1. The measured values with gross errors are shown in column 2. The third column gives the reconciled flow rates obtained using the modified iterative measurement test. The algorithm correctly identified all three gross errors, and correctly reconciled the flow rates, including the ones that had been grossly wrong. The flow rate for Stream 2, for instance, was adjusted downward by 40%. The sum of the absolute values of the errors is only 30.50 in the reconciled data, vs. 252.58 in the raw data; the MIMT© method reduced the error by 88%.



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MIMT© Performance

The performance of five well-known and industrially applied error detection algorithms with a steam-metering system has been tested. The data were created by computer, using a random-number generator to superimpose errors on flow rate data that were balanced to begin with. One hundred such cases were created. The percentage of measurements containing gross errors was permitted to be as high as 25%; the magnitude of the gross errors ranged from about 10% to 100% of the true flow rate.

The modified iterative measurement test MIMT© method identified over 80% of the gross errors, made few erroneous identifications, and reduced the total error by more than 60%. No other method tested provided this consistency, accuracy increase or identified the gross errors dependably. The MIMT© method is reliable and effective for detecting gross errors in data sets subject to linear constraints. Additionally, the modified iterative measurement test MIMT© has the advantage of being much faster than other methods. The average computing times per run were 3 seconds for the MIMT©, and more than 40 seconds for the other methods.

A number of chemical process companies are using MIMT© as an aid to process analysis, control, leak detection and inventory accounting. The modified iterative measurement test MIMT© method is effective at detecting gross errors and reconciling data. It is sufficiently robust to be used in an industrial and “real-time” environment. The methods have been extended to non-linear systems, such as those involving component mass balances, energy balances, physical property verification and sensor measurements with model based constraints.

Advantages of MIMT©

As has been seen throughout the last several decades, all chemical plants have bad flow meters. Often the percentage of bad meters is as high as 25%. Without Error Detection & Data Reconciliation, plant personnel live with these problems because they have no way of knowing which meters are bad. These bad meters result in inventory problems or accountability problems, such as not knowing where the fluids are going or why the inventory does not balance. When problems occur in the plant, since the meters are bad, it is difficult to tell what is happening. Therefore, the resolution of the real problem takes additional time and loss of profit. Often metering problems can cost several thousand dollars per day. Once the QMC MIMT© Program is applied, the problems can be corrected in one day.

The QMC MIMT© Program also enables one to find and pinpoint leaks and losses. It identifies the bad meters/sensors, and calculates what the bad meters/sensors should be reading, termed data reconciliation. Gross errors are pinpointed and corrected. Minor errors are corrected statistically based on weighted flow meter error analysis and mass balance constraints.



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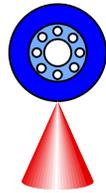


Table 23 - Fluid Flow Parameters

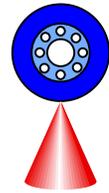
| Equation Basis | Equation |
|---|---|
| Velocity Distribution | $\frac{v_r}{v_{r=0}} = 1 - \left(\frac{r}{R}\right)^2$ |
| | $\frac{\kappa(v_r - v_{r=0})}{v_{avg} \sqrt{f/2}} = Ln\left(1 - \frac{r}{R}\right)$ |
| | $\frac{v_r}{v_{r=0}} \cong \left(1 - \frac{r}{R}\right)^{\frac{1}{n}}$ |
| Average Velocity | $\frac{v_{avg}}{v_{r=0}} = \frac{1}{2}$ |
| | $\frac{v_{avg}}{v_{r=0}} = \frac{1}{1 + 10\kappa\sqrt{f/2}} = \frac{2n^2}{(n+1)(2n+1)}$ |
| Momentum Correction Factor | $\beta = \frac{4}{3}$ |
| | $\beta = \frac{(1+n)^2(2+n)^2}{2(1+2n)(2+n)}$ |
| Kinetic Energy Correction Factor | $\alpha = \frac{(1+n)^3(2+n)^3}{4(1+3n)(2+3n)}$ |

Table 24 - Friction Factors with Relative Roughness Factor

| | |
|---|---|
| Turbulent Flow, Fully Rough $(D/e)/(Re\sqrt{f}) \leq 0.01$ | $\frac{1}{\sqrt{2f}} = Log\left(\frac{D}{e}\right) + B$ |
| Turbulent Smooth/Rough Transition | $\frac{1}{\sqrt{2f}} = Log\left(\frac{D}{e}\right) - Log\left(\frac{(\varepsilon - 1)(D/e)}{Re\sqrt{f}} + 1\right) + B$ |
| $f \pm 1.5\%, 10^8 \geq Re \geq 10^4, 0.05 \geq e/D \geq 0 \Rightarrow$ | $\frac{1}{\sqrt{f}} = -3.6 Log\left[\frac{6.9}{Re} + \left(\frac{e/D}{3.7}\right)^{\frac{10}{9}}\right]$ |
| $Re \equiv \frac{D\bar{G}}{\mu} \cong const., \bar{G}_{@any \bar{x}} \equiv m/A = const., \kappa \cong 0.407, \varepsilon \cong 5.67, B = 2.28, Log = Log_{10}$ | |



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PTK-810 Balanced Flow Meter Temperature Expansion ©

Temperature Expansion

When the temperature of a body is increased, the average distance between atoms increases. This leads to an expansion of the whole solid body as the temperature is increased. The change in any linear dimension of the solid, such as its length, width, or thickness, is called a linear expansion. If the length of the linear dimension is l , the change in length from a change in temperature ΔT is Δl . Based on experimental data, with small temperature changes, ΔT , the change in length Δl is proportional to the temperature change ΔT and to the original length l ,

$$\Delta l = \alpha l \Delta T, \quad \text{Equation 98}$$

where α , called the coefficient of linear expansion, has different values for different materials. Rewriting this formula gives,

$$\alpha = \frac{1}{l} \frac{\Delta l}{\Delta T}, \quad \text{Equation 99}$$

and α is the fractional change in length per degree temperature change.

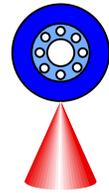
On the microscopic level, thermal expansion is an increase in the average separation between the atoms in the solid. The potential energy curve for two adjacent atoms in a crystalline solid as a function of their internuclear separation is an asymmetric curve. As the atoms move, close together, the separation decreases from the equilibrium radius. Strong repulsive forces come into play and the potential curve rises steeply ($F = -dU/dr$). As the atoms move farther apart, the separation distance increases from the equilibrium value and somewhat weaker attractive forces take over. The potential curve rises more slowly in this region.

If the potential energy curve were symmetric about the equilibrium separation, then no matter how large the amplitude of the vibration becomes the average separation would correspond to the equilibrium separation. Thermal expansion is a direct consequence of the deviation from symmetry (that is, the asymmetry) of the potential energy curve characteristic of solids.

Some crystalline solids in certain temperature regions may contract as the temperature rises. The above analysis remains valid where compression (i.e., longitudinal) modes of vibration exist and dominate. Solids may vibrate in shear-like (i.e., transverse) modes as well and these modes of vibration will allow the solid to contract as the temperature rises, the average separation of the planes of atoms decrease. For certain types of crystalline structure and in certain temperature regions, these transverse modes of



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the discharge coefficient and the gas expansion factor. The pipe diameter at flowing conditions is calculated by

$$D = F_{\alpha D} D_{meas} = [1 + \alpha_p (T_F - 68)] D_{meas} \quad \text{Equation 100}$$

D_{meas} is the measured pipe diameter in inches at a reference temperature of 68°F (15.5°C). $F_{\alpha D}$ is the pipe material's thermal expansion correction and α_p is the coefficient of thermal expansion for the pipe material in in/(in °F).

Similarly, the bore diameter(s) is corrected for the flowing temperature by

$$d = F_{\alpha d} d_{meas} = [1 + \alpha_{PE} (T_F - 68)] d_{meas} \quad \text{Equation 101}$$

where d_{meas} is the measured bore diameter in inches at a reference temperature of 68°F (15.5°C), $F_{\alpha d}$ is the thermal expansion correction factor and α_{PE} is the coefficient of thermal expansion for the primary element material in in/(in °F).

The orifice is manufactured and measured at approximately 68°F; this measured diameter is determined by combining the equations as

$$d_{meas} = \frac{d}{F_{\alpha d}} = \frac{\beta D}{F_{\alpha d}} \quad \text{Equation 102}$$

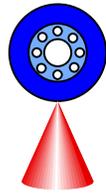
where by definition the equivalent bore diameter at the flowing temperature is $d = \beta D$. For the Balanced Flow Meter, the equivalent bore diameter is reported in the A+ FLOWTEK Sizing Program and the beta factor β is used for setting the hole-layout.

The following tables give the linear expansion coefficients values and equations for many pipe and plate materials.



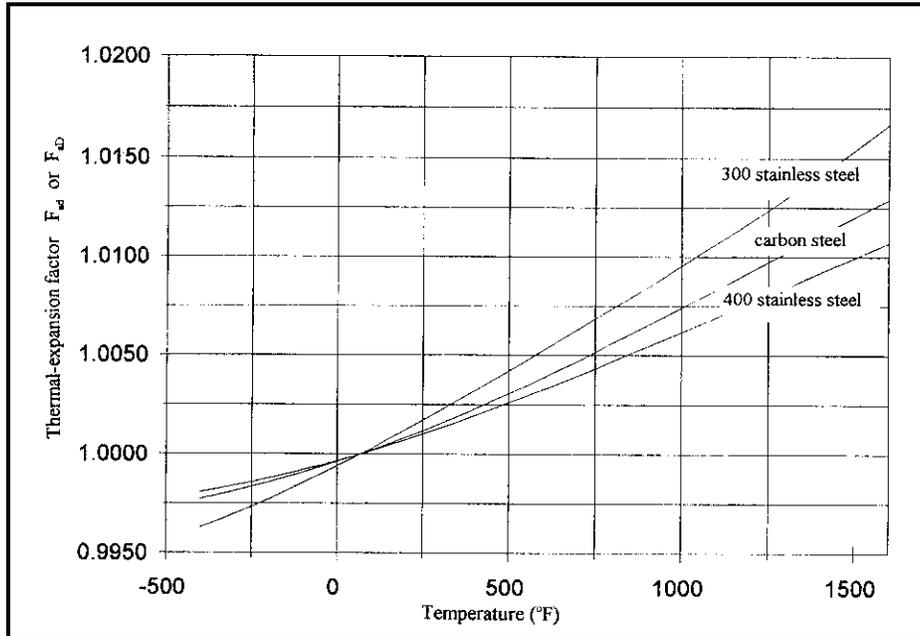
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The following figure graphically provides the F_{ad} and $F_{\alpha D}$ corrections. The values from these graphs are used for thermal expansion dimensional corrections for the Balanced Flow Meter plate and pipe.

Figure 29 - Thermal Expansion Plots of Common Steels



The above values, equations and figures are used to correct the flowing temperature β from the pipe and plate manufactured temperature and vice versa.

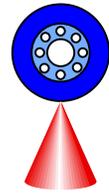
For plates and pipes made of the same material no beta-ratio (β) thermal expansion correction is required. For the typical plate design of 304 or 316 stainless steel with a carbon steel pipe, the ratio of $F_{ad} / F_{\alpha D}$ is typically close to one (1) in the temperature region of most industrial applications. The A+ FLOWTEK Balanced Flow Meter Sizing Program calculates the measured hole diameters based the 300 stainless steel series basis.

For highly accurate applications, $< 0.25\%$, the thermal correction ratio should be applied from real time temperature measurements. A fluid density correction should also be applied. The flowing fluid density correction is typically most important when compared to the pipe/plate thermal expansion ratio correction.



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PTK-811 Balanced Flow Meter Low Reynolds Number Corrections ©

Foreword

For fluid-flow processes, equation adjustments are required for the laminar, transition, and turbulent flow regimes. The orifice or venturi equation, as derived from the Bernoulli equation, is typically based on fully developed turbulent flow. However, for Reynolds numbers below 30,000 to 60,000 correction factors are required to adequately describe the Bernoulli head meter. These effects are reviewed and equations are provided to correct for low Reynolds number operations with the Balanced Flow Meter.

Engineering Bernoulli Equation

The steady state *engineering Bernoulli equation* for flow systems of any steady state, thermodynamic path and flow regime is given by,

$$\Delta \left(\frac{\alpha v^2}{2g_c} \right) + \Delta \frac{gZ}{g_c} + \int \frac{dP}{\rho} + W_{shaft} + LW_{friction} = 0 \quad \text{Equation 103}$$

For a head meter, such as an orifice, venturi, or Balanced Flow Meter, the generalized orifice equation is used as derived from the Bernoulli equation,

$$m = \frac{C_D \gamma A_b}{\sqrt{1 - \beta^4}} \sqrt{2g_c \rho_a \Delta P} \equiv \text{Orifice Equation} \quad \text{Equation 104}$$

The discharge coefficient C_D is used to correct for losses from frictional effects and to correct for low Reynolds numbers, where the kinetic energy correction factors α, s are not one (1). A more general equation is,

$$m = \frac{C_v C_c \gamma A_b}{\sqrt{\alpha_b - \alpha_a C_c^2 \beta^4}} \sqrt{2g_c \rho_a \Delta P} \quad \text{Equation 105}$$

For an orifice plate, C_v is the velocity coefficient and C_c is the contraction coefficient. The kinetic energy correction factors are for the upstream position, α_a , and downstream position, α_b .

Orifice Contraction and Velocity Coefficients

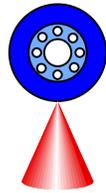
In the fully developed turbulent flow regime, the contraction coefficient C_c is given by,

$$C_c = \left(1 + \frac{(1 - \beta^4)}{(1 + C_D)} \right)^{-1} \quad \text{Equation 106}$$



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The beta factor β is the orifice hole to pipe diameter ratio, $D_{Orifice} / D_{Pipe}$, and the discharge coefficient C_D for any beta factor is,

$$C_D = \frac{\pi}{\pi + 2} \quad \text{Equation 107}$$

The orifice velocity coefficient C_V is given by,

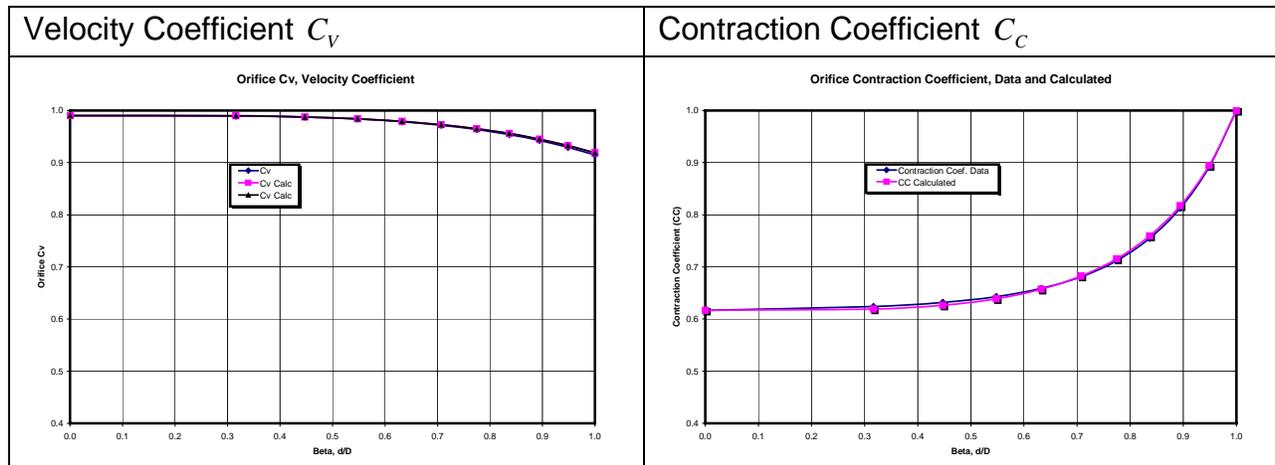
$$C_V = \left(\frac{C_D}{C_C} \right)_{\beta=0} \left(1 - C_D^4 \beta^4 \right)^{\frac{1}{2}} \quad \text{Equation 108}$$

On this basis,

$$C_D = \left(\frac{C_D}{C_C} \right)_{\beta=0} \left(1 - C_D^4 \beta^4 \right)^{\frac{1}{2}} \left(\frac{1 - \beta^4}{(1/C_C^2 - \beta^4)} \right)^{\frac{1}{2}} \quad \text{Equation 109}$$

Plots of these factors for measured and calculated orifice plate values are shown below,

Table 27 - Orifice Plate Flow Correction Factors at High Reynolds Numbers



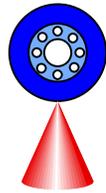
Kinetic Energy Correction Factor

The kinetic energy correction factor corrects the average pipe velocity for the actual velocity profile. The velocity profile in developed flow is dependent on the friction factor f . The kinetic energy correction factor α is given by,

$$\alpha = 1 + \frac{f}{8\kappa^2} \left(15 - \frac{9}{\kappa} \sqrt{f} \right), \quad \text{Equation 110}$$



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for turbulent flow, and $\alpha = 2$ for laminar flow.

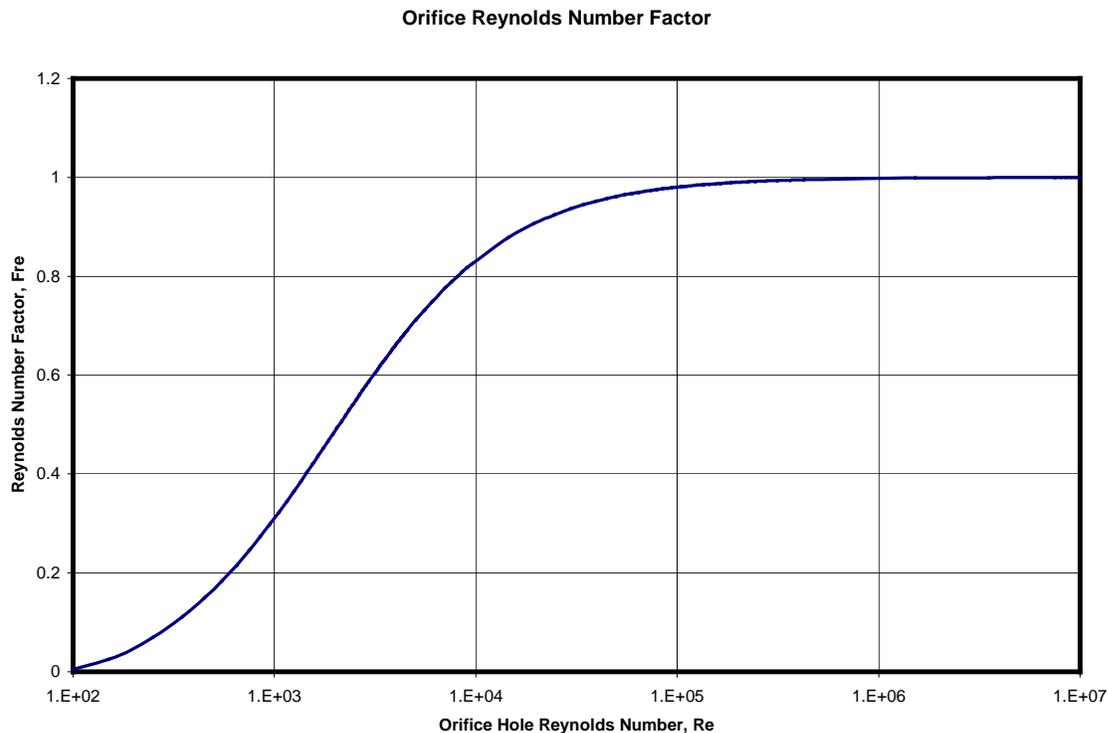
Reynolds Number Correction, Contraction Factor and Contraction Coefficient

The contraction coefficient is a function of Reynolds number. The Reynolds number correction factor for an orifice is given by,

$$F_{Re} = 1/(1 + 2000/(Re - 100)), \quad \text{Equation 111}$$

as shown in the following figure.

Figure 30 - Orifice Contraction Correction Factor



The orifice contraction factor exponent C_f corrects the contraction factor as a function of beta factor. The regressed equation is,

$$C_f = 29.442\beta^4 - 35.747\beta^3 + 15.731\beta^2 - 2.5684\beta + 0.5323 \quad \text{Equation 112}$$

The following figure provides a plot of the contraction factor exponent C_f as a function of beta factor, and regression results. As shown, the R squared value is one (1).



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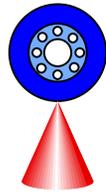
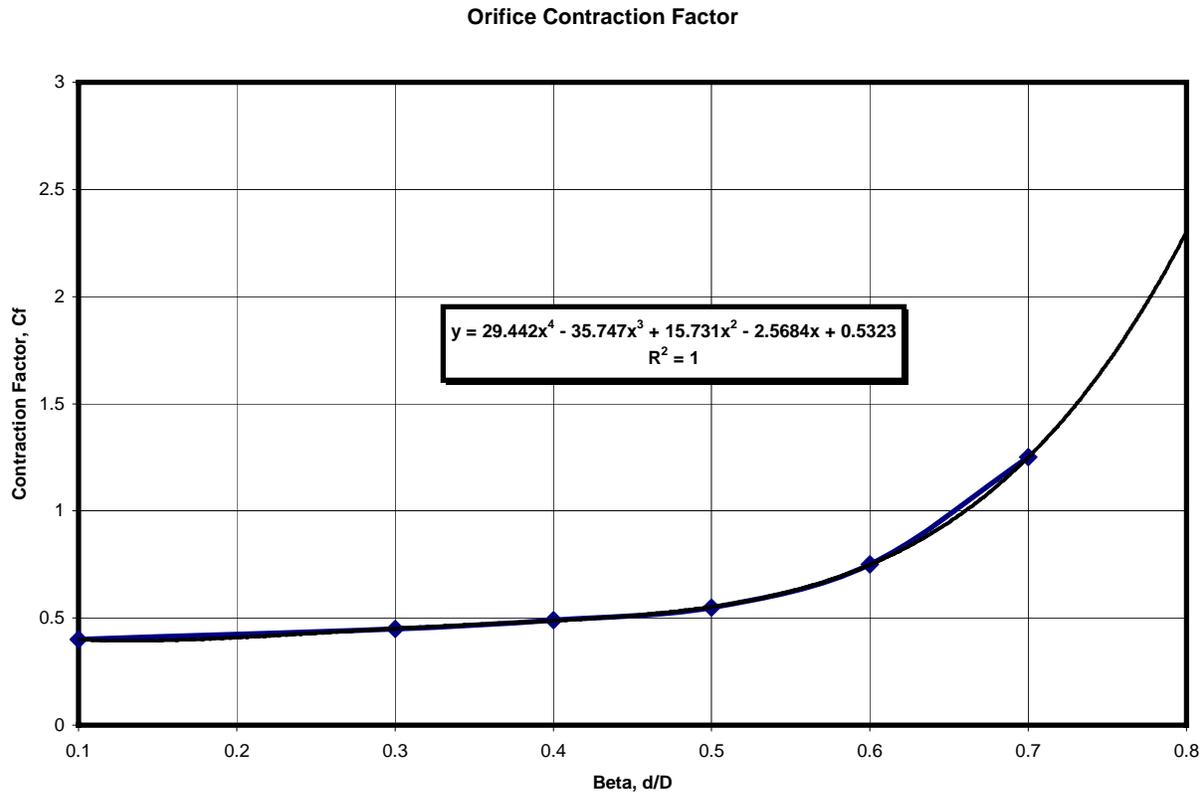


Figure 31 - Contraction Factor Exponent



With these factors, the orifice contraction coefficient, as a function of Reynolds number is given by,

$$C_c = \left(1 + \frac{(1 - \beta^4)}{F_{Re}^{C_f} (1 + C_D)} \right)^{-1} \quad \text{Equation 113}$$

Discharge Coefficient Calculation Method

The general calculation method for the discharge coefficient C_D is given by,

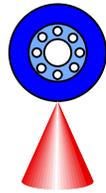
$$C_D = \frac{C_v C_c (1 - \beta^4)^{\frac{1}{2}}}{(\alpha_b - \alpha_a C_c^2 \beta^4)^{\frac{1}{2}}} \quad \text{Equation 114}$$

and $C_D / (1 - \beta^4)^{1/2}$ includes the velocity of approach. This method applies to any head meter, including the orifice, venturi, or Balanced Flow Meter. With $\alpha_a = \alpha_b = 1$ and with the contraction coefficient equal to a value of one, $C_c = 1$, the discharge coefficient C_D is



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equal to the velocity coefficient C_V , e.g. $C_D = C_V$. For other conditions the following method applies:

The velocity coefficient C_V is determined from,

$$C_V = \left(\frac{C_D}{C_C} \right)_{\beta=0} \left(1 - C_D^{4/m} \beta^{4/m} \right)^{\frac{m}{2}}, \quad \text{Equation 115}$$

and the contraction coefficient is given by,

$$C_C = \left(1 + \frac{(1 - \beta^4)}{F_{Re}^{C_f} (1 + C_D)} \right)^{-m}. \quad \text{Equation 116}$$

For an orifice ($m = 1$),

$$C_{D,O} = \left(\frac{\pi}{\pi + 2} \right)_{m=1}^m, \quad \text{Equation 117}$$

for the venturi ($m = 1/8 \dots 1/n'$),

$$C_{D,V} = \left(\frac{\pi}{\pi + 2} \right)_{m=1/8(\text{or } 1/n')}^m, \quad \text{Equation 118}$$

and for the Balanced Flow Meter ($m = 1/4$),

$$C_{D,BFM} = \left(\frac{\pi}{\pi + 2} \right)_{m=1/4}^m. \quad \text{Equation 119}$$

Discharge Coefficient Plots

With these Reynolds number correction factors, the discharge coefficients can be estimated. The following figures show the discharge coefficients for an orifice plate, venturi and Balanced Flow Meter as a function of Reynolds number. These figures may be used to estimate the discharge coefficients for Reynolds numbers less than 30,000 to 60,000. Surface tension may affect the discharge coefficient below a Reynolds number of about 300. For fluid flow systems where surface tension is not important, the discharge coefficients as shown on the following figures may be extrapolated (or calculated) for Reynolds number values below 300. A hole-tap is recommended for these fluid measurement systems. The hole-tap approach directly measures the Bernoulli effect. For any flow condition and design, field calibration is recommended.



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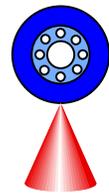


Figure 32 - Orifice Discharge Coefficients, Velocity of Approach Included

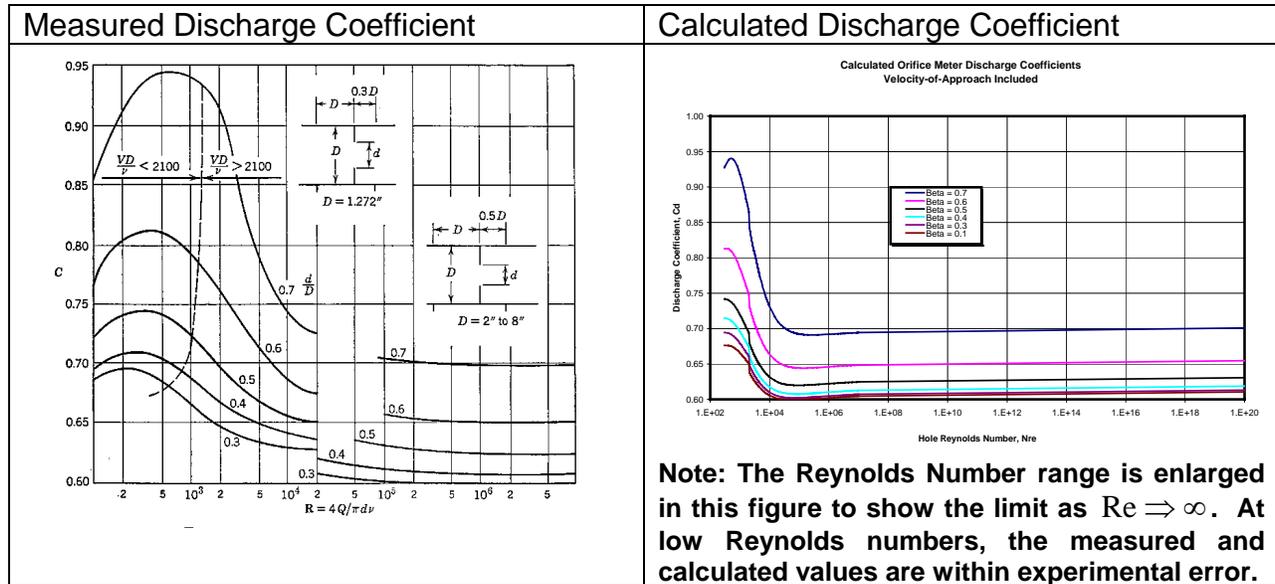
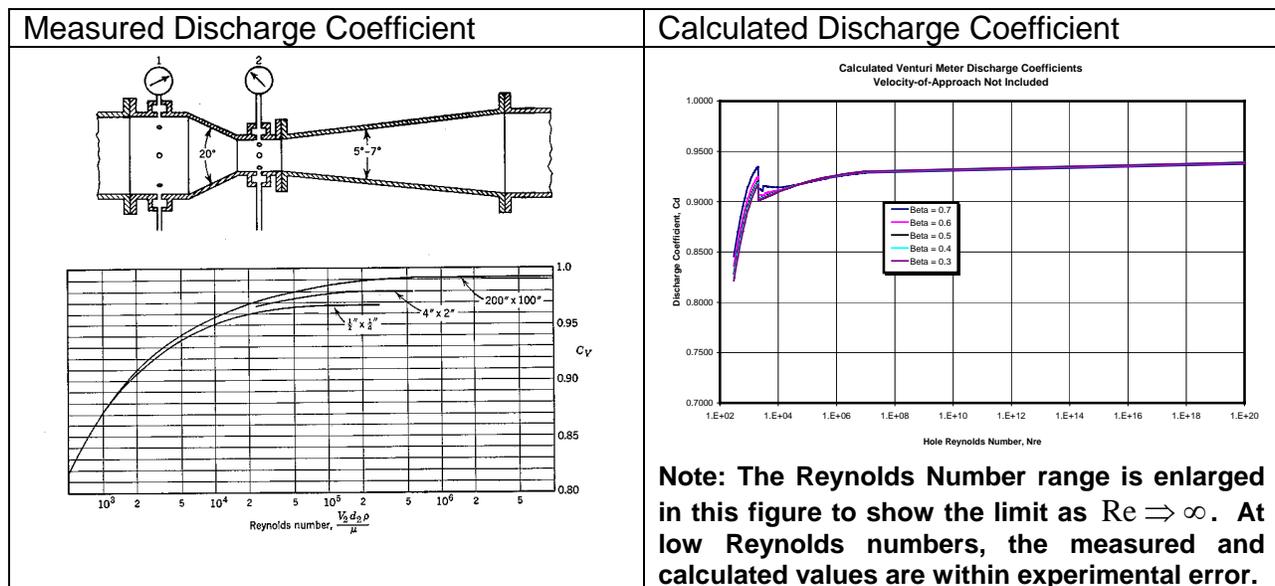


Figure 33 - Venturi Discharge Coefficients, Velocity of Approach Not Included





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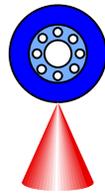
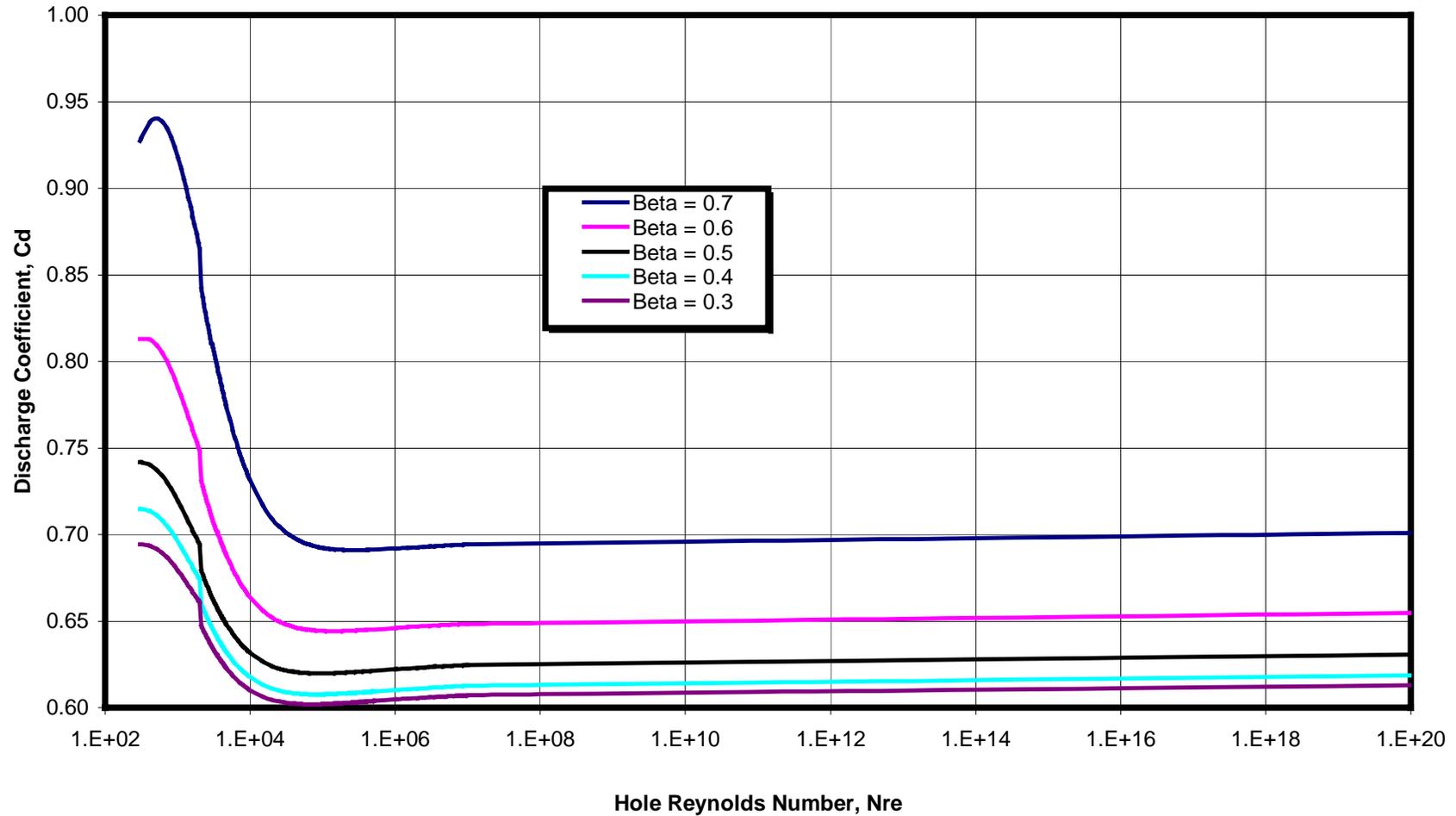


Figure 34 - Calculated Orifice Meter Discharge Coefficients, Velocity-of-Approach Included





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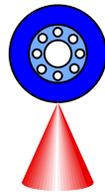
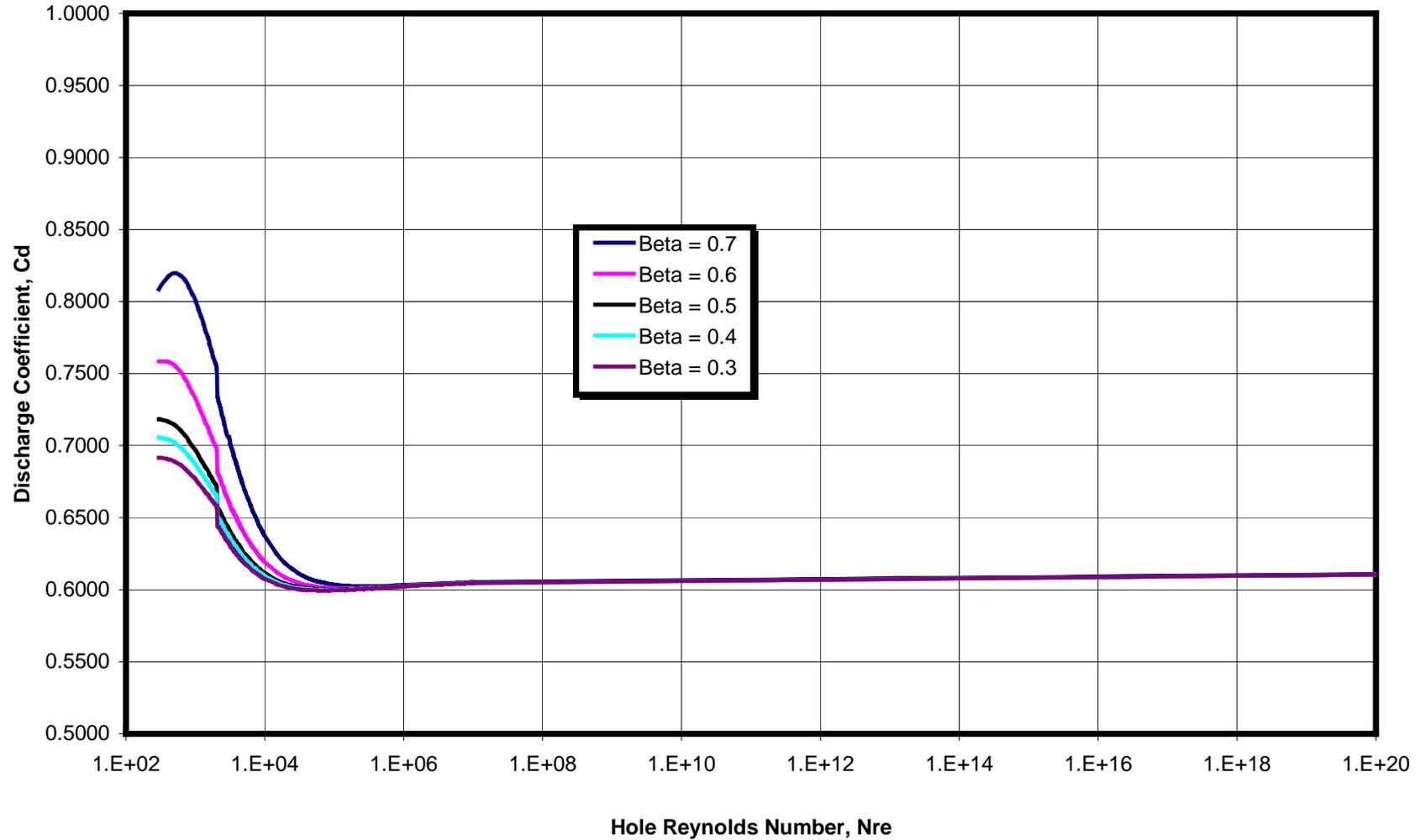


Figure 35 - Calculated Orifice Meter Discharge Coefficients, Velocity-of-Approach Not Included





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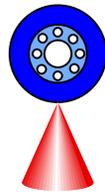
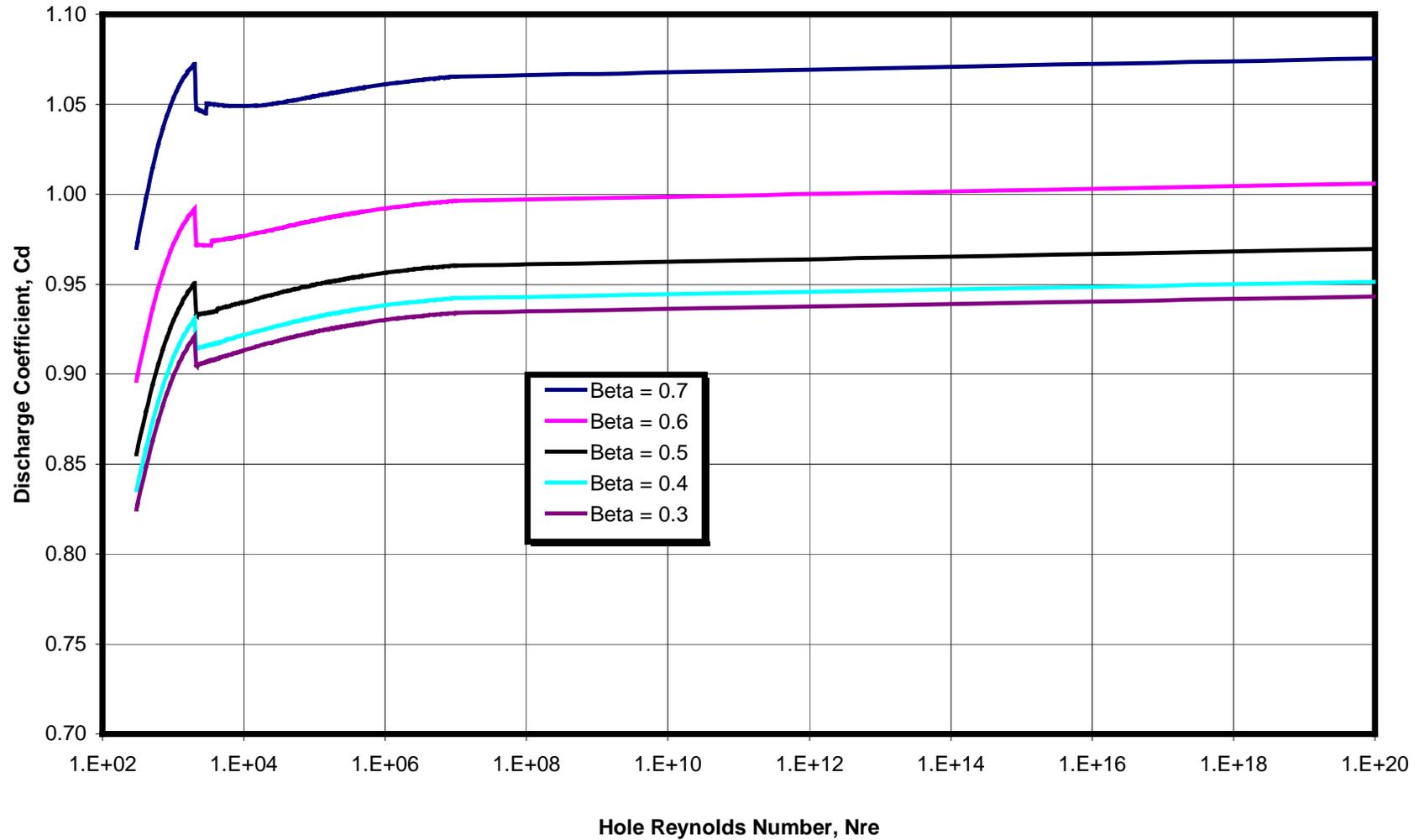


Figure 36 - Calculated Venturi Meter Discharge Coefficients, Velocity-of-Approach Included





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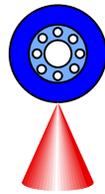
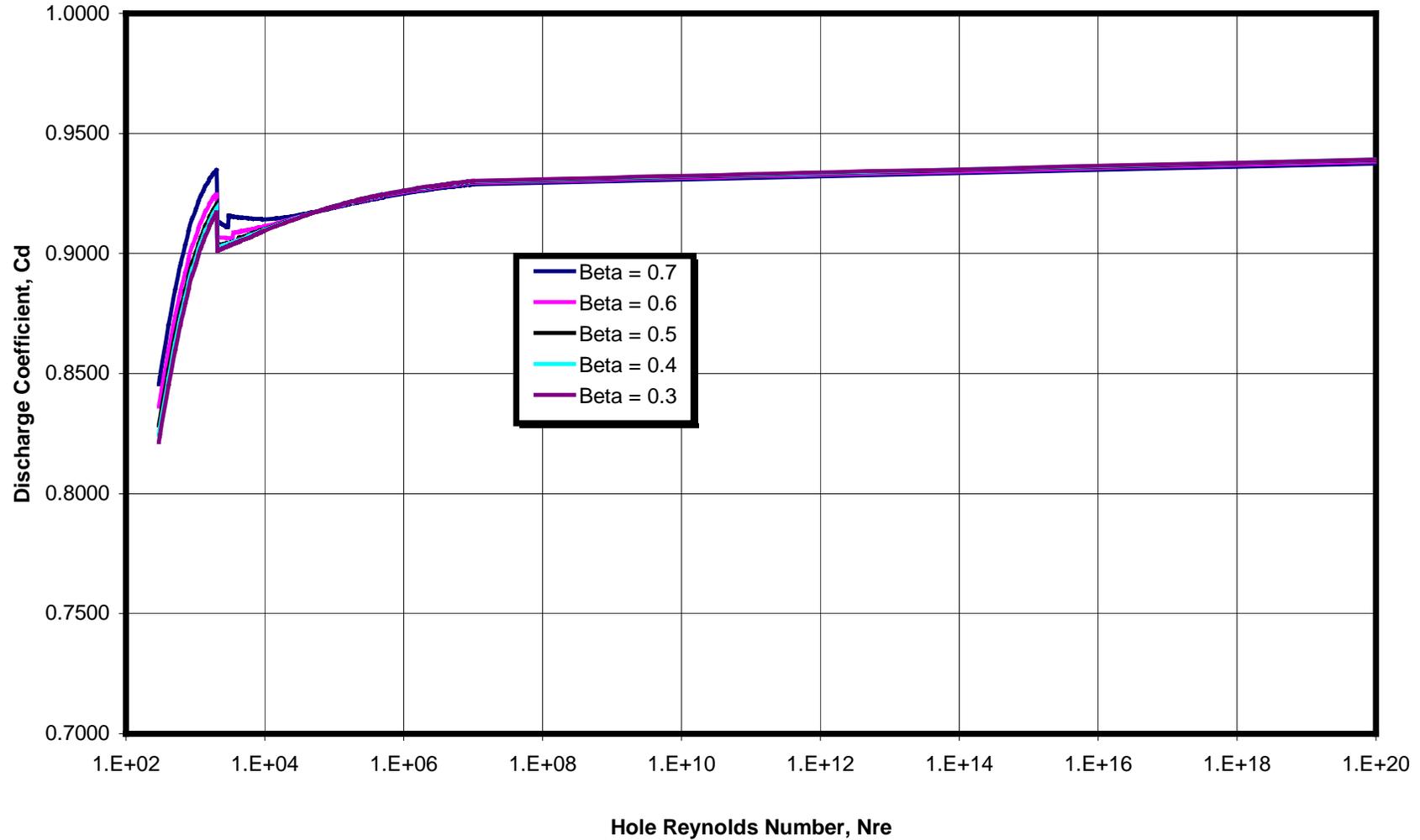


Figure 37 - Calculated Venturi Meter Discharge Coefficients, Velocity-of-Approach Not Included





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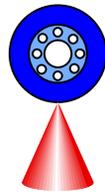
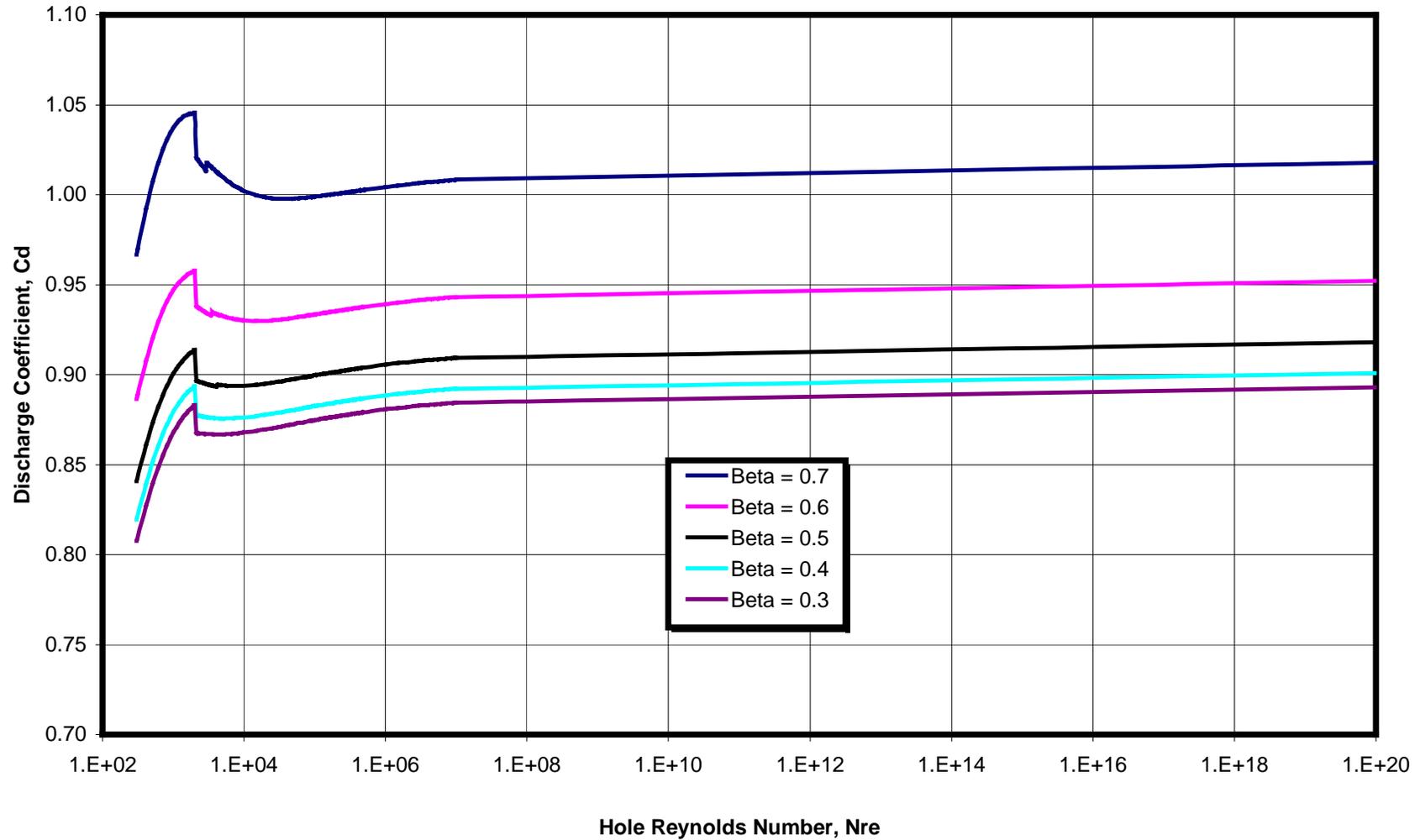


Figure 38 - Calculated Balanced Flow Meter Discharge Coefficients, Velocity-of-Approach Included





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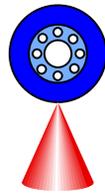
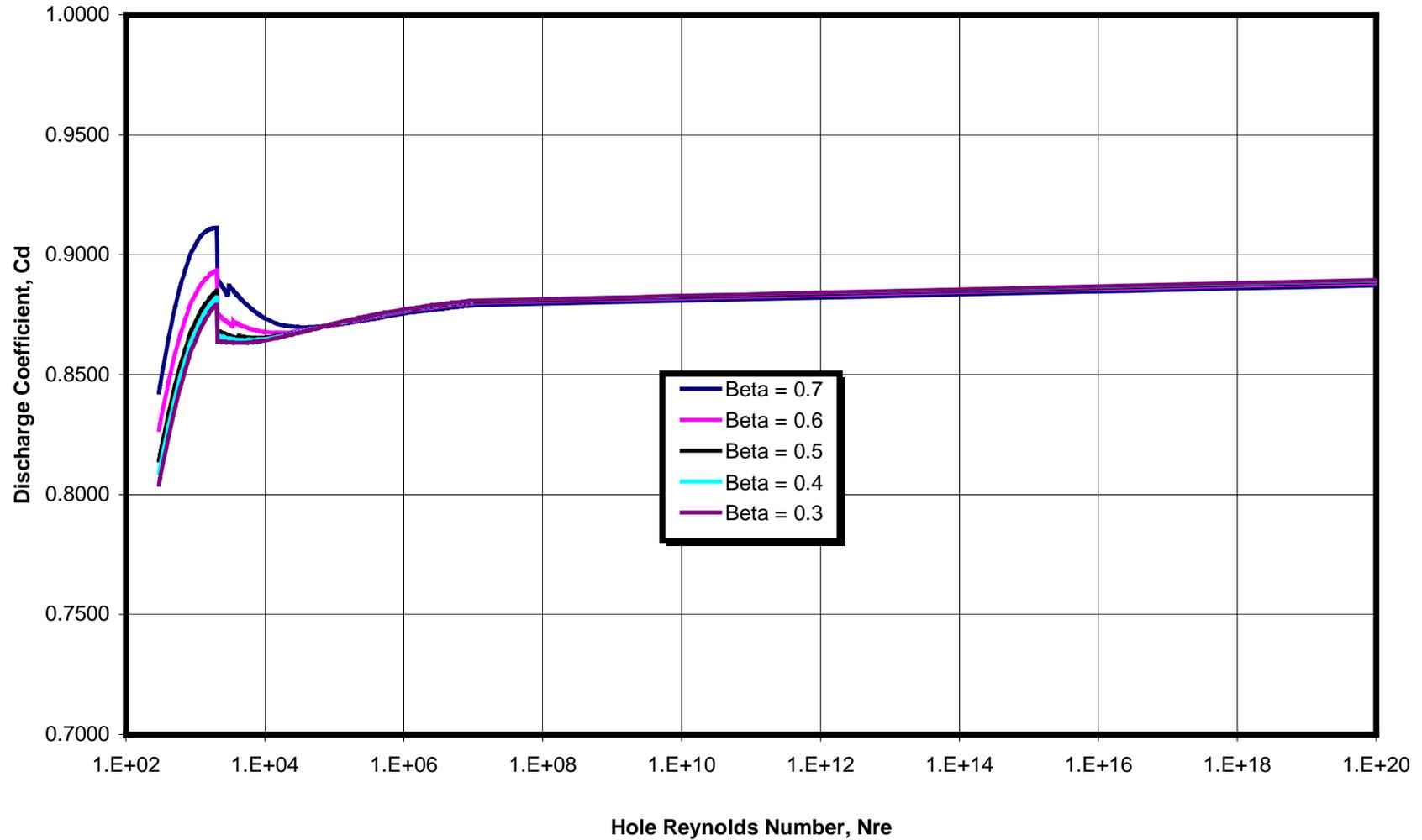
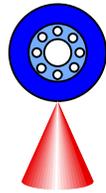


Figure 39 - Calculated Balanced Flow Meter Discharge Coefficients, Velocity-of-Approach Not Included





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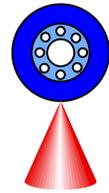
Acknowledgement

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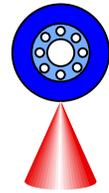
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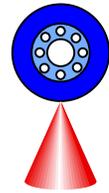


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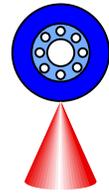


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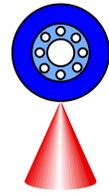


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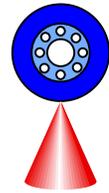


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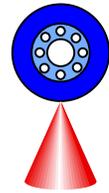
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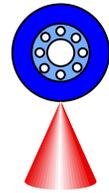
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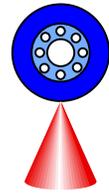
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Appendix A - Units, Dimensions and Dimensional Analysis

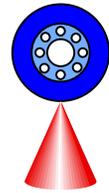
APPENDIX A

**UNITS, DIMENSIONS
AND DIMENSIONAL ANALYSIS**



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Appendix A - Units, Dimensions and Dimensional Analysis

Any physical quantity consists of two parts: 1) A unit, which tells what the quantity is and gives the standard by which it is measured and, 2) a number which tells how many units are needed to make up the quantity, e.g., 8 feet means that a definite length has been measured. The standard length is called the foot and there are eight one-foot units laid end to end that cover the distance. Physical quantities are divided into two groups, termed primary and secondary quantities. A minimum list of primary quantities for all engineering is length, mass, time, temperature and quantity of electric charge. Force and heat may also be added to this list. Secondary units are derived from the primary unit quantities.

The units and dimensions of three common mechanical properties, force, mass and weight, are commonly used. Mass is the quantity conserved in all non-nuclear unit operations; material balances are in terms of mass, not weight. The amount of mass is not affected by body forces, such as gravitational. Weight is often used for both force and mass. Relations exist among these quantities, and units and dimensions must be addressed correctly in engineering formulas and calculations. Force F , body mass M and acceleration a are related by

$$F = kMa. \quad \text{Equation 120}$$

Mass is defined as,

$$M = \lambda(g_c / g) \quad \text{Equation 121}$$

where λ is the weight of fluid.

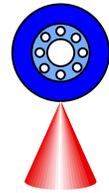
Several systems of units and dimensions are defined and used for the quantities in engineering equations. The various systems differ in the numbers and choice of the fundamental quantities, and in the choice of standard units for the fundamental units. One such system is the centimeter-gram-second (cgs) system. Another is the foot-pound-second (fps) system. The fps units are also called English units. In the fps system, two units, mass and force, have the same name, termed the pound. The lb-mass is designated lb and the lb-force is lb_f.

In the cgs system, only three fundamental mechanical quantities, mass, length and time are used. Force is then a derived or secondary unit. It is defined by making the constant k dimensionless and by choosing its numerical value as unity. To complete this system, mass is measured in grams, length in centimeters and time in seconds. The mks and SI (Système International d'Unités) systems are closely related to the cgs system and use mass measured in kilograms, length in meters, and time in seconds. The constant k is also dimensionless and equal to unity.



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In the fps system, force is a fundamental quantity along with mass, length and time. These four mechanical units are used as the primary quantities. The constant k is a dimensional quantity with a numeric value fixed with a numeric size of 1/32.174. The value of 32.174 is designated as g_c . This value for the gravitational constant g_c is chosen since it is the average acceleration of gravity at sea level. The units of g_c are $\text{ft}\cdot\text{lb}_f\cdot\text{sec}^{-2}$ and g_c is the Newton's Law conversion factor.

For comparison of the cgs/mks/SI and fps units, two common forms are used with mechanical energy: 1) The kinetic energy of translation of a body and 2) the potential energy of a body at a distance above the earth's surface or any arbitrary datum plane. Mechanical energies in the cgs system are in ergs, Newtons for the mks and SI systems, and are $\text{ft}\cdot\text{lb}_f$ in the fps system. For a body mass of M , Z as its distance above a datum plane, u as its kinetic velocity, and g as the local acceleration of gravity, the kinetic and potential energies in the dimensional systems are given in the table below.

Table 28 - Potential and Kinetic Energies

| Potential and Kinetic Energies | | |
|--------------------------------|------------------|---------------|
| Energy | cgs/mks/SI units | fps units |
| Potential | MgZ | MgZ / g_c |
| Kinetic | $Mu^2 / 2$ | $Mu^2 / 2g_c$ |

The fps or cgs/mks/SI units are employed dependent on the technical reference for the primary mechanical quantities. Equations in the fps system are converted to the cgs/mks/SI systems by deleting g_c ; correspondingly, the cgs/mks/SI systems are converted to the fps system by adding g_c .

Dimensional Analysis

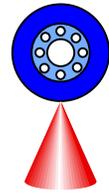
Dimensional analysis is based on the fact, if a theoretical equation exists among variables affecting a physical process, the equation must be dimensionally homogeneous. Because of this requirement, it is possible to group many factors into a smaller number of dimensionless groups of variables. The numerical values of these groups, in any given situation, are independent of the dimension system used, and the groups themselves rather than the separate factors appear in the final equation.

Dimensionless groups may be formed by analyzing the dimensions of the terms appearing in the differential equations governing the various operations. Such a procedure emphasizes the physical significance of the dimensionless groups and guards against inadvertent omission of important variables. Some dimensionless groups occur with such frequency that they have been given names and special



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symbols. The group $dv\rho/\mu$, for example, is called the Reynolds number N_{Re} , and the group $c_p\mu/k$ the Prandtl number N_{Pr} .

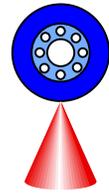
The numerical value of a dimensionless group for a given case is independent of the units chosen for the primary quantities provided consistent units are used within the group. The following provides a listing of dimensionless groups used with fluid systems.

Table 29 - Dimensionless Groups

| Symbol | Name | Definition |
|----------|-------------------------|---|
| C_D | Drag coefficient | $2g_c F_D / u^2 \rho A_p$ |
| f | Fanning friction factor | $-\Delta P g_c d / 2L_p u^2$ |
| j_H | Heat transfer factor | $(h / c_p G)(c_p \mu / k)^{2/3} (\mu_w / \mu)^{0.14}$ |
| j_M | Mass transfer factor | $(k / G_m)(\mu / D_m M)^{2/3}$ |
| N_{Fo} | Fourier number | $kt / c_p \rho A^2$ |
| N_{Fr} | Froude number | u^2 / gL |
| N_{Gr} | Grashof number | $L^3 \rho^2 \beta g \Delta T / \mu^2$ |
| N_{Gz} | Graetz number | mc_p / kL |
| N_{Ma} | Mach number | u / a |
| N_{Nu} | Nusselt number | hd / k |
| N_{Pc} | Peclet number | LGc_p / k |
| N_{Po} | Power number | $Pg_c / \rho n^3 d^5$ |
| N_{Pr} | Prandtl number | $c_p \mu / k$ |
| N_{Re} | Reynolds number | dG / μ |
| N_{Sc} | Schmidt number | $\mu / D_m M$ |
| N_{Sh} | Sherwood number | kd / D_m |



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Appendix B - Total Mass and Energy Equations of Steady Flow

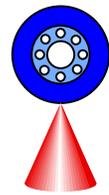
APPENDIX B

TOTAL MASS AND ENERGY EQUATIONS OF STEADY FLOW



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Appendix B - Total Mass and Energy Equations of Steady Flow

The most important processes in unit operations are fluid flow processes, in which fluids flow into, through, and out of pieces of equipment. Accurate and general equations based on the laws of conservation of mass and energy applies directly to flow-systems.

Steady Flow Process

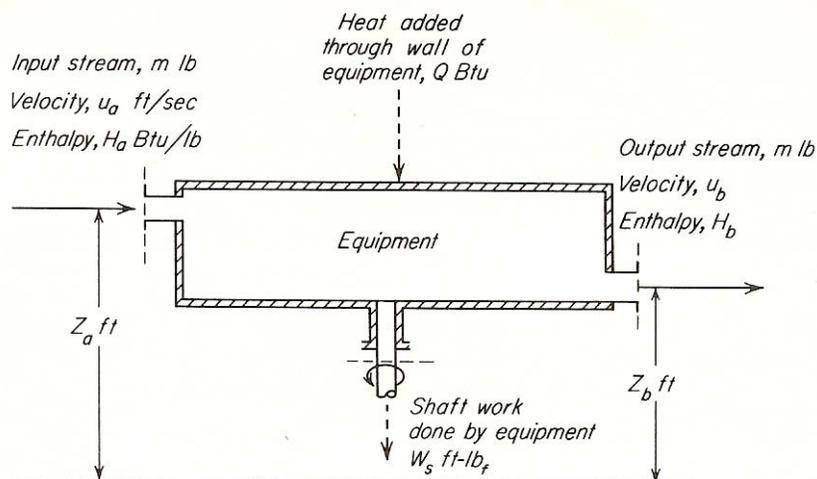
In a steady flow process, the flow rates and the properties of flowing materials such as temperature, pressure, composition, density and velocity, at each point in the apparatus including all entrance and exit ports, are constant with time. These quantities can and usually do vary from point to point in the system; at any one location, they do not change. Because of this constancy of local conditions, there is no accumulation or depletion of either mass or energy within the apparatus. In all material and energy balances the following simple expression applies,

$$\text{input} = \text{output.}$$

Energy Balance for a Single Stream Process

For a steady flow process in which a single stream of fluid material is treated, consider the unit operation as shown in the figure below. The equipment is any device through which a fluid is passing.

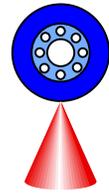
Figure 40 - Diagram for a Steady Flow Process





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The fluid is flowing through the system at constant mass rate with a flow of m lb-mass/time of material. The entering stream has a kinetic velocity of u_a ft/sec and is Z_a ft above a horizontal datum from which heights are measured. Its thermodynamic energy, or enthalpy, is H_a Btu/lb. The enthalpy is a function of the stream's thermodynamic pressure and temperature and is related to the fluid's internal energy (U) and mechanical (P/ρ) energy by, $H = U + P/\rho$. The corresponding quantities for the leaving stream are u_b , Z_b , and H_b . Heat, with the amount Q Btu/time, is being transferred through the boundaries of the equipment to the material flowing through. If the equipment includes a turbine, the equipment may do work by means of a turning shaft. If the unit includes a pump, work from the outside must be done on the equipment, again through the mechanism of a turning shaft. Work effects of this kind are called shaft work, W_s . Shaft work is equal to W_s ft-lb_f/time, which is the amount of work being done on the outside by the equipment. For this process, the following equation that is derived from technical thermodynamics applies,

$$m \left[\frac{u_a^2 - u_b^2}{2g_c J} + \frac{g(Z_b - Z_a)}{g_c J} + H_b - H_a \right] = Q - \frac{W_s}{J} \quad \text{Equation 122}$$

The constant J is the mechanical equivalent of heat, in ft-lb_f/Btu, and g and g_c have their usual meanings.

Limitations and Restrictions

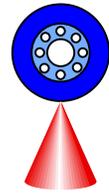
Limitations and restrictions of the above equation are:

1. Changes in electrical, magnetic, surface and mechanical-stress energies are not taken into account. Except in rare situations, these are absent or unimportant.
2. To apply the above equation to a specific situation, a precise choice of the boundaries of the equipment must be made. The inlet and outlet streams must be identified, the inlet and outlet ports located, and rotating shafts noted. All heat-transfer areas between the equipment and its surroundings must be located. The boundaries of the equipment and the cross sections of all shafts and inlet and outlet ports form the control surface. This must be a closed envelope, without gaps. The above equation applies to the equipment and material inside the control surface. The control surface of the process of the figure shown above is bounded by the walls of the equipment, the cross sections of the shaft and the inlet and outlet ports.
3. The constant J is a universal constant, the value of which depends only on the units chosen for heat and work. The fps value of J is 778.26 ft-lb_f/Btu.



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4. For the equation as shown, the heat effect Q is, by convention, positive when heat flows from the outside of the control surface into the equipment and negative when heat flows in the opposite direction. The shaft work W_s is taken as positive when the work is done on the outside of the control surface by the equipment and is negative when the work is supplied to the equipment from outside the control surface. Work required by a pump located within the control surface is negative. Both Q and W_s are net effects on the system. The signs of Q and W_s may be changed dependent on the sign convention used since the values are net effects.

5. No term appears in the above equation for friction or other energy losses. Friction is an internal transformation of mechanical energy into heat and occurs inside the control surface. These effects are included in the other terms of the equation.

Enthalpy

The quantities H_a and H_b in the above equation, the enthalpies of the inlet and outlet streams, respectively, are physical properties of the material. The enthalpy of a unit mass of a pure single-phase substance is a function of pressure and temperature. With a mixture, the enthalpy of a unit mass of a single-phase fluid is a function of pressure, temperature and composition. With a known and constant composition through the equipment, (such as with no chemical reaction) the enthalpy reduces to a function of temperature and pressure only. As with friction or other energy losses, the condition of chemical reaction is included in the other terms of the equation and the composition, temperature and pressure of the fluid at the inlet and at the outlet require measurement.

Bernoulli Equation for Flow Measurement

The total energy equation, without heat or work interaction ($W_s = Q = 0$) and when the inlet and outlet elevation planes are at the same reference elevation ($Z_a = Z_b$) is:

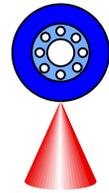
$$m \left[\frac{u_a^2 - u_b^2}{2g_c J} + H_a - H_b \right] = 0. \quad \text{Equation 123}$$

From the mass balance at steady flow, it can be shown that there is a density ρ and average velocity v , such that

$$(\rho Av)_{a,b} = m_a = m_b = \text{Const.} \quad \text{Equation 124}$$



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In the above equation, ρ_a or ρ_b is the average density at the inlet or outlet, v_a or v_b is the average velocity, and A_a or A_b is the flow area at the inlet or outlet respectively. The average velocity v is related to the kinetic energy velocity u by a common factor α , the kinetic energy correction factor, by the following relationship,

$$\alpha v^2 = u^2. \quad \text{Equation 125}$$

The mass flow rate through any apparatus, with the restraints as specified, is provided by substitution of the above relationships into the total energy equation. The mass flow rate is then determined by,

$$m = \rho_a A_a \sqrt{\frac{2g_c J(H_a - H_b)}{\alpha_a \left(\frac{\alpha_b \left(\frac{\rho_a A_a}{\rho_b A_b} \right)^2 - 1}{\alpha_a} \right)}}. \quad \text{Equation 126}$$

This equation is a thermodynamic form of the Bernoulli equation. The basis is very general, adiabatic flow in a horizontal plane. Proper design and utilization provides extremely accurate flow measurement. The kinetic energy correction factor is determined by integration of the velocity profile across the flow area. The numbers approach a value of one (1) for uniform velocity or for moderate to high Reynolds numbers in fully developed boundary layer pipe-flow. The enthalpy and density at the inlet and outlet are determined from the measured thermodynamic pressure and temperature at the inlet and outlet, since for enthalpy $H = f_H(T, P)$ and for density $\rho = f_\rho(T, P)$.

The Bernoulli equation is subject to the constraints of the mass, momentum, energy, and power balances for steady flow systems. It is also subject to the thermodynamics of state. As an example, from the power balance and thermodynamics of state,

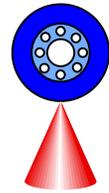
$$\left(\varpi \left(\frac{\alpha v^2}{2g_c J} + U + z\Re T \right) \right)_a - \left(\varpi \left(\frac{\alpha v^2}{2g_c J} + U + z\Re T \right) \right)_b = 0 \quad \text{Equation 127}$$

where v is calculated from the equation of continuity. \Re is calculated from the ideal gas constant R as,

$$\Re = R / M_w \quad \text{Equation 128}$$



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M_w is the fluid's molecular weight. The compressibility factor is calculated from,

$$z = P / \rho \mathcal{R}T \quad \text{Equation 129}$$

and the flow factor ϖ is calculated from,

$$\varpi = PA_v / z \mathcal{R}T = PQ / z \mathcal{R}T \quad \text{Equation 130}$$

where Q is the volumetric flow rate.

These equations were developed across the piece of flow equipment. Macroscopic balances with respect to the fluid are also repeated which provides another grouping of mass, momentum, energy, and power balances, with thermodynamics of state equations. The thermodynamics of state are provided by the Extended Lee-Kesler Equation-of-State (ELK-EoS).

On the macroscopic basis, additional state variables are constant. The variables for transport systems are enthalpy H for insulated systems and Gibbs free enthalpy G for isothermal systems. The isothermal system is defined as one in which heat absorbed or generated does not cause appreciable temperature change.

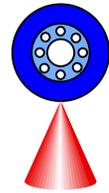
For these systems, the total enthalpy H^* or total Gibbs free enthalpy G^* is constant down the flow measurement device. This provides a cross checking procedure from the a, b, c, etc., locations of the flow tube and within various locations of the flow tube, such as a to b, c to a, b to c, etc. Additionally, with the total enthalpies known, the flow rate at a particular location (a, b, c, etc.) can be determined at that location without any additional information.

All equations with associated parameters and variables may then be solved by a global optimization procedure such as MIMT© to minimize flow measurement errors with respect to pressure and temperature measurements as positioned down the flow tube. This is the basis for high accurate flow measurement systems as developed in the fluid flow industry. This general approach and procedure may be extended to all unit operations, such as pipes, valves, pumps, compressors, turbines, mixers, reactors, etc., within the process industry. This method and procedure extends and significantly improves the global optimization procedure for unit operation analysis, and is attributed to the development of the ELK-EoS and MIMT©.

The basis for accurate flow measurement requires a reliable physical property method to determine density and thermodynamic properties, and requires good design to ensure proper measurement of the true thermodynamic pressure and temperature. These measurements are to be de-coupled of kinetic, potential or other dynamic head



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and thermal response effects. With precise area measurements and known mechanical and fluid-flow design, any flow measurement error can be traced to pressure measurement, temperature measurement or physical properties.

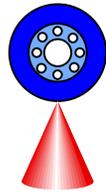
The thermodynamic form of the Bernoulli flow equation is rigorous and can be used for any real-fluid flow measurement within the restraints as listed. Corrections for out-of-plane gravitational fields or variable, directional accelerations are made by re-positioning the pressure tap locations or by use of the following equation,

$$m = \rho_a A_a \sqrt{\frac{2g_c J(H_a - H_b) + 2g(Z_a - Z_b) \cos(\phi)}{\alpha_a \left(\frac{\alpha_b}{\alpha_a} \left(\frac{\rho_a A_a}{\rho_b A_b} \right)^2 - 1 \right)}} . \quad \text{Equation 6}$$

With a two-phase fluid supplementary analysis is required for determining the enthalpy H , density ρ , and kinetic energy correction factor α . The enthalpy, $H = f_H(T, X)$ or $f_H(P, X)$, and density, $\rho = f_\rho(T, X)$ or $f_\rho(P, X)$, are functions of T or P , and X , where X is the amount vaporized. The amount vaporized, X , may be measured or calculated since it is related to the entropy, which is evaluated from additional thermodynamic and transport relations. The kinetic energy correction factor can be determined from two-phase flow analysis methods, e.g., annular, wave, slug or homogeneous two-phase flow regimes.



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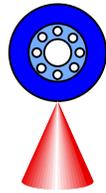
Appendix C - Physical Properties

**APPENDIX C
PHYSICAL PROPERTIES**



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Appendix C - Physical Properties

Volumetric and Thermodynamic Properties Relation

Volumetric and thermodynamic properties of a pure substance (or fluid mixture) are related. From volumetric data, across a wide range of temperatures and pressures for gases, vapors, and liquids, there are relations for calculation of the thermodynamic properties of the substance. Volumetric data can be measured or determined with the “Law of Corresponding States”.

A simple volumetric measurement method is to place a known amount of mass (or moles, $moles = mass / molecular\ weight$) into a small capsule for which the volume is accurately known for all pressure and temperatures. With heating or cooling, the temperature can be adjusted. At the steady state measured temperature, an average (gravity adjusted) pressure is measured. From these measurements, an Equation-of-State (EoS) can be developed of the form,

$$P = f_p(T, (\rho = m/Vol)) \quad \text{Equation 131}$$

where T is the temperature, P is the pressure, m is the mass or moles in the capsule, and Vol is the volume of the capsule. The above equation is developed for diverse amounts of known mass (or moles) charged into the small capsule. With the measured data, a new variable, called the compressibility factor z , can be determined from the measured data. The compressibility factor is defined as,

$$z = \frac{P}{\rho RT} \cdot \quad \text{Equation 132}$$

In the above equation, R is the ideal gas constant when the density units are on a molar basis.

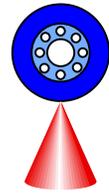
The compressibility z represents a departure from the Ideal Gas Equation-of-State (IG EoS) since $z = 1 = P/(\rho RT)$ for an ideal gas. The departure function is,

$$(1 - z) = \left(\frac{P}{\rho RT} \right)_{IdealGas} - \left(\frac{P}{\rho RT} \right)_{Measured} = \frac{P}{RT} \left(\frac{1}{\rho_{IdealGas}} - \frac{1}{\rho_{Measured}} \right) \cdot \quad \text{Equation 133}$$

With the equation $P = f_p(T, (V = 1/\rho))$ and the departure function $(1 - z)$, the thermodynamic departure functions (Helmholtz free energy A , internal energy U ,



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entropy S , and enthalpy H) are defined, and calculated at constant temperature by the following relations:

$$\frac{A^\circ - A}{RT} = \int_V^\infty (1-z) \frac{dV}{V} + \ln(z) \quad \text{Equation 134}$$

$$\frac{U^\circ - U}{RT} = \int_V^\infty T \left(\frac{\partial Z}{\partial T} \right)_V \frac{dV}{V} \quad \text{Equation 135}$$

$$\frac{S^\circ - S}{R} = \frac{U^\circ - U}{RT} - \frac{A^\circ - A}{RT} \quad \text{Equation 136}$$

$$\frac{H^\circ - H}{RT} = \frac{U^\circ - U}{RT} + (1-z) \quad \text{Equation 137}$$

In the above equations B° refers to the ideal gas state for any property B . The internal energy U , entropy S , and enthalpy H between any two states of pressure and temperature are calculated by,

$$U(T, P)_b - U(T, P)_a = -(U^\circ - U)_b + \int_{T_a}^{T_b} C_V^\circ dT + (U^\circ - U)_a \quad \text{Equation 138}$$

$$S(T, P)_b - S(T, P)_a = -(S^\circ - S)_b + \int_{T_a}^{T_b} \frac{C_P^\circ dT}{T} - R \ln \left(\frac{P_b}{P_a} \right) + (S^\circ - S)_a \quad \text{Equation 139}$$

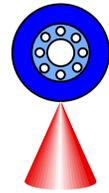
$$H(T, P)_b - H(T, P)_a = -(H^\circ - H)_b + \int_{T_a}^{T_b} C_P^\circ dT + (H^\circ - H)_a \quad \text{Equation 140}$$

For an ideal gas, $\frac{dU}{dP} = 0$, $C_p^\circ = C_v^\circ + R$ and $C_p^\circ \neq f(P, \text{ for any pressure})$.



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The thermodynamic path follows $(B^\circ - B)_a$ at constant T_a from P_a to 0 pressure, the ideal gas enthalpy change from T_a to T_b , and then $(B^\circ - B)_b$ at constant T_b from 0 to P_b pressure.

This powerful approach provides a rigorous and fundamental evaluation of all thermodynamic properties directly from only the fluid's volumetric and ideal gas properties. Volumetric and ideal gas properties are normally obtained from elementary measurements or can be estimated from well-known and accurate methods such as the Law of Corresponding States.

Law of Corresponding States

The basis for the pressure-volume-temperature relationships of constant component gases, vapors and liquids is the thermodynamic proof that for each substance there exists a unique relation among pressure P , temperature T , and specific volume V for any single phase. This thermodynamic postulate is not derived from the Gibbs phase rule, but from the fundamental Gibbs equation relating internal energy U , entropy S , temperature T , and volume V ,

$$dU_{fluid} = T_{fluid} dS_{fluid} - P_{fluid} dV_{fluid} \cdot \quad \text{Equation 141}$$

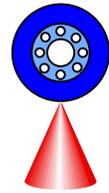
Methods for determination of the pure component volumetric, thermodynamic and transport behavior of liquids, vapors, gases and their mixtures as a function of temperature and pressure are derived from the "Law of Corresponding States". This approach asserts that dimensionless physical properties of materials follow universal variations with dimensionless variables of state. The general and specific forms are derived from molecular theory.

The properties P , V (or ρ), and T are necessary system parameters. Characterizing parameters in the appropriate intermolecular potential function are also necessary; for non-polar molecules, parameters are usually chosen to represent the collision diameter σ and the minimum potential energy ϵ_0 . For polar molecules, in addition to σ and ϵ_0 , the dipole moment μ , the quadrupole moment Q , and the polarizability α are direct choices. Additionally, some variable characteristic of the shape (or nonsphericity) is required; following the suggestion of Bird and Brock, the elliptic eccentricity β is chosen. Finally, the molecular mass m , Planck's constant h , and Boltzmann's constant k complete the list. The base relationship is,

$$\phi_1(P, V, T, \sigma, \epsilon_0, \mu, Q, \alpha, \beta, m, h, k) = 0. \quad \text{Equation 142}$$



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Bird and Brock suggest an arrangement in terms of dimensionless groups,

$$\phi_2(PV/kT, V/\sigma^3, kT/\epsilon_0, \mu^2/\epsilon_0 \sigma^3, Q^2/\epsilon_0 \sigma, \alpha \sigma^3, h/(\sigma(m\epsilon_0)^{1/2}), \beta) = 0. \quad \text{Equation 143}$$

Upon using simplifications, neglecting the quantum group, and applying dimensional analysis, the following equation is obtained,

$$\phi_3(z, P/P_c, T/T_c, \mu^2/V_c kT_c, Q^2/V_c^5 kT_c, \alpha/V_c, \beta) = 0. \quad \text{Equation 144}$$

The above three equations are the basic relations for all the corresponding state correlations.

This is the basis for the departure from an ideal gas condition at the specified pressure and temperature for determining the actual molar (or mass) density. The departure from an ideal gas state is $(1-z)$, where z is the fluid compressibility factor. An

Equation-of-State (EoS) calculates the compressibility factor $z = \frac{PV}{RT}$ and pressure

$P = f_p(T, (V=1/\rho))$. All thermodynamic properties can be determined rigorously with the departure function $(1-z)$, density ρ , and their variations along fixed paths (defined by “natural properties”, such as $\left(\frac{\partial A}{\partial V}\right)_T = P$) and ideal gas properties.

Compressibility and Volumetric Properties

The non-ideality of a fluid as expressed by the compressibility factor z is,

$$z = \frac{P}{\rho RT} = f_z(T, P) = f_z(T, \rho). \quad \text{Equation 145}$$

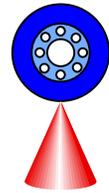
R is the ideal gas constant. P and T are the pressure and temperature, respectively. The molar-density ρ is the defined by

$$\rho = \text{Lim} \left(\frac{\Delta \text{mole}}{\Delta \text{Vol}} \right)_{\Delta \text{Vol} \rightarrow \delta}. \quad \text{Equation 146}$$



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The compressibility is applicable, by the defining equation, to any continuous state. The value of R depends on the units of the variables used. For an ideal gas $z^\circ = 1.0$. For real gases, z is somewhat less than one (< 1.0) except at high-reduced temperatures ($T/T_{critical}$) and pressures ($P/P_{critical}$) where values are greater than one (> 1.0). For liquids between the boiling and melting points (i.e. solids), z is small ($\sim < 0.1$). $Z_{critical}$ is typically between 0.15 and 0.4 at the vapor-liquid critical point of $T_{critical}$, $P_{critical}$ and $\rho_{critical}$.

Calculation of the compressibility z is by the Extended Lee-Kesler Equation-of-State (ELK-EoS) as jointly developed and co-patented with NASA/MSFC, Patent Number 11/152,810. The basis of this EoS is a modified Taylor series expansion that includes the deviation functions of argon with octane and water, and uses the acentric factor (ω), radius-of-gyration (σ), and a Q factor (Θ) as correlating parameters. A similar ELK-EoS relation applies to the transport properties, i.e. acoustic velocity, viscosity, thermal conductivity and surface tension.

Fluid property comparisons are based on the Yaws physical property database, containing over 1700 organic and inorganic materials. Results demonstrate that for a large number of components the volumetric and thermodynamic fluid properties can be evaluated with relative errors typically within 3 to 5 percent over a wide range of operating conditions. For the gas viscosity, relative errors are within 5 percent over the entire range of Yaws Data Base temperatures for over 1700 organic and inorganic compounds (over 4,400 data points). This accuracy range is of the same magnitude as the original experimental data. Improved accuracies will result with improved data. For high accuracy, the Q factor (Θ) may be adjusted to experimental data and calibrated as $\Theta = f_\Theta(T_{reduced}, P_{reduced}) = f_\Theta(T_{reduced}, \rho_{reduced})$. With this adjustment factor, the volumetric, thermodynamic, and transport property accuracy can be of the same order of magnitude as any provided experimental data.

By utilization of a modified Taylor series expansion, the ELK-EoS for compressibility is,

$$z_{fluid} = z^{(0)} + z^{(1)} + z^{(4)} - (\Theta / \Theta^{(0)}) (z^{(1)} - z^{(2)} - z^{(3)} + z^{(4)}). \quad \text{Equation 147}$$

The reference Q factor ($\Theta^{(0)}$) value is 1.16395. $z^{(0)}$ is the compressibility of the simple fluid and the $z^{(i)}$'s represent the deviation functions defined by,

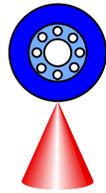
$$z^{(1)} = ((\omega - \omega^{(0)}) / (\omega^{(R)} - \omega^{(0)})) (z^{(R)} - z^{(0)}) \quad \text{Equation 148}$$

$$z^{(2)} = ((\sigma - \sigma^{(0)}) / (\sigma^{(R)} - \sigma^{(0)})) (z^{(R)} - z^{(0)}) \quad \text{Equation 149}$$



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$$z^{(3)} = ((\omega - \omega^{(0)}) / (\omega^{(W)} - \omega^{(0)})) (z^{(W)} - z^{(0)})$$

Equation 150

$$z^{(4)} = ((\sigma - \sigma^{(0)}) / (\sigma^{(W)} - \sigma^{(0)})) (z^{(W)} - z^{(0)}).$$

Equation 151

The correlating factors are listed in the following table.

Table 30 - ELK EoS Correlating Factors

| Component | Superscript | Acentric Factor, ω | Radius-of-Gyration, σ | Q Factor, Θ |
|-----------|-------------|------------------------------|------------------------------|-----------------------|
| Argon | (0) | -0.004605018 | 1.076 | 1 |
| Octane | (R) | 0.397732487 | 4.546 | 1 |
| Water | (W) | 0.3214248 | 0.615 | 1 |

The compressibility factor $z^{(0)}$ of a simple fluid, such as argon, the reference fluid octane $z^{(R)}$, and water $z^{(W)}$ are represented by the twelve (12) constant Modified Benedict-Webb-Rubens (MBWR) equation-of-state,

$$z = (P_r V_r / T_r) = 1 + B/V_r + C/V_r^2 + D/V_r^5 + c4/(T_r^3 V_r^2) (\beta + \gamma/V_r^2) \exp(-\gamma/V_r^2)$$

Equation 152

where

$$B = b1 - b2/T_r - b3/T_r^2 - b4/T_r^3$$

Equation 153

$$C = c1 - c2/T_r + c3/T_r^3$$

Equation 154

$$D = d1 + d2/T_r$$

Equation 155

$$V_r = 1/\rho_r$$

Equation 156

The ELK-EoS MBWR constants are given in the following table for the three specified fluids. The argon and octane constants are the same as provided by Lee-Kesler for their simple and reference fluids. The water constants are derived from the ASME Steam Tables for both water and steam.



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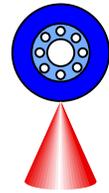


Table 31 - ELK-EoS MBWR Constants

| Constant | Argon | Octane | Water |
|----------------------------------|-----------|-----------|--------------|
| b ₁ | 0.1181193 | 0.2026579 | 0.1263781 |
| b ₂ | 0.265728 | 0.331511 | 0.1786672 |
| b ₃ | 0.15479 | 0.027655 | 0.1308736 |
| b ₄ | 0.030323 | 0.203488 | 0.1944523 |
| c ₁ | 0.0236744 | 0.0313385 | -0.00355054 |
| c ₂ | 0.0186984 | 0.0503618 | -0.000736033 |
| c ₃ | 0 | 0.016901 | 0.017420037 |
| c ₄ | 0.042724 | 0.041577 | 0.006247606 |
| d ₁ x 10 ⁴ | 0.155488 | 0.48736 | 0.131647 |
| d ₂ x 10 ⁴ | 0.623689 | 0.0740336 | -0.0391698 |
| β | 0.65392 | 1.226 | 6.6484506 |
| γ | 0.060167 | 0.03754 | 0.01861792 |

Thermodynamic Properties

The thermodynamic and thermodynamic departure functions derived from the compressibility relation are identical to those as originally developed by Lee-Kesler. These thermodynamic functions are as follows:

Fugacity Coefficient

$$\ln(f/P) = z - 1 - \ln(z) + B/V_r + C/(2V_r^2) + D/(5V_r^5) + E \quad \text{Equation 157}$$

where

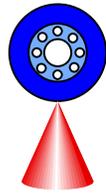
$$E = c_4/(2T_r^3 \gamma) \{ \beta + 1 - (\beta + 1 + \gamma/V_r^2) \exp(-\gamma/V_r^2) \} \quad \text{Equation 158}$$

Enthalpy Departure

$$(H - H^\circ)/RT_c = \{ T_r z - 1 - (b_2 + 2b_3/T_r + 3b_4/T_r^2)/(T_r V_r) - (c_2 - 3c_3/T_r^2)/2T_r V_r^2 + d_2/(5T_r V_r^5) + 3E \} \quad \text{Equation 159}$$



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Entropy Departure

$$(S - S^\circ) / R + \ln(P / P^\circ) = \ln(z) - (b1 + b3/T_r^2 + 2b4/T_r^3) / V_r - (c1 - 2c3/T_r^3) / (2V_r^2) - d1 / (5V_r^5) + 2E \quad \text{Equation 160}$$

Isochoric Heat Capacity Departure

$$(C_v - C_v^\circ) / R = 2(b3 + 3b4/T_r) / (T_r^2 V_r) - 3c3 / (T_r^3 V_r^2) - 6E \quad \text{Equation 161}$$

Isobaric Heat Capacity Departure

$$(C_p - C_p^\circ) / R = (C_v - C_v^\circ) / R - 1 - T_r ((\partial P_r / \partial T_r)_{V_r})^2 / (\partial P_r / \partial V_r)_{T_r} \quad \text{Equation 162}$$

and

$$\begin{aligned} (\partial P_r / \partial T_r)_{V_r} = & (1/V_r) \{1 + (b1 + b3/T_r^2 + 2b4/T_r^3) / V_r \\ & + (c1 - 2c3/T_r^3) / V_r^2 + d1/V_r^5 \\ & - 2c4 / (T_r^3 V_r^2) [(\beta + \gamma/V_r^2) \exp(-\gamma/V_r^2)]\} \end{aligned} \quad \text{Equation 163}$$

$$\begin{aligned} (\partial P_r / \partial V_r)_{T_r} = & -(T_r / V_r^2) \{1 + 2B/V_r + 3C/V_r^2 + 6D/V_r^5 \\ & + c4 / (T_r^3 V_r^2) [3\beta + \{5 - 2(\beta \\ & + \gamma/V_r^2)\} \gamma / V_r^2] \exp(-\gamma/V_r^2)\} \end{aligned} \quad \text{Equation 164}$$

where $P^\circ = 1$ atm, H° = the ideal-gas enthalpy, S° = the ideal-gas entropy, C_v° = ideal-gas constant volume specific heat, and C_p° = ideal-gas constant pressure specific heat, all at the reduced temperature, T_r .

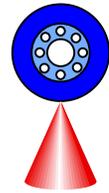
ELK-EoS Calculation Procedure

The following procedure is used to calculate the volumetric or any of the thermodynamic departure functions (X):



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1. Determine the molar volume of the simple fluid, reference fluid and water at the given reduced temperature and pressure from the MBWR compressibility equation.
2. Calculate the departure functions for the volumetric or thermodynamic property of interest, $X^{(0)}$, $X^{(R)}$ and $X^{(W)}$ from the appropriate thermodynamic departure function
3. With the fluids acentric factor and radius-of-gyration, calculate the volumetric or thermodynamic deviation function $X^{(1)}$ through $X^{(4)}$ using the following deviation functions:

$$X^{(1)} = ((\omega - \omega^{(0)}) / (\omega^{(R)} - \omega^{(0)}))(X^{(R)} - X^{(0)}) \quad \text{Equation 165}$$

$$X^{(2)} = ((\sigma - \sigma^{(0)}) / (\sigma^{(R)} - \sigma^{(0)}))(X^{(R)} - X^{(0)}) \quad \text{Equation 166}$$

$$X^{(3)} = ((\omega - \omega^{(0)}) / (\omega^{(W)} - \omega^{(0)}))(X^{(W)} - X^{(0)}) \quad \text{Equation 167}$$

$$X^{(4)} = ((\sigma - \sigma^{(0)}) / (\sigma^{(W)} - \sigma^{(0)}))(X^{(W)} - X^{(0)}) \quad \text{Equation 168}$$

4. The volumetric or thermodynamic property can then be calculated from,

$$X_{fluid} = X^{(0)} + X^{(1)} + X^{(4)} - (\Theta / \Theta^{(0)})(X^{(1)} - X^{(2)} - X^{(3)} + X^{(4)}) \quad \text{Equation 169}$$

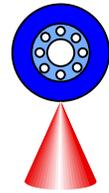
A value of one (1) is normally used for the fluid's Q factor (Θ) unless specific data indicates otherwise. For improved precision in density and thermodynamic property determination, the Q factor (Θ) values are included in the Yaws database for selected components and at various reduced temperatures and pressures. When using these supplied Q factors the volumetric (and associated state functions) errors are zero on the liquid saturation curve, from the melting temperature to the critical point. Alternatively, an average Q factor (Θ) value may be used since the standard deviation across a wide range of reduced temperatures is typically low. For the thermodynamic and vapor pressure fluid properties, a Q factor (Θ) value of one (1) is typically acceptable for engineering applications.

The enthalpy change from state a to state b ($H_b - H_a$) is calculated from,



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$$H_b - H_a = RT_c H^d(T_b, P_b) + \int_{T_a}^{T_b} C_p^\circ dT - RT_c H^d(T_a, P_a)$$

Equation 170

where $H^d(T, P) = ((H - H^\circ) / RT_c)_{fluid}$ as given above, and the superscript $^\circ$ is the ideal gas reference condition. The ideal gas C_p° - A_i values are provided in the Yaws database as a polynomial of form $C_p^\circ = \sum_{i=1}^J A_i T^i$. Similar approaches are used for the other thermodynamic properties. For reaction systems, the above enthalpy equation must be altered to include the heat of formation. For Gibbs or Helmholtz free energy reaction equilibrium calculations, the absolute entropy of formation must be included for the entropy relation.

Certain flow system designs provide a reversible conversion of velocity to enthalpy as given by $dH = u dv / g_c$, where u is the kinetic velocity. This equation is based on the total energy balance, in differential form, with the restrictions as previously listed. The thermodynamic equivalent for a reversible enthalpy change is $dH = V dP$. Integration and solving for the average specific-volume gives $V_{avg} = 1 / \rho_{avg} = v^2 / (2 g_c \Delta P)$. This relation is based on the zero velocity stagnation enthalpy, which is a constant for adiabatic flow through a horizontal pipe and/or through valves and fittings. This density/specific-volume relationship provides redundant volumetric verification that can be used to verify all-fluid thermodynamic properties in flow systems. Pressure taps located in high-flow and stagnation zones can be readily incorporated into the hardware for any head-meter flow system.

Transport Properties

With the transport properties the correlating parameters are given by the same X_{fluid} equations as given above with the following modifications for properties evaluation:

1. For the simple fluid, reference fluid and water the respective transport property is evaluated at the reference temperature as calculated by:

$$T^{(j)} = T_c^{(i)} (T / T_c^{(j)})$$

Equation 171

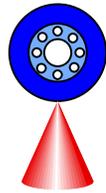
2. A log reduced transport property is based on:

$$X^{(0)} = Ln(f(T^{(0)})), \text{ for the simple fluid}$$

Equation 172



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$$X^{(R)} = Ln(f(T^{(R)})), \text{ for the reference fluid}$$

Equation 173

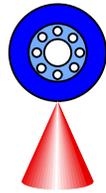
$$X^{(W)} = Ln(f(T^{(W)})), \text{ for water.}$$

Equation 174

3. These equations are used in the formulas as provided above, with specific correlating factors for the transport property of interest.



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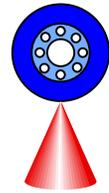
Appendix D - Thermodynamics of Fluid Flow

APPENDIX D
THERMODYNAMICS OF FLUID FLOW



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Appendix D - Thermodynamics of Fluid Flow

Fluid Physical Properties as Thermodynamic State Properties

Fixed state properties connecting any number of thermodynamic paths are independent of the path. The path determines the energy interaction of the system with the environment, typically heat Q or work W . Physical properties, such as density ρ , temperature T , pressure P , viscosity η , etc., are state properties. Any state property that can be expressed uniquely as a function of these properties is itself a state property. All thermodynamic properties, such as enthalpy H , entropy S , internal energy U , etc., are functions of temperature and pressure alone, and are variables of state. Any state variable can be determined by $B = f_B(T, P)$. For temperature T , pressure P , or density ρ , one of many forms for the state representation is:

$$T = (\partial U / \partial S)_V, \quad P = -(\partial U / \partial V)_S, \quad \& \quad 1/\rho = V = (\partial H / \partial P)_S \quad \text{Equation 175}$$

The measured values of temperature, pressure or density are simply a measurement of the respective slopes, or derivatives, of other state properties. As an example, the density of any constant composition fluid can be determined by taking the enthalpy derivative with respect to pressure, at constant entropy, from an equation of state, at the corresponding system temperature and pressure. Since enthalpy and entropy are functions of temperature and pressure only, density and density variations can be determined at any state within a process.

The uniqueness of this equation of state approach provides redundancy, consistency and measurement validation. It also provides closure to the mass, momentum and energy balances as the fluid (system) flows from state a to state b. With fluid flow systems, certain properties are conserved with respect to a reference thermodynamic state, which can be represented by the thermodynamic convenience functions.

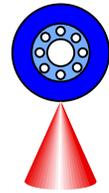
Thermodynamic Convenience Functions

The following equation is the Gibbs fundamental property relationship and represents the third of the three basic equations of thermodynamics. (For fluid flow analysis, the energy and entropy balances are the other two equations coupled with an equation-of-state and the mass and momentum balances.) The Gibbs equation relates the fundamental physical properties of internal energy U , specific volume V , pressure P , entropy S and temperature T , to one another when the system is defined as the fluid. Since it relates only the state functions to each other it is an exact differential equation with extremely powerful mathematical properties in the analysis of fluid flow systems. When put in the form of PdV , other work terms, such as magnetic, electric, gravitation, kinetic, etc., can be added to the fundamental property relationship. These added terms do not affect the basic physical property relationships.



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$$dU_{syst} = T_{syst} dS_{syst} - P_{syst} dV_{syst} \quad \text{Equation 176}$$

It is possible to express all thermodynamic relations in terms of the fundamental properties U , V , P , S , and T . However, with fluid flow analysis, certain properties are conserved when following a thermodynamic path from state a to state b. These properties are convenience functions and provide unique relationships as groupings of new mathematical variables.

The enthalpy convenience function is defined as

$$H = U + PV \quad \text{or} \quad dH = TdS + VdP. \quad \text{Equation 177}$$

The Helmholtz free energy A , and the Gibbs free energy, or Gibbs free enthalpy G , are defined respectively as:

$$A = U - TS \quad \text{or} \quad dA = -SdT - PdV \quad \text{Equation 178}$$

and

$$G = H - TS \quad \text{or} \quad dG = -SdT + VdP. \quad \text{Equation 179}$$

The internal energy U , enthalpy H , Helmholtz free energy A , and Gibbs free enthalpy G provide unique properties when other functions of state are constrained to fixed values. As examples, for an isentropic process,

$$(\partial H / \partial P)_S = V \quad \text{Equation 180}$$

and for an isothermal process,

$$(\partial G / \partial P)_T = V. \quad \text{Equation 181}$$

Integration of the enthalpy term for an isentropic pressure change gives,

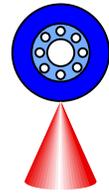
$$\int_a^b (VdP)_S = \int_a^b (\partial H / \partial P)_S dP_S = \int_a^b (dH)_S = \Delta H_S \quad \text{Equation 182}$$

Integration of the Gibbs free enthalpy term for an isothermal pressure change gives,



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$$\int_a^b (VdP)_T = \int_a^b (\partial G / \partial P)_T dP_T = \int_a^b (dG)_T = \Delta G_T . \quad \text{Equation 183}$$

From the Gibbs free enthalpy property relation, the equation for specific volume is,

$$(\partial G / \partial P)_T = V \quad \text{Equation 184}$$

from which a new property is defined as the fugacity f .

Where,

$$dG_T = RTd(\text{Ln}(f)) \text{ or } \Delta G_T = \text{Ln}(f_b / f_a) . \quad \text{Equation 185}$$

The fugacity thermodynamic-property function is defined as

$$f \rightarrow P \quad \text{as } P \rightarrow 0 \quad \text{Equation 186}$$

$$\frac{f}{P} \rightarrow 1 \quad \text{as } P \rightarrow 0 \quad \text{Equation 187}$$

and

$$RT \left(\frac{\partial \text{Ln } f}{\partial P} \right)_T = V = 1 / \rho . \quad \text{Equation 188}$$

For fluids, other than an ideal gas or incompressible liquid, the fugacity representation for density provides improvements in state property calculations when compared to the density. As an example, for saturated liquids and vapors,

$$G_i^{\text{Liquid}} = G_i^{\text{Vapor}} \quad \text{Equation 189}$$

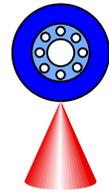
where G_i is the chemical potential for the particular component. The fugacity representation is,

$$f_i^{\text{Liquid}} = f_i^{\text{Vapor}} . \quad \text{Equation 190}$$



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Therefore, for any single component or fluid mixture at the saturation point or within the vapor/liquid envelope, the component fugacity function provides a constant value whereas the density may change by factors of several magnitudes. The fugacity approach provides a constraining-constant factor for rigorous analysis of saturated and two-phase fluid systems.

Reversible and Irreversible Processes

A reversible process requires that the change of states from a to b may be reversed from b to a. Any thermodynamic process or path that does not meet these conditions is called an irreversible process.

Entropy S is used as a measure of this irreversibility and since it is a thermodynamic function of state, its value can be determined. There are two forms in flow systems that generate an entropy change. The first is energy transfer Q from the fluid system to its surroundings; the second is due to frictional effects, termed lost mechanical work LW . The Q and LW can be determined as separate terms from knowledge of the fluid flow process and subsequently related individually to the entropy S .

An irreversible process always involves the degradation of energy potential without producing the maximum amount of work or a corresponding increase in another energy potential other than temperature. An irreversible process includes the flow of a viscous fluid through a pipe and fittings where the system is defined as a fluid.

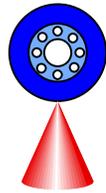
The irreversible degradation of an energy potential in fluid flow processes results from frictional effects or from an imbalance of mechanical-energy potentials. The imbalance occurs when the rate of the process does not allow molecular adjustment of the system. Most fluid flow processes operate at a rate much slower than the molecular processes. Fluid flow analysis is referred to as a quasi-static or quasi-equilibrium and is amenable to analysis by a reversible process technique. The dissipation of a shock wave is an example of mechanical-molecular energy non-uniformity. However, even this process can be adequately evaluated with thermodynamic principles. Frictional losses are evaluated to a high degree of accuracy with mechanical-thermodynamic principles.

Any frictional process is irreversible. However, when the fluid is defined as the system, there may not be degradation of particular convenience-function energy potential. Therefore, even if a process is irreversible, there may be reversible state components within the process that provides a powerful analysis tool. As an example, for an adiabatic and isolated (no Q and W_s) process with frictional lost work LW affects,

$$H^* = H + \sum \text{energy potentials} + \sum \text{frictional lost work} = \text{const.} \quad \text{Equation 191}$$



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Bernoulli Equation or Mechanical Energy Balance

For the flow of fluid through a piece of process equipment, such as a pipeline, pump or expander, the first law differential energy balance is,

$$\left(H + \frac{u^2}{2g_c} + \frac{gZ}{g_c} \right)_a \delta m_a - \left(H + \frac{u^2}{2g_c} + \frac{gZ}{g_c} \right)_b \delta m_b + \delta Q - \delta W_s = d \left[M \left(U + \frac{u^2}{2g_c} + \frac{gZ}{g_c} \right) \right] \quad \text{Equation 192}$$

The entropy balance is,

$$\Delta(\delta m S) + \int_{\text{surface area}} \frac{\delta q dA}{T} + \int_{\text{volume}} \frac{\delta lw dV}{T} = d(MS)_{\text{sys}} \quad \text{Equation 193}$$

For a steady state system, the energy and entropy balances become,

$$\Delta \left(H + \frac{u^2}{2g_c} + \frac{gZ}{g_c} \right)_a m + Q - W_s = 0 \quad \text{Equation 194}$$

$$-TdS + \delta LW + \delta Q = 0 \quad \text{Equation 195}$$

Taking differentials and equating the energy and entropy balances with the Gibbs fundamental relation provides,

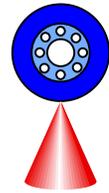
$$-LW - W_s - \int_{P_1}^{P_2} V dP - \Delta \left(\frac{u^2}{2g_c} + \frac{gZ}{g_c} \right) = 0 \quad \text{Equation 196}$$

The Bernoulli representations for isentropic and isothermal fluid flow systems, respectively are,

$$\Delta H_s + \Delta \left(\frac{u^2}{2g_c} + \frac{gZ}{g_c} \right) + LW = 0 \quad \text{Equation 197}$$



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$$\Delta G_T + \Delta \left(\frac{u^2}{2g_c} + \frac{gZ}{g_c} \right) + LW = 0.$$

Equation 198

The total enthalpy H_S^* and Gibbs free enthalpy G_T^* “head” representations for these systems are,

$$H_S^* = (H_S + u^2 / 2g_c + gZ / g_c + \delta LW)_a = \\ (H_S + u^2 / 2g_c + gZ / g_c + \delta LW)_b = Const$$

Equation 199

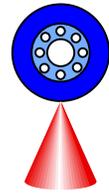
$$G_T^* = (G_T + u^2 / 2g_c + gZ / g_c + \delta LW)_a = \\ (G_T + u^2 / 2g_c + gZ / g_c + \delta LW)_b = Const$$

Equation 200

For two-phase flow systems, a fugacity representation can be substituted in the G_T^* equation. This approach provides the necessary equilibrium calculations throughout the saturated and two-phase regions.



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Appendix E - The Bernoulli Equation for Fluid Flow Measurement

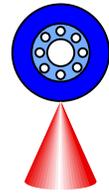
APPENDIX E

THE BERNOULLI EQUATION FOR FLUID FLOW MEASUREMENT



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Appendix E - The Bernoulli Equation for Fluid Flow Measurement

Thermodynamics of fluid-flow measurement systems are based on the first and second laws of thermodynamics and the Gibbs fundamental equation. For a steady state, adiabatic fluid-flow system, with no external work, the first law based on an energy balance is,

$$\Delta(H + \alpha v^2 / 2g_c + gZ / g_c)m = 0 . \quad \text{Equation 201}$$

The second law based on an entropy balance that includes any lost-work LW is,

$$\partial LW = TdS . \quad \text{Equation 202}$$

The Gibbs equation for the fluid is,

$$dU = TdS - PdV \quad \text{Equation 203}$$

and the enthalpy function is,

$$dH = dU + PdV + VdP . \quad \text{Equation 204}$$

Combining equations for dH with $V = 1/\rho$ gives,

$$dH = \delta LW + dP / \rho . \quad \text{Equation 205}$$

The lost work LW is derived from a momentum balance, and is commonly given by,

$$\int \delta LW = LW = kv_b^2 / 2g_c . \quad \text{Equation 206}$$

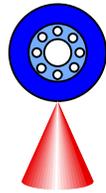
Where k is the velocity head loss, or number of velocity heads.

The total change in enthalpy is,

$$\Delta H = \frac{kv_b^2}{2g_c} + \int_a^b dP / \rho . \quad \text{Equation 207}$$



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Substitution into the total energy balance provides the mechanical energy balance, for $Z_b - Z_a = 0$, in the form of the Bernoulli equation,

$$\frac{(k + \alpha_b)v_b^2 - \alpha_a v_a^2}{2g_c} + \int_a^b dP / \rho = 0. \quad \text{Equation 208}$$

Applying the equation of continuity ($\rho A v = m$) and solving for the mass flow rate in the above equation gives the following general representation of the Bernoulli equation for flow through a horizontal head meter,

$$m = \left(\frac{2g_c \int_a^b dP / \rho}{\left(\frac{k + \alpha}{(\rho A)^2} \right)_b - \left(\frac{\alpha}{(\rho A)^2} \right)_a} \right)^{1/2}. \quad \text{Equation 209}$$

The enthalpy H representation is,

$$m = \left(\frac{2g_c \Delta H}{\left(\frac{k + \alpha}{(\rho A)^2} \right)_b - \left(\frac{\alpha}{(\rho A)^2} \right)_a} \right)^{1/2} = \left(\frac{2g_c \Delta H}{\left(\frac{k + \alpha}{((\partial H / \partial P)_s A)^2} \right)_b - \left(\frac{\alpha}{((\partial H / \partial P)_s A)^2} \right)_a} \right)^{1/2} \quad \text{Equation 210}$$

The Gibbs free enthalpy G representation is,

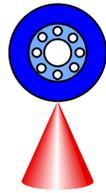
$$m = \left(\frac{2g_c \Delta G}{\left(\frac{k + \alpha}{(\rho A)^2} \right)_b - \left(\frac{\alpha}{(\rho A)^2} \right)_a} \right)^{1/2} = \left(\frac{2g_c \Delta G}{\left(\frac{k + \alpha}{((\partial G / \partial P)_T A)^2} \right)_b - \left(\frac{\alpha}{((\partial G / \partial P)_T A)^2} \right)_a} \right)^{1/2} \quad \text{Equation 211}$$

The Bernoulli equation for flow through a head meter of any orientation is,

$$m = \left(\frac{2g_c \int_a^b dP / \rho + g / g_c (Z_b - Z_a)}{\left(\frac{k + \alpha}{(\rho A)^2} \right)_b - \left(\frac{\alpha}{(\rho A)^2} \right)_a} \right)^{1/2}. \quad \text{Equation 212}$$



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The enthalpy H representation is,

$$m = \left(\frac{2g_c \Delta H + g / g_c (Z_b - Z_a)}{\left(\frac{k + \alpha}{(\rho A)^2} \right)_b - \left(\frac{\alpha}{(\rho A)^2} \right)_a} \right)^{1/2} = \left(\frac{2g_c \Delta H + g / g_c (Z_b - Z_a)}{\left(\frac{k + \alpha}{((\partial H / \partial P)_s A)^2} \right)_b - \left(\frac{\alpha}{((\partial H / \partial P)_s A)^2} \right)_a} \right)^{1/2} \quad \text{Equation 213}$$

The Gibbs free enthalpy G representation is,

$$m = \left(\frac{2g_c \Delta G + g / g_c (Z_b - Z_a)}{\left(\frac{k + \alpha}{(\rho A)^2} \right)_b - \left(\frac{\alpha}{(\rho A)^2} \right)_a} \right)^{1/2} = \left(\frac{2g_c \Delta G + g / g_c (Z_b - Z_a)}{\left(\frac{k + \alpha}{((\partial G / \partial P)_T A)^2} \right)_b - \left(\frac{\alpha}{((\partial G / \partial P)_T A)^2} \right)_a} \right)^{1/2} \quad \text{Equation 214}$$

where $Z_b - Z_a$ is the vertical distance parallel with the local gravitational field.

Another method is the total head method:

$$H_a^* = (H + \alpha V^2 / 2g_c)_a = H_b^* = (H + \alpha' V^2 / 2g_c)_b = \text{Const} \quad \text{Equation 215}$$

where $\alpha' = k + \alpha$.

For which $H_a^* - H_b^* = 0$, and for the a location,

$$m_a = (\rho A ((2g_c / \alpha) (H^* - H)^{1/2}))_a \quad \text{Equation 216}$$

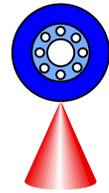
For the b location

$$m_b = (\rho A ((2g_c / \alpha) (H^* - H)^{1/2}))_b \quad \text{Equation 217}$$

Closure is provided when $m_a = m_b$, for the same flow rates as calculated above. A similar approach is made for the isothermal $G^* - G$ condition.



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Additionally, from

$$H^* - H = \alpha V^2 / 2g_c = (\alpha / 2g_c)(m^2 / A^2)(dH / dP)_s^2 \quad \text{Equation 218}$$

and by taking the square roots,

$$(H^* - H)^{1/2} = (\alpha / 2g_c)^{1/2}(m / A)(dH / dP)_s. \quad \text{Equation 219}$$

Then,

$$\int (H^* - H)^{1/2} dP = (\alpha / 2g_c)^{1/2}(m / A) \int (\partial H / \partial P) dP = (\alpha / 2g_c)^{1/2}(m / A)(H^* - H). \quad \text{Equation 220}$$

Solving for the mass flow rate gives,

$$m = (2g_c / \alpha)^{1/2} A \int (H^* - H)^{1/2} dP / (H^* - H) \quad \text{Equation 221}$$

where $H^* = \text{Const}$.

Referencing H^* to a zero (0) datum line gives,

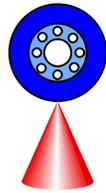
$$m_a = \left((2g_c / \alpha)^{1/2} \frac{A \int_a^* H^{1/2} dP}{H} \right)_a \quad \text{Equation 222}$$

and

$$m_b = \left((2g_c / \alpha)^{1/2} \frac{A \int_b^* H^{1/2} dP}{H} \right)_b \quad \text{Equation 223}$$



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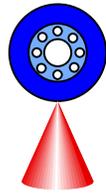


Appendix F - Summary of Macroscopic Balances

APPENDIX F
SUMMARY OF MACROSCOPIC BALANCES



QUALITY MONITORING & CONTROL THERMODYNAMIC AND TRANSPORT ANALYSIS



Appendix F - Summary of Macroscopic Balances

Table 32 - Summary of Macroscopic Balances

| Summary of the Macroscopic Balances for Nonisothermal Flow Systems | | |
|--|--|--|
| Balance | Special Form | Steady State |
| Mass | | $\Delta m = 0$ |
| Momentum | | $F = -\Delta \left(\frac{\langle v^2 \rangle}{\langle v \rangle} m + PS \right) + m_{tot} g$ |
| Energy | | $\Delta \left(U + P/\rho + \frac{1}{2} \frac{\langle v^3 \rangle}{\langle v \rangle} + \Phi - W \right)$ |
| Mechanical Energy | Isothermal | $\Delta \left(\frac{1}{2} \frac{\langle v^3 \rangle}{\langle v \rangle} + \Phi + G \right) + W + E_v = 0$ |
| | Isentropic | $\Delta \left(\frac{1}{2} \frac{\langle v^3 \rangle}{\langle v \rangle} + \Phi + H \right) + W + E_v = 0$ |
| Macroscopic Energy Balance for Variable Potential Energy Systems | | |
| $\frac{d}{dt}(U_{tot} + \phi_{tot} + K_{tot}) = -\Delta[(\hat{U} + \hat{K} + \hat{\phi})_w] + Q - W + \int_V \rho \frac{\partial \phi}{\partial t} dV$ | | |
| Energy quantities defined as: | $U_{tot} = \int_V \rho \hat{U} dVol$ | Internal energy |
| | $K_{tot} = \int_V \frac{1}{2} \rho v^2 dVol$ | Kinetic energy |
| | $\phi_{tot} = \int_V \rho \phi dVol$ | Potential energy |